

(Held On Thursday 28th July, 2022)

TIME : 9 : 00 AM to 12 : 00 NOON

MATHEMATICS

SECTION-A

1. Let the solution curve of the differential equation $xdy = (\sqrt{x^2 + y^2} + y)dx$, $x > 0$, intersect the line $x = 1$ at $y = 0$ and the line $x = 2$ at $y = \alpha$. Then the value of α is :

- (A) $\frac{1}{2}$ (B) $\frac{3}{2}$ (C) $-\frac{3}{2}$ (D) $\frac{5}{2}$

Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol. $xdy = (\sqrt{x^2 + y^2} + y)dx$

$$xdy - ydx = \sqrt{x^2 + y^2}dx$$

$$\frac{xdy - ydx}{x^2} = \sqrt{1 + \frac{y^2}{x^2}} \cdot \frac{dx}{x}$$

$$\frac{d(y/x)}{\sqrt{1 + \left(\frac{y}{x}\right)^2}} = \frac{dx}{x}$$

$$\ln \left(\frac{y}{x} + \sqrt{\left(\frac{y}{x}\right)^2 + 1} \right) = \ln x + R$$

$$\frac{y + \sqrt{y^2 + x^2}}{x} = cx$$

$$y + \sqrt{y^2 + x^2} = cx^2$$

$$x = 1, y = 0 \Rightarrow 0 + 1 = C \Rightarrow C = 1$$

$$\text{Curve is } y + \sqrt{x^2 + y^2} = x^2$$

$$x = 2, y = \alpha$$

$$2 + \sqrt{4 + \alpha^2} = 4$$

$$4 + \alpha^2 = 16 + \alpha^2 = 8\alpha$$

$$\alpha = \frac{3}{2}$$

TEST PAPER WITH SOLUTION

2. Considering only the principal values of the inverse trigonometric functions, the domain of the function $f(x) = \cos^{-1} \left(\frac{x^2 - 4x + 2}{x^2 + 3} \right)$ is :

- (A) $\left[-\infty, \frac{1}{4} \right]$ (B) $\left[-\frac{1}{4}, \infty \right)$
(C) $\left(-\frac{1}{3}, \infty \right)$ (D) $\left(-\infty, \frac{1}{3} \right]$

Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol.

$$\left| \frac{x^2 + 4x + 2}{x^2 + 3} \right| \leq 1$$

$$\Leftrightarrow (x^2 - 4x + 2)^2 \leq (x^2 + 3)^2$$

$$\Leftrightarrow (x^2 - 4x + 2)^2 - (x^2 + 3)^2 \leq 0$$

$$\Leftrightarrow (2x^2 - 4x + 5)(-4x - 1) \leq 0$$

$$\Leftrightarrow -4x - 1 \leq 0 \rightarrow x \geq -\frac{1}{4}$$

3. Let the vectors $\vec{a} = (1+t)\hat{i} + (1-t)\hat{j} + \hat{k}$, $\vec{b} = (1-t)\hat{i} + (1+t)\hat{j} + 2\hat{k}$ and $\vec{c} = t\hat{i} - t\hat{j} + \hat{k}$, $t \in \mathbb{R}$ be such that for $\alpha, \beta, \gamma \in \mathbb{R}$, $\alpha\vec{a} + \beta\vec{b} + \gamma\vec{c} = \vec{0} \Rightarrow \alpha = \beta = \gamma = 0$. Then, the set of all values of t is :

- (A) a non-empty finite set
(B) equal to $\mathbb{N}$
(C) equal to $\mathbb{R} - \{0\}$
(D) equal to $\mathbb{R}$

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. By its given condition

: $\vec{a}, \vec{b}, \vec{c}$ are linearly independent vectors

$$\Rightarrow [\vec{a} \vec{b} \vec{c}] \neq 0 \quad \dots(i)$$

Now, $[\vec{a} \vec{b} \vec{c}]$

$$= \begin{vmatrix} 1+t & 1-t & 1 \\ 1-t & 1+t & 2 \\ t & -t & 1 \end{vmatrix}$$

$$C_2 \rightarrow C_1 + C_2$$

$$\begin{vmatrix} 1+t & 2 & 1 \\ 1-t & 2 & 2 \\ t & 0 & 1 \end{vmatrix}$$

$$= 2 \begin{vmatrix} 1+t & 1 & 1 \\ 1-t & 1 & 2 \\ t & 0 & 1 \end{vmatrix}$$

$$= 2[(1+t) - (1-t) + t]$$

$$= 2[3t] = 6t$$

$$[\vec{a} \vec{b} \vec{c}] \neq 0 \Rightarrow t \neq 0$$

4. Considering the principal values of the inverse trigonometric functions, the sum of all the solutions of the equation $\cos^{-1}(x) - 2\sin^{-1}(x) = \cos^{-1}(2x)$ is equal to :

- (A) 0 (B) 1 (C) $\frac{1}{2}$ (D) $-\frac{1}{2}$

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol. $\cos^{-1} x = 2\sin^{-1} x = \cos^{-1} 2x$

$$\cos^{-1} x - 2\left(\frac{\pi}{2} - \cos^{-1} x\right) = \cos^{-1} 2x$$

$$\cos^{-1} x - \pi + 2\cos^{-1} x = \cos^{-1} 2x$$

$$3\cos^{-1} x = \pi + \cos^{-1} 2x \quad \dots(1)$$

$$\cos(3\cos^{-1} x) = \cos(\pi + \cos^{-1} 2x)$$

$$4x^3 - 3x = -2x$$

$$4x^3 = x \Rightarrow x = 0, \pm \frac{1}{2}$$

All satisfy the original equation

$$\text{sum} = -\frac{1}{2} + 0 + \frac{1}{2} = 0$$

5. Let the operations $*, \odot \in \{\wedge, \vee\}$. If $(p*q)\odot(p\odot\sim q)$ is a tautology, then the ordered pair $(*, \odot)$ is :

- (A) $(\vee, \wedge)$ (B) $(\vee, \vee)$ (C) $(\wedge, \wedge)$ (D) $(\wedge, \vee)$

Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol. Well check each option

For A $\pi = \vee$ of $0 = \wedge$

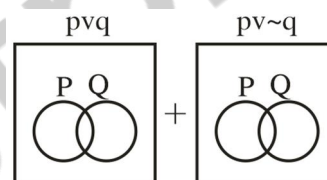
$$(pvq) \wedge (pv \sim q)$$

$$\equiv pv(q \wedge \sim q)$$

$$\equiv pv(0) \equiv p$$

For B : $* = \vee, \odot = \vee$

$$(pvq) \vee (pv \sim q) \equiv t \quad \text{using Venn Diagrams}$$



6. Let a vector $\vec{a}$ has a magnitude 9. Let a vector $\vec{b}$ be such that for every $(x, y) \in \mathbb{R} \times \mathbb{R} - \{(0, 0)\}$, the vector $(x\vec{a} + y\vec{b})$ is perpendicular to the vector $(6y\vec{a} - 18x\vec{b})$. Then the value of $|\vec{a} \times \vec{b}|$ is equal to:

- (A) $9\sqrt{3}$ (B) $27\sqrt{3}$ (C) 9 (D) 81

Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol. $|\vec{a}| = 9$ & $(x\vec{a} + y\vec{b}) \cdot (6y\vec{a} - 18x\vec{b}) = 0$

$$\Rightarrow 6xy|\vec{a}|^2 - 18x^2(\vec{a} \cdot \vec{b}) + 6y^2(\vec{a} \cdot \vec{b}) - 18xy|\vec{b}|^2 = 0$$

$$\Rightarrow 6xy(|\vec{a}|^2 - 3|\vec{b}|^2) + (\vec{a} \cdot \vec{b})(y^2 - 3x^2) = 0$$

This should hold $\forall x, y \in \mathbb{R} \times \mathbb{R}$

$$\therefore |\vec{a}|^2 = 3|\vec{b}|^2 \text{ \& } (\vec{a} \cdot \vec{b}) = 0$$

$$\text{Now } |\vec{a} \times \vec{b}|^2 = |\vec{a}|^2 |\vec{b}|^2 - (\vec{a} \cdot \vec{b})^2$$

$$= |\vec{a}|^2 \cdot \frac{|\vec{a}|^2}{3}$$

$$\therefore |\vec{a} \times \vec{b}| = \frac{|\vec{a}|^2}{\sqrt{3}} = \frac{81}{\sqrt{3}} = 27\sqrt{3}$$



7. For $t \in (0, 2\pi)$, if ABC is an equilateral triangle with vertices $A(\sin t, -\cos t)$, $B(\cos t, \sin t)$ and $C(a, b)$ such that its orthocentre lies on a circle with centre $\left(1, \frac{1}{3}\right)$, then $(a^2 - b^2)$ is equal to :

- (A) $\frac{8}{3}$ (B) 8
(C) $\frac{77}{9}$ (D) $\frac{80}{9}$

Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol. $s \equiv \sin t, c \equiv \cos t$

Let orthocentre be (h, k)

Since it is an equilateral triangle hence orthocentre coincides with centroid.

$$\therefore a + s + c = 3h, b + s - c = 3k$$

$$\therefore (3h - a)^2 + (3k - b)^2 = (s + c)^2 + (s - c)^2 = 2(s^2 + c^2) = 2$$

$$\therefore \left(h - \frac{a}{3}\right)^2 + \left(k - \frac{b}{3}\right)^2 = \frac{2}{9},$$

circle centre at $\left(\frac{a}{3}, \frac{b}{3}\right)$

$$\text{Gives, } \frac{a}{3} = 1, \frac{b}{3} = \frac{1}{3} \Rightarrow a = 3, b = 1$$

$$\therefore a^2 - b^2 = 8$$

8. For $\alpha \in \mathbb{N}$, consider a relation R on $\mathbb{N}$ given by $R = \{(x, y) : 3x + \alpha y \text{ is a multiple of } 7\}$. The relation R is an equivalence relation if and only if :
- (A) $\alpha = 14$
(B) α is a multiple of 4
(C) 4 is the remainder when α is divided by 10
(D) 4 is the remainder when α is divided by 7

Official Ans. by NTA (D)

QuestPix Ans. (D)

Sol. For R to be reflexive $\Rightarrow x R x$

$$\Rightarrow 3x + \alpha x = 7x \Rightarrow (3 + \alpha)x = 7K$$

$$\Rightarrow 3 + \alpha = 7\lambda \Rightarrow \alpha = 7\lambda - 3 = 7N + 4, K, \lambda, N \in \mathbb{I}$$

$\therefore$ when α divided by 7, remainder is 4.

R to be symmetric $xRy \Rightarrow yRx$

$$3x + \alpha y = 7N_1, 3y + \alpha x = 7N_2$$

$$\Rightarrow (3 + \alpha)(x + y) = 7(N_1 + N_2) = 7N_3$$

Which holds when $3 + \alpha$ is multiple of 7

$$\therefore \alpha = 7N + 4 \text{ (as did earlier)}$$

R to be transitive

$$xRy \text{ \& } yRz \Rightarrow xRz.$$

$$3x + \alpha y = 7N_1 \text{ \& } 3y + \alpha z = 7N_2 \text{ and}$$

$$3x + \alpha z = 7N_3$$

$$\therefore 3x + 7N_2 - 3y = 7N_3$$

$$\therefore 7N_1 - \alpha y + 7N_2 - 3y = 7N_3$$

$$\therefore 7(N_1 + N_2) - (3 + \alpha)y = 7N_3$$

$$\therefore (3 + \alpha)y = 7N$$

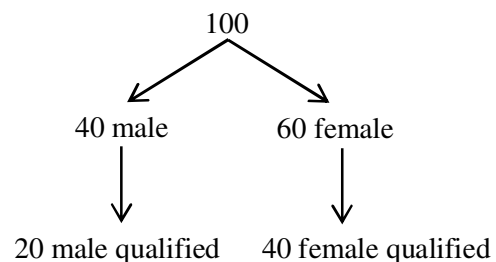
Which is true again when $3 + \alpha$ divisible by 7, i.e. when α divided by 7, remainder is 4.

9. Out of 60% female and 40% male candidates appearing in an exam, 60% candidates qualify it. The number of females qualifying the exam is twice the number of males qualifying it. A candidate is randomly chosen from the qualified candidates. The probability, that the chosen candidate is a female, is :

- (A) $\frac{3}{4}$ (B) $\frac{11}{16}$
(C) $\frac{23}{32}$ (D) $\frac{13}{16}$

Official Ans. by NTA (A)

QuestPix Ans. (Bonus)



$$\text{Probability that chosen candidate is female} = \frac{40}{60} = \frac{2}{3}$$

10. If $y = y(x)$, $x \in \left(0, \frac{\pi}{2}\right)$ be the solution curve of the differential equation

$$(\sin^2 2x) \frac{dy}{dx} + (8 \sin^2 2x + 2 \sin 4x)y =$$

$$2e^{-4x} (2 \sin 2x + \cos 2x), \quad \text{with} \quad y\left(\frac{\pi}{4}\right) = e^{-\pi},$$

then $y\left(\frac{\pi}{6}\right)$ is equal to :

- (A) $\frac{2}{\sqrt{3}}e^{-2\pi/3}$ (B) $\frac{2}{\sqrt{3}}e^{2\pi/3}$
(C) $\frac{1}{\sqrt{3}}e^{-2\pi/3}$ (D) $\frac{1}{\sqrt{3}}e^{2\pi/3}$

Official Ans. by NTA (A)

QuestPix Ans. (A)

- Sol.** Given differential equation can be re-written as

$$\frac{dy}{dx} + (8 + 4 \cot 2x)y = \frac{2e^{-4x}}{\sin^2 2x} (2 \sin x + \cos 2x)$$

which is a linear diff. equation.

$$\text{I.f.} = e^{\int (8+4 \cot 2x) dx} = e^{8x+2 \ln(\sin 2x)}$$

$$= e^{8x} \cdot \sin^2 2x$$

$\therefore$ solution is

$$y(e^{8x} \cdot \sin^2 2x) = \int 2e^{4x} (2 \sin 2x + \cos 2x) dx + C$$

$$= e^{4x} \cdot \sin 2x + C$$

$$\text{Given } y\left(\frac{\pi}{4}\right) = e^{-\pi} \Rightarrow C = 0$$

$$\therefore y = \frac{e^{-4x}}{\sin 2x}$$

$$\therefore y\left(\frac{\pi}{6}\right) = \frac{e^{-4 \cdot \frac{\pi}{6}}}{\sin\left(2 \cdot \frac{\pi}{6}\right)} = \frac{2}{\sqrt{3}} e^{-\frac{2\pi}{3}}$$

11. If the tangents drawn at the points P and Q on the parabola $y^2 = 2x - 3$ intersect at the point R(0, 1), then the orthocentre of the triangle PQR is :

- (A) (0, 1) (B) (2, -1)
(C) (6, 3) (D) (2, 1)

Official Ans. by NTA (B)

QuestPix Ans. (B)

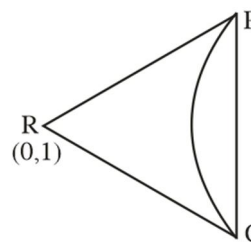
Sol. $y^2 = 2x - 3 \quad \dots(i)$

Equation of chord of contact

$$PQ : r = 0$$

$$yx1 = (x + 0) - 3$$

$$y = x - 3 \quad \dots(2)$$



from (1) and (2)

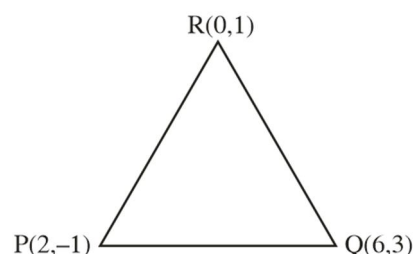
$$(x \cdot 3)^2 = 2x - 3$$

$$x^2 - 8x + 12 = 0$$

$$(x - 2)(x - 6) = 0$$

$$x = 2 \text{ or } 6$$

$$y = -1 \text{ or } 3$$



$$MPQ = \frac{1}{4} = 1$$

$$MQR = \frac{2}{6} = \frac{1}{3}$$

$$MPR = \frac{2}{6} = \frac{1}{3}$$

$$MPR = \frac{2}{-2} = -1$$

$$MPQ \times MPR = - \Rightarrow PQ \perp PR$$

$$\text{Orthocentre} = P(2, -1)$$

$$B_4 = \text{Adj}(\text{Adj}(\text{Adj}(\text{Adj} B_0)))$$

$$= |B_0|^{(n-1)^4}$$

$$= |B_0|^{16}$$

$$B_0 = \begin{bmatrix} 0 & 1 & 0 \\ 0 & 0 & 1 \\ 1 & 0 & 0 \end{bmatrix} + \begin{bmatrix} 2 & 0 & 0 \\ 0 & 2 & 0 \\ 0 & 0 & 2 \end{bmatrix}$$

$$= \begin{bmatrix} 2 & 1 & 0 \\ 0 & 2 & 1 \\ 1 & 0 & 2 \end{bmatrix}$$

$$= 2(4-0) - 1(0-1)$$

$$= 9$$

$$B_4(9)^{16} = (3)^{32}$$

15. Let $S_1 = \left\{ z_1 \in \mathbb{C} : |z_1 - 3| = \frac{1}{2} \right\}$ and

$$S_2 = \left\{ z_2 \in \mathbb{C} : |z_2 - |z_2 + 1|| = |z_2 + |z_2 - 1|| \right\}. \quad \text{Then,}$$

for $z_1 \in S_1$ and $z_2 \in S_2$, the least value of $|z_2 - z_1|$ is :

(A) 0 (B) $\frac{1}{2}$ (C) $\frac{3}{2}$ (D) $\frac{5}{2}$

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. $|z_2 + |z_2 - 1||^2 = |z_2 - |z_2 + 1||^2$

$$\Rightarrow |z_2 + |z_2 - 1||(\bar{z}_2 + |z_2 - 1|) = (z_2 - |z_2 + 1|)(\bar{z}_2 - (z_2 + 1))$$

$$\Rightarrow z_2 \bar{z}_2 + 12z_2 - 1 - (\bar{z}_2 - |z_2 + 1|) + \bar{z}_2 (|z_2 - 1| + |z_2 + 1|)$$

$$= |z_2 + 1|^2 = |z_2 - 1|^2$$

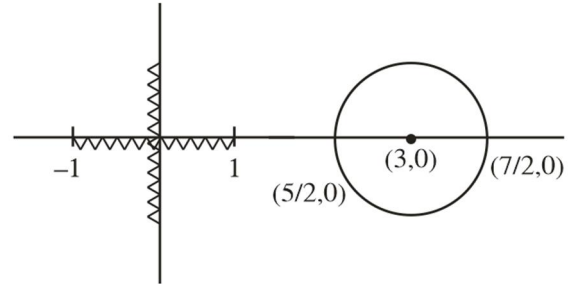
$$[z_2 + \bar{z}_2] (|z_2 - 1| + |z_2 + 1|) = 2(z_2 + \bar{z}_2)$$

$$\Rightarrow (z_2 + \bar{z}_2) (|z_2 - 1| + |z_2 + 1| - 2) = 0$$

$$\therefore z_2 + \bar{z}_2 = 0 \text{ or } |z_2 - 1| + |z_2 + 1| - 2 = 0$$

$\therefore z_2$ lie on imaginary axis. Or on real axis with in $[-1, 1]$

Also $|z_1 - 3| = \frac{1}{2}$ lie on circle having centre 3 and radius $\frac{1}{2}$.



$$\text{Clearly } |z_1 - z_2|_{\min} = \frac{5}{2} - 1 = \frac{3}{2}$$

16. The foot of the perpendicular from a point on the circle $x^2 + y^2 = 1$, $z = 0$ to the plane $2x + 3y + z = 6$ lies on which one of the following curves ?

(A) $(6x + 5y - 12)^2 + 4(3x + 7y - 8)^2 = 1$,

$$z = 6 - 2x - 3y$$

(B) $(5x + 6y - 12)^2 + 4(3x + 5y - 9)^2 = 1$,

$$z = 6 - 2x - 3y$$

(C) $(6x + 5y - 14)^2 + 9(3x + 5y - 7)^2 = 1$,

$$z = 6 - 2x - 3y$$

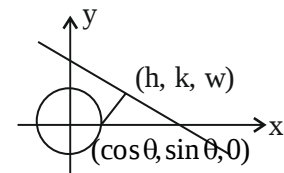
(D) $(5x + 6y - 14)^2 + 9(3x + 7y - 8)^2 = 1$,

$$z = 6 - 2x - 3y$$

Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol.



$$\frac{h - \cos \theta}{2} = \frac{k - \sin \theta}{3} = \frac{w - 0}{1}$$

$$= \frac{-1(2 \cos \theta + 3 \sin \theta - 6)}{14}$$

$$h = \cos \frac{-2(2 \cos \theta + 3 \sin \theta - 6)}{14}$$

$$= \frac{10 \cos \theta - 6 \sin \theta + 12}{14}$$

$$k = \sin \theta - \frac{3}{14}(2 \cos \theta + 3 \sin \theta - 6)$$

$$k = \frac{5 \sin \theta - 6 \cos \theta + 18}{14}$$

Elementary $\sin \theta$ and $\cos \theta$

$$(5h + 6k - 12)^2 + 4(3h + 5k - 9)^2 = 1$$

17. If the minimum value of $f(x) = \frac{5x^2}{2} + \frac{\alpha}{x^5}$, $x > 0$, is

14, then the value of α is equal to :

- (A) 32 (B) 64
(C) 128 (D) 256

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. $\frac{x^2}{2} + \frac{x^2}{2} + \frac{x^2}{2} + \frac{x^2}{2} + \frac{x^2}{2} + \frac{\alpha}{2x^5} + \frac{\alpha}{2x^5}$

$$\geq 7 \left(\frac{\alpha^2}{2^7} \right)^{\frac{1}{7}}$$

$$\frac{7 \cdot (\alpha)^{2/7}}{2} = 14$$

$$(\alpha^2)^{1/7} = 2^2$$

$$\alpha = (2^2)^{7/2} = 2^7$$

$$\alpha = 128$$

18. Let α, β and γ be three positive real numbers. Let

$$f(x) = \alpha x^5 + \beta x^3 + \gamma x, x \in \mathbb{R} \text{ and } g: \mathbb{R} \rightarrow \mathbb{R} \text{ be}$$

such that $g(f(x)) = x$ for all $x \in \mathbb{R}$. If $a_1, a_2, a_3, \dots, a_n$ be in arithmetic progression with mean zero, then

the value of $f\left(g\left(\frac{1}{n} \sum_{i=1}^n f(a_i)\right)\right)$ is equal to :

- (A) 0 (B) 3
(C) 9 (D) 27

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol. Consider a case when $\alpha = \beta = 0$ then

$$f(x) = \gamma x$$

$$g(x) = \frac{x}{\gamma}$$

$$\frac{1}{n} \sum_{i=1}^n f(a_i) \Rightarrow \frac{\gamma}{n} (a_1 + a_2 + \dots + a_n)$$

$$= 0$$

$$\Rightarrow f(g(0)) \Rightarrow f(0)$$

$$\Rightarrow 0$$

19. Consider the sequence $a_1, a_2, a_3, \dots$ such that

$$a_1 = 1, a_2 = 2 \text{ and } a_{n+2} = \frac{2}{a_{n+1}} + a_n \text{ for } n = 1, 2, 3, \dots$$

$$\text{If } \left(\frac{a_1 + \frac{1}{a_2}}{a_3} \right) \cdot \left(\frac{a_2 + \frac{1}{a_3}}{a_4} \right) \cdot \left(\frac{a_3 + \frac{1}{a_4}}{a_5} \right) \dots \left(\frac{a_{30} + \frac{1}{a_{31}}}{a_{32}} \right) = 2^\alpha ({}^{61}C_{31}),$$

then α is equal to :

- (A) -30 (B) -31
(C) -60 (D) -61

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. $a_{n+2} a_{n+1} - a_{n+1} a_n = 2$

Series will satisfy

$$a_1 a_2, a_2 a_3, a_3 a_4, a_4 a_5, \dots$$

$$1.2 \quad 2.2 \quad 2.3 \quad 2.4$$

$$\frac{a_n + \frac{1}{a_{n+1}}}{a_{n+2}} = \frac{a_{n+2} - \frac{1}{a_{n+1}}}{a_{n+2}}$$

$$= 1 - \frac{1}{a_{n+1} a_{n+2}}$$

$$= 1 - \frac{1}{2(r+1)}$$

$$= \frac{2r+1}{2(r+1)}$$

Now proof is given by

$$= \prod_{r=1}^{30} \frac{(2r+1)}{2(r+1)}$$

$$= \frac{(1 \cdot 3 \cdot 5 \cdot \dots \cdot 61)}{2^{30} \cdot (2 \cdot 3 \cdot \dots \cdot 31)}$$

$$\Rightarrow \frac{(1 \cdot 3 \cdot 5 \cdot \dots \cdot 61)}{[31 \cdot 2^{30}]} \times \frac{2^{30} \times [30]}{2^{30} \times [30]}$$

$$= \frac{[61]}{2^{60} [31 \cdot 30]}$$

$$\alpha = -60$$

20. The minimum value of the twice differentiable function $f(x) = \int_0^x e^{x-t} f'(t) dt - (x^2 - x + 1)e^x$, $x \in \mathbb{R}$, is :

(A) $-\frac{2}{\sqrt{e}}$ (B) $-2\sqrt{e}$
 (C) $-\sqrt{e}$ (D) $\frac{2}{\sqrt{e}}$

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol. $f(x) = e^x \cdot \int_0^x \frac{f'(t)}{e^t} dt$
 $f'(x) = e^x \cdot \int_0^x \frac{f'(t)}{e^t} dt + e^x \cdot \frac{f'(x)}{e^x}$

$$-\left[(2x-1) \cdot e^x + (x^2 - x + 1) \cdot e^x\right]$$

 $\int_0^x \frac{f'(t)}{e^t} dt = x^2 + x$
 $\frac{f'(x)}{e^x} = 2x + 1$
 $f'(x) = (2x + 1) \cdot e^x$
 $f'(x) = 0 \Rightarrow x = -\frac{1}{2}$
 $f(x) = (2x + 1) \cdot e^x - 2e^x + C$
 $\left. \begin{array}{l} f(0) = -1 \\ -1 = 1 - 2 + C \\ C = 0 \end{array} \right\}$
 $f(x) = e^x(2x - 1)$
 $f\left(-\frac{1}{2}\right) = \frac{-2}{\sqrt{e}}$

SECTION-B

21. Let S be the set of all passwords which are six to eight characters long, where each character is either an alphabet from {A, B, C, D, E} or a number from {1, 2, 3, 4, 5} with the repetition of characters allowed. If the number of passwords in S whose at least one character is a number from {1, 2, 3, 4, 5} is $\alpha \times 5^6$, then α is equal to _____.

Official Ans. by NTA (7073)

QuestPix Ans. (7073)

Sol. Required no. = Total – no character from {1, 2, 3, 4, 5}
 $= (10^6 - 5^6) + (10^7 - 5^7) + (10^8 - 5^8)$
 $= 10^6(1 + 10 + 100) - 5^6(1 + 5 + 25)$
 $= 10^6 \times 111 - 5^6 \times 31$
 $= 2^6 \times 5^6 \times 111 - 5^6 \times 31$
 $= 5^6(2^6 \times 111 - 31)$
 $= 5^6 \times \underbrace{7073}_\alpha$
 $\therefore \alpha = 7073$

22. Let $P(-2, -1, 1)$ and $Q\left(\frac{56}{17}, \frac{43}{17}, \frac{111}{17}\right)$ be the vertices of the rhombus PQRS. If the direction ratios of the diagonal RS are $\alpha, -1, \beta$, where both α and β are integers of minimum absolute values, then $\alpha^2 + \beta^2$ is equal to _____.

Official Ans. by NTA (450)

QuestPix Ans. (450)

Sol. $RS \equiv (\alpha, -1, \beta)$
 DR of PQ $\equiv \left(\frac{56}{17} + 2, \frac{43}{17} + 1, \frac{111}{17} - 1\right)$
 $\equiv \left(\frac{90}{17}, \frac{60}{17}, \frac{94}{17}\right)$
 $\frac{90}{17}\alpha + \frac{60}{17}(-1) + \frac{94}{17}\beta = 0$
 $90\alpha + 94\beta = 60$
 $\beta = \frac{60 - 90\alpha}{94}$
 $\beta = \frac{30(2 - 3\alpha)}{94}$
 $\beta = -30 \frac{(3\alpha - 2)}{94}$
 $\beta = \frac{-15}{47}(3\alpha - 2)$
 $\Rightarrow \frac{\beta}{-15} = \frac{3\alpha - 2}{47}$
 $\Rightarrow \beta = -15, \alpha = -15$
 $\alpha^2 + \beta^2 = 225 + 225$
 $= 450$

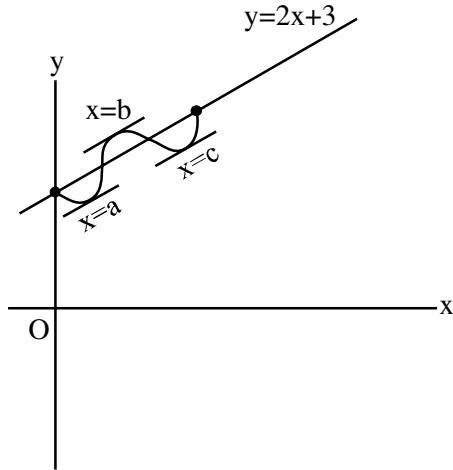


23. Let $f : [0, 1] \rightarrow \mathbb{R}$ be a twice differentiable function in $(0, 1)$ such that $f(0) = 3$ and $f(1) = 5$. If the line $y = 2x + 3$ intersects the graph of f at only two distinct points in $(0, 1)$, then the least number of points $x \in (0, 1)$, at which $f''(x) = 0$, is _____.

Official Ans. by NTA (2)

QuestPix Ans. (2)

Sol.



$f'(a) = f'(b) = f'(c) = 2$
 $\Rightarrow f''(x)$ is zero
 for atleast $x_1 \in (a, b)$ & $x_2 \in (b, c)$

24. If $\int_0^{\sqrt{3}} \frac{15x^3}{\sqrt{1+x^2} + \sqrt{(1+x^2)^3}} dx = \alpha\sqrt{2} + \beta\sqrt{3}$, where α, β are integers, then $\alpha + \beta$ is equal to

Official Ans. by NTA (10)

QuestPix Ans. (10)

Sol. Put $1 + x^2 = t^2$
 $2x dx = 2t dt$
 $x dx = t dt$
 $\therefore \int_1^2 \frac{15(t^2 - 1)t dt}{\sqrt{t^2 + t^3}}$
 $15 \int_1^2 \frac{t(t^2 - 1)}{t\sqrt{1+t}} dt$
 Put $1 + t = u^2$
 $dt = 2u du$
 $15 \int_{\frac{\sqrt{2}}{2}}^{\frac{\sqrt{3}}{2}} \frac{(u^2 - 1)^2 - 1}{u} \times 2u du$
 $30 \int_{\frac{\sqrt{2}}{2}}^{\frac{\sqrt{3}}{2}} (u^4 - 2u^2) du$
 $30 \left(\frac{u^5}{5} - \frac{2u^3}{3} \right)_{\frac{\sqrt{2}}{2}}^{\frac{\sqrt{3}}{2}}$

$$30 \left[\frac{1}{5} (\sqrt{3}^5 - \sqrt{2}^5) - \frac{2}{3} (\sqrt{3}^3 - \sqrt{2}^3) \right]$$

$$30 \left[\frac{1}{5} (9\sqrt{3} - 4\sqrt{2}) - \frac{2}{3} (3\sqrt{3} - 2\sqrt{2}) \right]$$

$$30 \left[-\frac{1}{5} \times \sqrt{3} + \frac{8}{15} \sqrt{2} \right]$$

$$-6\sqrt{3} + 16\sqrt{2} = \alpha\sqrt{2} + \beta\sqrt{3}$$

$$\alpha = 16, \beta = -6$$

$$\therefore \alpha + \beta = 10$$

25. Let $A = \begin{bmatrix} 1 & -1 \\ 2 & \alpha \end{bmatrix}$ and $B = \begin{bmatrix} \beta & 1 \\ 1 & 0 \end{bmatrix}$, $\alpha, \beta \in \mathbb{R}$. Let α_1 be the value of α which satisfies $(A+B)^2 = A^2 + \begin{bmatrix} 2 & 2 \\ 2 & 2 \end{bmatrix}$ and α_2 be the value of α which satisfies $(A+B)^2 = B^2$. Then $|\alpha_1 - \alpha_2|$ is equal to _____.

Official Ans. by NTA (2)

QuestPix Ans. (2)

Sol. $A+B = \begin{bmatrix} \beta+1 & 0 \\ 3 & \alpha \end{bmatrix}$

$$(A+B)^2 = \begin{bmatrix} \beta+1 & 0 \\ 3 & \alpha \end{bmatrix} \begin{bmatrix} \beta+1 & 0 \\ 3 & \alpha \end{bmatrix}$$

$$= \begin{bmatrix} (\beta+1)^2 & 0 \\ 3(\beta+1) + 3\alpha & \alpha^2 \end{bmatrix}$$

$$A^2 = \begin{bmatrix} 1 & -1 \\ 2 & \alpha \end{bmatrix} \begin{bmatrix} 1 & -1 \\ 2 & \alpha \end{bmatrix}$$

$$= \begin{bmatrix} -1 & -1-\alpha \\ 2+2\alpha & \alpha^2-2 \end{bmatrix}$$

$$\therefore \begin{bmatrix} 1 & -\alpha+1 \\ 2\alpha+4 & \alpha^2 \end{bmatrix} = \begin{bmatrix} (\beta+1)^2 & 0 \\ 3(\alpha+\beta+1) & \alpha^2 \end{bmatrix}$$

$$\boxed{\alpha=1} = \alpha_1$$

$$B^2 = \begin{bmatrix} \beta & 1 \\ 1 & 0 \end{bmatrix} \begin{bmatrix} \beta & 1 \\ 1 & 0 \end{bmatrix}$$

$$= \begin{bmatrix} \beta^2+1 & \beta \\ \beta & 1 \end{bmatrix} = \begin{bmatrix} (\beta+1)^2 & 0 \\ 3(\beta+1) + 3\alpha & \alpha^2 \end{bmatrix}$$

$$\therefore \beta = 0, \alpha = -1 = \alpha_2$$

$$|\alpha_1 - \alpha_2| = |1 - (-1)| = 2$$



26. For $p, q \in \mathbb{R}$, consider the real valued function $f(x) = (x - p)^2 - q$, $x \in \mathbb{R}$ and $q > 0$. Let a_1, a_2, a_3 and a_4 be in an arithmetic progression with mean p and positive common difference. If $|f(a_i)| = 500$ for all $i = 1, 2, 3, 4$, then the absolute difference between the roots of $f(x) = 0$ is

Official Ans. by NTA (50)

QuestPix Ans. (50)

Sol. $f(x) = 0 \Rightarrow (x - p)^2 - q = 0$.

Roots are $p + \sqrt{q}$, $p - \sqrt{q}$ absolute difference between roots $2\sqrt{q}$.

Now, $|f(a_i)| = 500$

Let a_1, a_2, a_3, a_4 are $a_1, a + d, a + 2d, a + 3d$

$|f(a_4)| = 500$

$|(a_1 - p)^2 - q| = 500$

$\Rightarrow (a_1 - p)^2 - q = 500$

$\Rightarrow \frac{9}{4}d^2 - q = 500$ _____(1)

and $|f(a_1)|^2 = |f(a_2)|^2$

$((a_1 - p)^2 - q)^2 = ((a_2 - p)^2 - q)^2$

$\Rightarrow ((a_1 - p)^2 - (a_2 - p)^2)((a_1 - p)^2 - q + (a_2 - p)^2 - q) = 0$

$\Rightarrow \frac{9}{4}d^2 - q + \frac{d^2}{4} - q = 0$

$2q = \frac{10d^2}{4} \Rightarrow q = \frac{5d^2}{4}$

$\Rightarrow d^2 = \frac{4q}{5}$

From equation (1) $\frac{9}{4} \cdot \frac{4q}{5} - q = 500$

$\frac{4q}{5} = 500$

$\frac{4q}{5} = 500$

and $2\sqrt{q} = 2 \times \frac{50}{2} = 50$

27. For the hyperbola $H : x^2 - y^2 = 1$ and the ellipse

$E : \frac{x^2}{a^2} + \frac{y^2}{b^2} = 1, a > b > 0$, let the

- (1) eccentricity of E be reciprocal of the eccentricity of H , and

- (2) the line $y = \sqrt{\frac{5}{2}}x + K$ be a common tangent of E and H .

Then $4(a^2 + b^2)$ is equal to _____.

Official Ans. by NTA (3)

QuestPix Ans. (3)

Sol. $e_E = \sqrt{1 - \frac{b^2}{a^2}}, e_H = \sqrt{2}$

If $\Rightarrow e_E = \frac{1}{e_H}$

$\Rightarrow \frac{a^2 - b^2}{a^2} = \frac{1}{2}$

$2a^2 - 2b^2 = a^2$

$a^2 = 2b^2$

and $y = \sqrt{\frac{5}{2}}x + k$ is tangent to ellipse then

$K^2 = a^2 \times \frac{5}{2} + b^2 = \frac{3}{2}$

$6b^2 = \frac{3}{2} \Rightarrow b^2 = \frac{1}{4}$ and $a^2 = \frac{1}{2}$

$\therefore 4(a^2 + b^2) = 3$

28. Let $x_1, x_2, x_3, \dots, x_{20}$ be in geometric progression with $x_1 = 3$ and the common ratio $\frac{1}{2}$. A new data is constructed replacing each x_i by $(x_i - i)^2$. If $\bar{x}$ is the mean of new data, then the greatest integer less than or equal to $\bar{x}$ is _____.

Official Ans. by NTA (142)

QuestPix Ans. (142)

Sol.
$$\sum x_0^1 = \frac{3\left(1 - \left(\frac{1}{2}\right)\right)^{20}}{1 - \frac{1}{2}} = 6\left(1 - \frac{1}{2^{20}}\right)$$

$$= \sum_{i=1}^{20} (x_{i-1})^2$$

$$= \sum_{i=1}^{20} (x_i)^2 + (i)^2 - 2x_i i$$

Now
$$= \sum_{i=1}^{20} (x_i)^2 = \frac{9\left(1 - \left(\frac{1}{4}\right)\right)^{20}}{1 - \frac{1}{4}} = 12\left(1 - \frac{1}{2^{40}}\right)$$

$$\sum_{i=1}^{20} i^2 = \frac{1}{6} \times 20 \times 21 \times 41 = 2870$$

$$\sum_{i=1}^{20} x_i i = s = 3 + 2.3\frac{1}{2} + 3.3\frac{1}{2^2} + 4.3\frac{1}{2^3} + \dots \text{AGP}$$

$$= 6\left(2 - \frac{22}{2^{20}}\right)$$

$$\bar{x} = \frac{12 - \frac{12}{2^{40}} + 2870 - 12\left(2 - \frac{22}{2^{20}}\right)}{20}$$

$$\bar{x} = \frac{2858}{20} + \left(\frac{-12}{2^{40}} + \frac{22}{2^{20}}\right) \times \frac{1}{20}$$

$$[\bar{x}] = 142$$

29.
$$\lim_{x \rightarrow 0} \left(\frac{(x+2\cos x)^3 + 2(x+2\cos x)^2 + 3\sin(x+2\cos x)}{(x+2)^3 + 2(x+2)^2 + 3\sin(x+2)} \right)^{\frac{100}{x}}$$

is equal to _____.

Official Ans. by NTA (1)

QuestPix Ans. (1)

Sol.

$$\lim_{x \rightarrow 10} \left(\frac{(x+2\cos x)^3 + 2(x+2\cos x)^2 + 3\sin(x+2\cos x)}{(x+2)^3 + 2(x+2)^2 + 3\sin(x+2)} \right)^x$$

Form 1^∞

$$= e^{\lim_{x \rightarrow 0} \left[\left(\frac{(x+2\cos x)^3 + 2(x+2\cos x)^2 + 3\sin(x+2\cos x)}{(x+2)^3 + 2(x+2)^2 + 3\sin(x+2)} \right) - 1 \right] \times \frac{100}{x}}$$

$$= e^{\lim_{x \rightarrow 0} \left[\frac{100}{x} \left(\frac{(x+2\cos x)^3 + 2(x+2\cos x)^2 + 3\sin(x+2\cos x) - ((x+2)^3 + 2(x+2)^2 + 3\sin(x+2))}{(x+2)^3 + 2(x+2)^2 + 3\sin(x+2)} \right) \right]}$$

$$= e^{\lim_{x \rightarrow 0} \frac{100}{x} \left[\frac{(x+2\cos x)^3 + (x+2)^3 + 2(x+2\cos x)^2 - 2(x+2)^2 + 3\sin(x+2\cos x) - 3\sin(x+2)}{8+8+3\sin^2} \right]}$$

$$= e^{\frac{100}{16+3\sin^2} \lim_{x \rightarrow 0} \frac{3(x+2\cos x)^2 \times (1+2\sin x) - 3(x+2)^2 - 4(x+2\cos x)}{x(1-2\sin x) - 4(x+2) + 3\cos(x+2\cos x) \times (1-2\sin x) - 3\cos(x+2)}}$$

$$= e^{\frac{100}{16+3\sin^2} \left(\frac{12 - 3(4) + 8 \times 1 - 8 + 3\cos 2 - 3\cos 2}{1} \right)}$$

Using L'H rule.

$$= e^0 = 1$$

30. The sum of all real values of x for which

$$\frac{3x^2 - 9x + 17}{x^2 + 3x + 10} = \frac{5x^2 - 7x + 19}{3x^2 + 5x + 12} \text{ is equal to } \underline{\quad}.$$

Official Ans. by NTA (6)

QuestPix Ans. (6)

Sol.
$$\frac{3x^2 - 9x + 17}{x^2 + 3x + 10} = \frac{5x^2 - 7x + 19}{3x^2 + 5x + 12}$$

$$\frac{x^2 + 3x + 10 + 2x^2 - 12x + 7}{x^2 + 3x + 10} = \frac{3x^2 + 5x + 12 + 2x^2 - 12x + 7}{3x^2 + 5x + 12}$$

$$1 + \frac{2x^2 - 12x + 7}{x^2 + 3x + 10} = 1 + \frac{2x^2 - 12x + 7}{3x^2 + 5x + 12}$$

$$(2x^2 - 12x + 7) \left(\frac{1}{x^2 + 3x + 10} - \frac{1}{3x^2 + 5x + 12} \right) = 0$$

$$2x^2 - 12x + 7 = 0 \text{ OR } 3x^2 + 5x + 12 = x^2 + 3x + 10$$

$$x = \frac{12 \pm \sqrt{D}}{4}$$

$$2x^2 + 2x + 2 = 0$$

$$x^2 + x + 1 = 0$$

Sum of Roots = 6

No solution.

(Held On Thursday 28th July, 2022)

TIME : 9 : 00 AM to 12 : 00 NOON

PHYSICS

SECTION-A

31. The dimensions of $\left(\frac{B^2}{\mu_0}\right)$ will be :

(if μ_0 : permeability of free space and
B : magnetic field)

- (A) $[M L^2 T^{-2}]$ (B) $[M L T^{-2}]$
(C) $[M L^{-1} T^{-2}]$ (D) $[M L^2 T^{-2} A^{-1}]$

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. $u = \frac{B^2}{2\mu_0}$

u $\rightarrow$ Energy per unit volume

$$\left[\frac{B^2}{\mu_0}\right] = [u] = \frac{[ML^2T^{-2}]}{[L^3]} = [ML^{-1}T^{-2}]$$

32. A NCC parade is going at a uniform speed of 9 km/h under a mango tree on which a monkey is sitting at a height of 19.6 m. At any particular instant, the monkey drops a mango. A cadet will receive the mango whose distance from the tree at time of drop is :

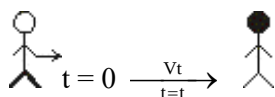
(Given $g = 9.8 \text{ m/s}^2$)

- (A) 5 m (B) 10 m
(C) 19.8 m (D) 24.5 m

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol. Monkey



$$\begin{aligned} \text{Time taken by mango} &= \sqrt{\frac{2n}{g}} \\ &= \sqrt{\frac{2 \times 19.6}{9.8}} = 2 \text{ second} \end{aligned}$$

Distance = vt

$$= 9 \times \frac{5}{18} \times 2 = 5 \text{ m}$$

TEST PAPER WITH SOLUTON

33. In two different experiments, an object of mass 5 kg moving with a speed of 25 ms^{-1} hits two different walls and comes to rest within (i) 3 second, (ii) 5 seconds, respectively. Choose the correct option out of the following :

- (A) Impulse and average force acting on the object will be same for both the cases.
(B) Impulse will be same for both the cases but the average force will be different.
(C) Average force will be same for both the cases but the impulse will be different.
(D) Average force and impulse will be different for both the cases.

Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol. Impulse = change in momentum

$$I = \Delta P$$

$$F_{\text{avg}} = \frac{\Delta P}{\Delta t}$$

$$\Delta t_1 = 3 \quad \Delta t_2 = 5$$

$$\Delta P_1 = \Delta P_2$$

$$I_1 = I_2$$

F_{avg} in case (i) is more than (ii)

34. A balloon has mass of 10 g in air. The air escapes from the balloon at a uniform rate with velocity 4.5 cm/s. If the balloon shrinks in 5 s completely. Then, the average force acting on that balloon will be (in dyne).

- (A) 3 (B) 9 (C) 12 (D) 18

Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol. $F = \frac{dm}{dt} v$

$$= \frac{10g}{5s} \left(4.5 \frac{\text{cm}}{s} \right) = 9 \frac{\text{gcm}}{s^2} = 9 \text{ dyne}$$

35. If the radius of earth shrinks by 2% while its mass remains same. The acceleration due to gravity on the earth's surface will approximately :

(A) decrease by 2% (B) decrease by 4%
(C) increase by 2% (D) increase by 4%

Official Ans. by NTA (D)

QuestPix Ans. (D)

Sol. $g = \frac{GM}{R^2}$

$$M = \text{constant} \quad g \propto \frac{1}{R^2}$$

$$100 \frac{\Delta g}{g} = -2 \frac{\Delta R}{R} 100$$

$$\% \text{ change} = -2 (-2)$$

$$\% \text{ change in } g = 4\%$$

$$\text{increase by } 4\%$$

36. The force required to stretch a wire of cross-section 1 cm^2 to double its length will be :

(Given Yong's modulus of the wire $= 2 \times 10^{11} \text{ N/m}^2$)

(A) $1 \times 10^7 \text{ N}$ (B) $1.5 \times 10^7 \text{ N}$
(C) $2 \times 10^7 \text{ N}$ (D) $2.5 \times 10^7 \text{ N}$

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. $F = \gamma A \frac{\Delta \ell}{\ell}$

$$= 2 \times 10^{11} \times 10^{-4} \left(\frac{2\ell - \ell}{\ell} \right)$$

$$= 2 \times 10^7 \text{ N}$$

37. A Carnot engine has efficiency of 50%. If the temperature of sink is reduced by 40°C , its efficiency increases by 30%. The temperature of the source will be :

(A) 166.7 K (B) 255.1 K
(C) 266.7 K (D) 367.7 K

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. $\eta = 1 - \frac{T_L}{T_H}$

$$\frac{1}{2} = 1 - \frac{T_L}{T_H}$$

$$\frac{1}{2} (1.3) = 1 - \left(\frac{T_L - 40}{T_H} \right)$$

$$\frac{1}{2} (1.3) = \frac{1}{2} + \frac{40}{T_H} \quad T_H = 266.7 \text{ K}$$

38. Given below are two statements :

Statement I : The average momentum of a molecule in a sample of an ideal gas depends on temperature.

Statement II : The rms speed of oxygen molecules in a gas is v . If the temperature is doubled and the oxygen molecules dissociate into oxygen atoms, the rms speed will become $2v$.

In the light of the above statements, choose the correct answer from the options given below :

(A) Both Statement I and Statement II are true
(B) Both Statement I and Statement II are false
(C) Statement I is true but Statement II is false
(D) Statement I is false but Statement II is true

Official Ans. by NTA (D)

QuestPix Ans. (D)

Sol. $[P_{\text{avg}} = 0]$ (due to random motion)

$$v_{\text{rms}} = \sqrt{\frac{3RT}{M}}$$

$$T_{\text{new}} = 2T$$

$$M_{\text{new}} = \frac{M}{2}$$

$$\frac{v_{\text{new}}}{v} = \frac{\sqrt{\frac{2T}{M/2}}}{\sqrt{\frac{T}{M}}}$$

$$v_{\text{new}} = 2v$$

39. In the wave equation

$$y = 0.5 \sin \frac{2\pi}{\lambda} (400t - x) \text{ m}$$

the velocity of the wave will be :

- (A) 200 m/s (B) $200\sqrt{2}$ m/s
(C) 400 m/s (D) $400\sqrt{2}$ m/s

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. $y = 0.5 \sin \left(\frac{2\pi}{\lambda} 400t - \frac{2\pi}{\lambda} x \right)$

$$\omega = \frac{2\pi}{\lambda} 400$$

$$K = \frac{2\pi}{\lambda}$$

$$v = \frac{\omega}{K} \quad [v = 400 \text{ m/s}]$$

40. Two capacitors, each having capacitance $40 \mu\text{F}$ are connected in series. The space between one of the capacitors is filled with dielectric material of dielectric constant K such that the equivalence capacitance of the system became $24 \mu\text{F}$. The value of K will be :

- (A) 1.5 (B) 2.5
(C) 1.2 (D) 3

Official Ans. by NTA (A)

QuestPix Ans. (A)



$$C_{\text{eq}} = \frac{C(KC)}{C + KC} = \frac{KC}{K + 1}$$

$$24 = \frac{K40}{K + 1}$$

$$[K = 1.5]$$

41. A wire of resistance R_1 is drawn out so that its length is increased by twice of its original length.

The ratio of new resistance to original resistance is:

- (A) 9 : 1 (B) 1 : 9
(C) 4 : 1 (D) 3 : 1

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol. $R_1 = \rho \frac{L_1}{A_1}$

$$R_2 = \rho \left(\frac{3L_1}{A_1/3} \right) = 9\rho \frac{L_1}{A_1}$$

$$\therefore \frac{R_2}{R_1} = 9$$

42. The current sensitivity of a galvanometer can be increased by :

- (A) decreasing the number of turns
(B) increasing the magnetic field
(C) decreasing the area of the coil
(D) decreasing the torsional constant of the spring

Choose the most appropriate answer from the options given below :

- (A) (B) and (C) only (B) (C) and (D) only
(C) (A) and (C) only (D) (B) and (D) only

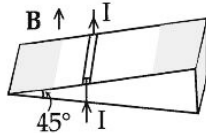
Official Ans. by NTA (D)

QuestPix Ans. (D)

Sol. $i = \left(\frac{K}{NAB} \right) \theta$

$$\therefore \frac{d\theta}{di} = \frac{NAB}{K}$$

43. As shown in the figure, a metallic rod of linear density 0.45 kg m^{-1} is lying horizontally on a smooth incline plane which makes an angle of 45° with the horizontal. The minimum current flowing in the rod required to keep it stationary, when 0.15 T magnetic field is acting on it in the vertical upward direction, will be :
{Use $g = 10 \text{ m/s}^2$ }

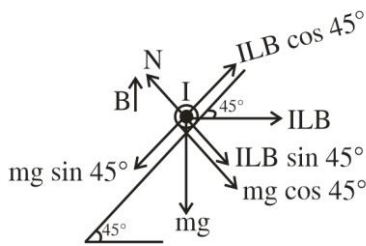


- (A) 30 A (B) 15 A
(C) 10 A (D) 3 A

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol.



$$mg \sin 45^\circ = ILB \cos 45^\circ$$

$$\therefore I = \left(\frac{m}{L} \right) \frac{g}{B}$$

$$= \frac{(0.45)(10)}{0.15} = 30 \text{ A}$$

44. The equation of current in a purely inductive circuit is $5\sin(49\pi t - 30^\circ)$. If the inductance is 30 mH then the equation for the voltage across the inductor, will be :

$$\left\{ \text{Let } \pi = \frac{22}{7} \right\}$$

- (A) $1.47\sin(49\pi t - 30^\circ)$ (B) $1.47\sin(49\pi t + 60^\circ)$
(C) $23.1\sin(49\pi t - 30^\circ)$ (D) $23.1\sin(49\pi t + 60^\circ)$

Official Ans. by NTA (D)

QuestPix Ans. (D)

Sol. $v_0 = i_0 X_L$

$$= i_0 (\omega L)$$

$$= (5)(49\pi)(30 \times 10^{-3})$$

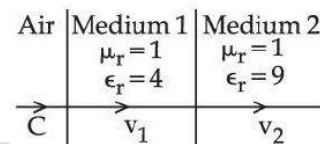
$$= 23.1$$

Voltage will lead current by 90° .

$$\therefore V = 23.1 \sin(49\pi t + 60^\circ)$$

45. As shown in the figure, after passing through the medium 1. The speed of light v_2 in medium 2 will be :

(Given $c = 3 \times 10^8 \text{ ms}^{-1}$)



- (A) $1.0 \times 10^8 \text{ ms}^{-1}$ (B) $0.5 \times 10^8 \text{ ms}^{-1}$
(C) $1.5 \times 10^8 \text{ ms}^{-1}$ (D) $3.0 \times 10^8 \text{ ms}^{-1}$

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol. $\frac{\mu_2}{\mu_{\text{air}}} = \frac{C}{v_2}$

$$\therefore \frac{\sqrt{\mu_{r_2} \epsilon_{r_2}}}{(1)} = \frac{C}{v_2}$$

$$\therefore \sqrt{(1)(9)} = \frac{C}{v_2}$$

$$\therefore v_2 = \frac{C}{3}$$

46. In normal adjustment, for a refracting telescope, the distance between objective and eye piece is 30 cm . The focal length of the objective, when the angular magnification of the telescope is 2, will be:

- (A) 20 cm (B) 30 cm
(C) 10 cm (D) 15 cm

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol. $f_0 + f_e = 30$

$$m = \frac{f_0}{f_e}$$

$$2 = \frac{f_0}{f_e} \Rightarrow f_0 = 2f_e$$

So $f_0 + \frac{f_0}{2} = 30$

$$f_0 = 20 \text{ cm}$$

- 47.** The equation $\lambda = \frac{1.227}{x} \text{ nm}$ can be used to find the de-Broglie wavelength of an electron. In this equation x stands for :

Where,

m = mass of electron

P = momentum of electron

K = Kinetic energy of electron

V = Accelerating potential in volts for electron

(A) $\sqrt{mK}$ (B) $\sqrt{P}$

(C) $\sqrt{K}$ (D) $\sqrt{V}$

Official Ans. by NTA (D)

QuestPix Ans. (D)

Sol. $\lambda = \frac{h}{mv}$ (de-Broglie's wavelength)

$$\lambda = \frac{h}{\sqrt{2m(K \cdot E)}}$$

$$h = \frac{h}{\sqrt{2mqV}}$$

Putting the values of m ; q

We get $\lambda = \frac{1.227}{\sqrt{V}} \text{ nm}$

- 48.** The half life period of a radioactive substance is 60 days. The time taken for $\frac{7}{8}$ th of its original mass to disintegrate will be :
- (A) 120 days (B) 130 days
(C) 180 days (D) 20 days

Official Ans. by NTA (C)

QuestPix Ans. (C)

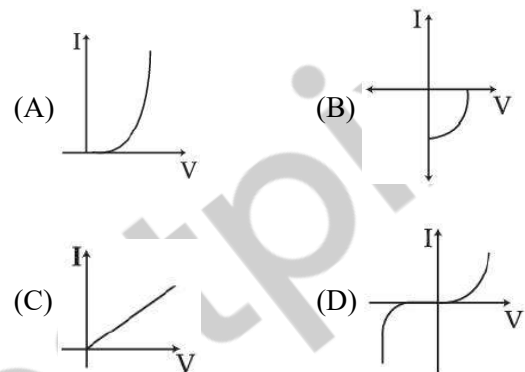
Sol. $\frac{7}{8}$ disintegrates means $\frac{1}{8}$ remains

Or $\left(\frac{1}{2}\right)^3$

$\therefore$ 3 half lives

= 180 days

- 49.** Identify the solar cell characteristics from the following options :



Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol. Conceptual / theory

- 50.** In the case of amplitude modulation to avoid distortion the modulation index (μ) should be :
- (A) $\mu \leq 1$ (B) $\mu \geq 1$
(C) $\mu = 2$ (D) $\mu = 0$

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol. $\mu = \frac{A_m}{A_c}$

$\mu \leq 1$ to avoid distortion

because $\mu > 1$ will result in interference between carrier frequency & message frequency.

SECTION-B

51. If the projection of $2\hat{i} + 4\hat{j} - 2\hat{k}$ on $\hat{i} + 2\hat{j} + \alpha\hat{k}$ is zero. Then, the value of α will be

Official Ans. by NTA (5)

QuestPix Ans. (5)

Sol. $\vec{a} \cdot \vec{b} = 0$

$$\therefore \vec{a} \cdot \vec{b} = 0$$

$$\therefore 2 \times 1 + 4 \times 2 - 2 \times \alpha = 0$$

$$\therefore \alpha = \boxed{5}$$

52. A freshly prepared radioactive source of half life 2 hours 30 minutes emits radiation which is 64 times the permissible safe level. The minimum time, after which it would be possible to work safely with source, will be _____ hours.

Official Ans. by NTA (15)

QuestPix Ans. (15)

Sol. $A = A_0 \times 2^{-t/T}$

$$\frac{A_0}{64} = A_0 \times 2^{-t/T}$$

$$\therefore t = 6T = 6 \times 2 \cdot 5 = \boxed{15} \text{ hours}$$

53. In a Young's double slit experiment, a laser light of 560 nm produces an interference pattern with consecutive bright fringes' separation of 7.2 mm. Now another light is used to produce an interference pattern with consecutive bright fringes' separation of 8.1 mm. The wavelength of second light is _____ nm.

Official Ans. by NTA (630)

QuestPix Ans. (630)

Sol. $\beta \propto \lambda$

$$\lambda_2 = \frac{9}{8} \lambda_1$$

$$\therefore \beta_2 = \frac{9}{8} \beta_1 = \frac{9}{8} \times 560 = \boxed{630} \text{ nm.}$$

54. The frequencies at which the current amplitude in an LCR series circuit becomes $\frac{1}{\sqrt{2}}$ times its maximum value, are 212 rad s^{-1} and 232 rad s^{-1} . The value of resistance in the circuit is $R = 5 \Omega$. The self inductance in the circuit is _____ mH.

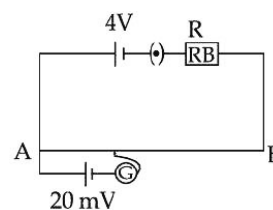
Official Ans. by NTA (250)

QuestPix Ans. (250)

Sol. Band width $= 232 - 212 = \frac{R}{L}$

$$\therefore L = \frac{5}{20} = \boxed{250} \text{ mH}$$

55. As shown in the figure, a potentiometer wire of resistance 20Ω and length 300 cm is connected with resistance box (R.B.) and a standard cell of emf 4 V. For a resistance 'R' of resistance box introduced into the circuit, the null point for a cell of 20 mV is found to be 60 cm. The value of 'R' is _____ Ω .



Official Ans. by NTA (780)

QuestPix Ans. (780)

Sol. $E = \frac{AC}{AB} (V_A - V_B)$

$$\therefore 20 \times 10^{-3} = \frac{60}{300} \times \frac{4 \times 20}{R + 20}$$

$$\therefore R = \boxed{780} \Omega$$

56. Two electric dipoles of dipole moments 1.2×10^{-30} cm and 2.4×10^{-30} cm are placed in two different uniform electric fields of strengths 5×10^4 NC⁻¹ and 15×10^4 NC⁻¹ respectively. The ratio of maximum torque experienced by the electric dipoles will be $\frac{1}{x}$. The value of x is _____.

Official Ans. by NTA (6)

QuestPix Ans. (6)

Sol. $|\tau|_{\max} = PE$

$$\frac{\tau_1}{\tau_2} = \frac{P_1 E_1}{P_2 E_2} = \frac{1.2 \times 10^{-30} \times 5 \times 10^4}{2.4 \times 10^{-30} \times 15 \times 10^4} = \frac{1}{6}$$

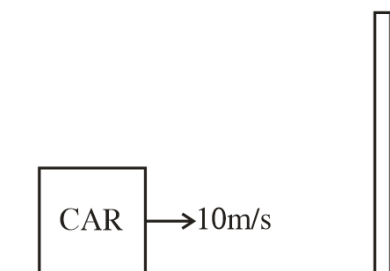
Hence $x = 6$

57. The frequency of echo will be _____ Hz if the train blowing a whistle of frequency 320 Hz is moving with a velocity of 36 km/h towards a hill from which an echo is heard by the train driver. Velocity of sound in air is 330 m/s.

Official Ans. by NTA (340)

QuestPix Ans. (340)

Sol. The hill will be a secondary source.



f_1 = frequency of the car w.r.t. the hill

$$f_1 = \left(\frac{v}{v - v_s} \right) f = \left(\frac{330}{330} \right) \times 320 = 320 \text{ Hz}$$

f_2 = Frequency of the sound reflected by hill w.r.t. the car (echo)

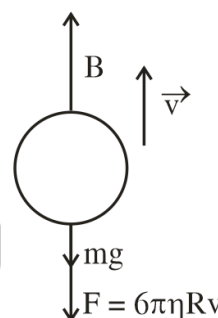
$$f_2 = \left(\frac{v + v_0}{v} \right) f_1 = \left(\frac{330 + 10}{330} \right) \times 320 = 340 \text{ Hz}$$

58. The diameter of an air bubble which was initially 2 mm, rises steadily through a solution of density 1750 kg m^{-3} at the rate of 0.35 cms^{-1} . The coefficient of viscosity of the solution is _____ poise (in nearest integer). (the density of air is negligible).

Official Ans. by NTA (11)

QuestPix Ans. (11)

Sol. As the bubble is rising steadily the net force acting on it will be zero



(Because of density of air the value of mg can be neglected)

$$\text{So } B = F \Rightarrow \frac{4\pi}{3} R^3 \rho g = 6\pi\eta Rv$$

Putting $R = 1 \text{ mm} = 10^{-3} \text{ m}$

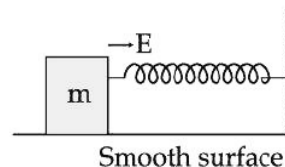
$$\rho = 1.75 \times 10^3 \text{ kg / m}^3$$

$$g = 10 \text{ m / s}^2$$

$$v = 0.35 \times 10^{-2} \text{ m / s}$$

$$\eta = \frac{10}{9} \approx 1.11 \text{ SI unit} = 11 \text{ poise (CGS)}$$

59. A block of mass 'm' (as shown in figure) moving with kinetic energy E compresses a spring through a distance 25 cm when, its speed is halved. The value of spring constant of used spring will be $nE \text{ Nm}^{-1}$ for $n =$ _____.



Official Ans. by NTA (24)

QuestPix Ans. (24)

Sol. Using work – energy theorem

$$W_{\text{net}} = (K_f - K_i)$$

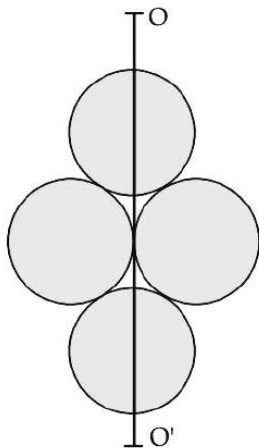
$$\Rightarrow -\frac{1}{2}Kx^2 = \frac{1}{2}m\left(\frac{v}{2}\right)^2 - \frac{1}{2}mv^2 = \frac{E}{4} - E$$

$$\Rightarrow \frac{1}{2}Kx^2 = \frac{3E}{4} \Rightarrow K = \frac{3E}{2x^2}$$

$$\Rightarrow K = \frac{3E}{2 \times \left(\frac{1}{4}\right)^2} = 24E$$

$$n = 24$$

- 60.** Four identical discs each of mass 'M' and diameter 'a' are arranged in a small plane as shown in figure. If the moment of inertia of the system about OO' is $\frac{x}{4}Ma^2$. Then, the value of x will be _____.

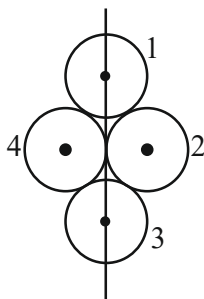


Official Ans. by NTA (3)

QuestPix Ans. (3)

Sol. $I_1 = I_3 = \frac{MR^2}{4}$

$$I_2 = \frac{MR^2}{4} + MR^2 = \frac{5}{4}MR^2 = I_4$$



$$\text{So } I = I_1 + I_2 + I_3 + I_4$$

$$= \frac{MR^2}{4} + \frac{5}{4}MR^2$$

$$= 3MR^2, \text{ Putting } R = \frac{a}{2}$$

$$I = \frac{3Ma^2}{4}, \text{ So } x = 3$$

(Held On Thursday 28th July, 2022)
TIME : 9 : 00 AM to 12 : 00 NOON

CHEMISTRY	TEST PAPER WITH SOLUTION																																								
SECTION-A 61. Identify the incorrect statement from the following. (A) A circular path around the nucleus in which an electron moves is proposed as Bohr's orbit. (B) An orbital is the one electron wave function (Ψ) in an atom. (C) The existence of Bohr's orbits is supported by hydrogen spectrum. (D) Atomic orbital is characterised by the quantum numbers n and l only Official Ans. by NTA (D) QuestPix Ans. (D) Sol. Atomic orbital is characterised by n, l, m. 62. Which of the following relation is not correct ? (A) $\Delta H = \Delta U - P\Delta V$ (B) $\Delta U = q + W$ (C) $\Delta S_{\text{sys}} + \Delta S_{\text{surr}} \geq 0$ (D) $\Delta G = \Delta H - T\Delta S$ Official Ans. by NTA (A) QuestPix Ans. (A) Sol. If $U + Pv$ (By definition) $\Delta H = \Delta U + P\Delta V$ 63. Match List-I with List-II. <table><tr><th></th><th>List-I</th><th></th><th>List-II</th></tr><tr><td>(A)</td><td>$\text{Cd(s)} + 2\text{Ni(OH)}_3\text{(s)} \rightarrow \text{CdO(s)} + 2\text{Ni(OH)}_2\text{(s)} + \text{H}_2\text{O(l)}$</td><td>(I)</td><td>Primary battery</td></tr><tr><td>(B)</td><td>$\text{Zn(Hg)} + \text{HgO(s)} \rightarrow \text{ZnO(s)} + \text{Hg(l)}$</td><td>(II)</td><td>Discharging of secondary battery</td></tr><tr><td>(C)</td><td>$2\text{PbSO}_4\text{(s)} + 2\text{H}_2\text{O(l)} \rightarrow \text{Pb(s)} + \text{PbO}_2\text{(s)} + 2\text{H}_2\text{SO}_4\text{(aq)}$</td><td>(III)</td><td>Fuel cell</td></tr><tr><td>(D)</td><td>$2\text{H}_2\text{(g)} + \text{O}_2\text{(g)} \rightarrow 2\text{H}_2\text{O(l)}$</td><td>(IV)</td><td>Charging of secondary battery</td></tr></table> Choose the correct answer from the options given below : (A) (A) – (I), (B) – (II), (C) – (III), (D) – (IV) (B) (A) – (IV), (B) – (I), (C) – (II), (D) – (III) (C) (A) – (II), (B) – (I), (C) – (IV), (D) – (III) (D) (A) – (II), (B) – (I), (C) – (III), (D) – (IV) Official Ans. by NTA (C) QuestPix Ans. (C)		List-I		List-II	(A)	$\text{Cd(s)} + 2\text{Ni(OH)}_3\text{(s)} \rightarrow \text{CdO(s)} + 2\text{Ni(OH)}_2\text{(s)} + \text{H}_2\text{O(l)}$	(I)	Primary battery	(B)	$\text{Zn(Hg)} + \text{HgO(s)} \rightarrow \text{ZnO(s)} + \text{Hg(l)}$	(II)	Discharging of secondary battery	(C)	$2\text{PbSO}_4\text{(s)} + 2\text{H}_2\text{O(l)} \rightarrow \text{Pb(s)} + \text{PbO}_2\text{(s)} + 2\text{H}_2\text{SO}_4\text{(aq)}$	(III)	Fuel cell	(D)	$2\text{H}_2\text{(g)} + \text{O}_2\text{(g)} \rightarrow 2\text{H}_2\text{O(l)}$	(IV)	Charging of secondary battery	Sol. (a) $\text{Cd(s)} + 2\text{Ni(OH)}_3\text{(s)} \rightarrow \text{CdO(s)} + 2\text{Ni(OH)}_2\text{(s)} + \text{H}_2\text{O(l)}$ Discharge of secondary Battery (b) $\text{Zn(Hg)} + \text{HgO(s)} \rightarrow \text{ZnO(s)} + \text{Hg(l)}$ (Primary Battery Mercury cell) (c) $2\text{PbSO}_4\text{(s)} + 2\text{H}_2\text{O(l)} \rightarrow \text{Pb(s)} + \text{PbO}_2\text{(s)} + 2\text{H}_2\text{SO}_4\text{(aq)}$ Charging of secondary Battery (d) $2\text{H}_2\text{(g)} + \text{O}_2\text{(g)} \rightarrow 2\text{H}_2\text{O(l)}$ – Fuel cell 64. Match List-I with List-II. <table><tr><th></th><th>List-I Reaction</th><th></th><th>List-II Catalyst</th></tr><tr><td>(A)</td><td>$4\text{NH}_3\text{(g)} + 5\text{O}_2\text{(g)} \rightarrow 4\text{NO(g)} + 6\text{H}_2\text{O(g)}$</td><td>(I)</td><td>$\text{NO(g)}$</td></tr><tr><td>(B)</td><td>$\text{N}_2\text{(g)} + 3\text{H}_2\text{(g)} \rightarrow 2\text{NH}_3\text{(g)}$</td><td>(II)</td><td>$\text{H}_2\text{SO}_4\text{(l)}$</td></tr><tr><td>(C)</td><td>$\text{C}_{12}\text{H}_{22}\text{O}_{11}\text{(aq)} + \text{H}_2\text{O(l)} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 \text{ (Glucose)} + \text{C}_6\text{H}_{12}\text{O}_6 \text{ (Fructose)}$</td><td>(III)</td><td>$\text{Pt(s)}$</td></tr><tr><td>(D)</td><td>$2\text{SO}_2\text{(g)} + \text{O}_2\text{(g)} \rightarrow 2\text{SO}_3\text{(g)}$</td><td>(IV)</td><td>$\text{Fe(s)}$</td></tr></table> Choose the correct answer from the options given below : (A) (A) – (II), (B) – (III), (C) – (I), (D) – (IV) (B) (A) – (III), (B) – (II), (C) – (I), (D) – (IV) (C) (A) – (III), (B) – (IV), (C) – (II), (D) – (I) (D) (A) – (III), (B) – (II), (C) – (IV), (D) – (I) Official Ans. by NTA (C) QuestPix Ans. (C) Sol. (a) $4\text{NH}_3\text{(g)} + 5\text{O}_2\text{(g)} \xrightarrow{\text{Pt(s)}} 4\text{NO(g)} + 6\text{H}_2\text{O(g)}$ Ostwald process 500 K (b) $\text{N}_2 + 3\text{H}_2 \xrightarrow{\text{Fe(s)}} 2\text{NH}_3\text{(g)}$ Haber's process (c) $\text{C}_{12}\text{H}_{22}\text{O}_{11}\text{(aq.)} + \text{H}_2\text{O(l)} \xrightarrow{\text{H}^+} \underset{\text{(glucose)}}{\text{C}_6\text{H}_{12}\text{O}_6} + \underset{\text{(fructose)}}{\text{C}_6\text{H}_{12}\text{O}_6}$ Inversion of sugar cane (d) $2\text{SO}_2\text{(g)} + \text{O}_2\text{(g)} \xrightarrow{\text{NO(g)}} 2\text{SO}_3\text{(g)}$		List-I Reaction		List-II Catalyst	(A)	$4\text{NH}_3\text{(g)} + 5\text{O}_2\text{(g)} \rightarrow 4\text{NO(g)} + 6\text{H}_2\text{O(g)}$	(I)	NO(g)	(B)	$\text{N}_2\text{(g)} + 3\text{H}_2\text{(g)} \rightarrow 2\text{NH}_3\text{(g)}$	(II)	$\text{H}_2\text{SO}_4\text{(l)}$	(C)	$\text{C}_{12}\text{H}_{22}\text{O}_{11}\text{(aq)} + \text{H}_2\text{O(l)} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 \text{ (Glucose)} + \text{C}_6\text{H}_{12}\text{O}_6 \text{ (Fructose)}$	(III)	Pt(s)	(D)	$2\text{SO}_2\text{(g)} + \text{O}_2\text{(g)} \rightarrow 2\text{SO}_3\text{(g)}$	(IV)	Fe(s)
	List-I		List-II																																						
(A)	$\text{Cd(s)} + 2\text{Ni(OH)}_3\text{(s)} \rightarrow \text{CdO(s)} + 2\text{Ni(OH)}_2\text{(s)} + \text{H}_2\text{O(l)}$	(I)	Primary battery																																						
(B)	$\text{Zn(Hg)} + \text{HgO(s)} \rightarrow \text{ZnO(s)} + \text{Hg(l)}$	(II)	Discharging of secondary battery																																						
(C)	$2\text{PbSO}_4\text{(s)} + 2\text{H}_2\text{O(l)} \rightarrow \text{Pb(s)} + \text{PbO}_2\text{(s)} + 2\text{H}_2\text{SO}_4\text{(aq)}$	(III)	Fuel cell																																						
(D)	$2\text{H}_2\text{(g)} + \text{O}_2\text{(g)} \rightarrow 2\text{H}_2\text{O(l)}$	(IV)	Charging of secondary battery																																						
	List-I Reaction		List-II Catalyst																																						
(A)	$4\text{NH}_3\text{(g)} + 5\text{O}_2\text{(g)} \rightarrow 4\text{NO(g)} + 6\text{H}_2\text{O(g)}$	(I)	NO(g)																																						
(B)	$\text{N}_2\text{(g)} + 3\text{H}_2\text{(g)} \rightarrow 2\text{NH}_3\text{(g)}$	(II)	$\text{H}_2\text{SO}_4\text{(l)}$																																						
(C)	$\text{C}_{12}\text{H}_{22}\text{O}_{11}\text{(aq)} + \text{H}_2\text{O(l)} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 \text{ (Glucose)} + \text{C}_6\text{H}_{12}\text{O}_6 \text{ (Fructose)}$	(III)	Pt(s)																																						
(D)	$2\text{SO}_2\text{(g)} + \text{O}_2\text{(g)} \rightarrow 2\text{SO}_3\text{(g)}$	(IV)	Fe(s)																																						

65. In which of the following pairs, electron gain enthalpies of constituent elements are nearly the same or identical ?

(A) Rb and Cs (B) Na and K
(C) Ar and Kr (D) I and At

Choose the correct answer from the options given below :

(A) (A) and (B) only
(B) (B) and (C) only
(C) (A) and (C) only
(D) (C) and (D) only

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. Rb & Cs have nearly same electron gain enthalpy
electron gain enthalpy = - 46 kJ/mol

Ar & Kr have same ΔH_{eq} . Value is + 96 kJ/mol

66. Which of the reaction is suitable for concentrating ore by leaching process ?

(A) $2\text{Cu}_2\text{S} + 3\text{O}_2 \rightarrow 2\text{Cu}_2\text{O} + 2\text{SO}_2$
(B) $\text{Fe}_3\text{O}_4 + \text{CO} \rightarrow 3\text{FeO} + \text{CO}_2$
(C) $\text{Al}_2\text{O}_3 + 2\text{NaOH} + 3\text{H}_2\text{O} \rightarrow 2\text{Na}[\text{Al}(\text{OH})_4]$
(D) $\text{Al}_2\text{O}_3 + 6\text{Mg} \rightarrow 6\text{MgO} + 4\text{Al}$

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. $\text{Al}_2\text{O}_3 + 2\text{NaOH} + 3\text{H}_2\text{O} \rightarrow 2\text{Na}[\text{Al}(\text{OH})_4]$
Leaching.

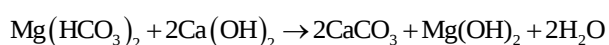
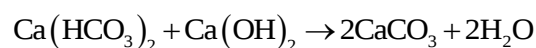
67. The metal salts formed during softening of hardwater using Clark's method are :

(A) $\text{Ca}(\text{OH})_2$ and $\text{Mg}(\text{OH})_2$
(B) CaCO_3 and $\text{Mg}(\text{OH})_2$
(C) $\text{Ca}(\text{OH})_2$ and MgCO_3
(D) CaCO_3 and MgCO_3

Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol. Clark's Method Reaction



68. Which of the following statement is incorrect ?

(A) Low solubility of LiF in water is due to its small hydration enthalpy.
(B) KO_2 is paramagnetic.
(C) Solution of sodium in liquid ammonia is conducting in nature.
(D) Sodium metal has higher density than potassium metal

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol. Low solubility of LiF in water is due to high lattice enthalpy

69. Match List-I with List-II, match the gas evolved during each reaction.

	List-I		List-II
(A)	$(\text{NH}_4)_2\text{Cr}_2\text{O}_7 \xrightarrow{\Delta}$	(I)	H_2
(B)	$\text{KMnO}_4 + \text{HCl} \rightarrow$	(II)	N_2
(C)	$\text{Al} + \text{NaOH} + \text{H}_2\text{O} \rightarrow$	(III)	O_2
(D)	$\text{NaNO}_3 \xrightarrow{\Delta}$	(IV)	Cl_2

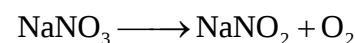
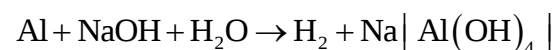
Choose the correct answer from the options given below :

(A) (A) - (II), (B) - (III), (C) - (I), (D) - (IV)
(B) (A) - (III), (B) - (I), (C) - (IV), (D) - (II)
(C) (A) - (II), (B) - (IV), (C) - (I), (D) - (III)
(D) (A) - (III), (B) - (IV), (C) - (I), (D) - (II)

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. $(\text{NH}_4)_2\text{Cr}_2\text{O}_7 \xrightarrow{\Delta} \text{N}_2 + \text{Cr}_2\text{O}_3 + 4\text{H}_2\text{O}$



70. Which of the following has least tendency to liberate H_2 from mineral acids ?

(A) Cu (B) Mn
(C) Ni (D) Zn

Official Ans. by NTA (A)

QuestPix Ans. (A)

Sol. Copper is least electropositive among the given metals and it lies below H in reactivity series

71. Given below are two statements :

Statement I : In polluted water values of both dissolved oxygen and BOD are very low.

Statement II : Eutrophication results in decrease in the amount of dissolved oxygen.

In the light of the above statements, choose the most appropriate answer from the options given below :

- (A) Both Statement I and Statement II are true
(B) Both Statement I and Statement II are false
(C) Statement I is true but Statement II is false
(D) Statement I is false but Statement II is true

Official Ans. by NTA (D)

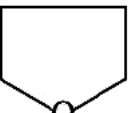
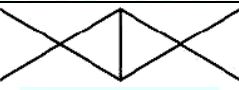
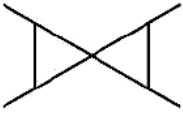
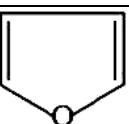
QuestPix Ans. (D)

Sol. Since eutrophication is result of excessive growth of weed in water bodies, which consume dissolved oxygen of water bodies.

∴ Eutrophication decreases amount of dissolved oxygen in water bodies.

Polluted water has low value of dissolved oxygen, but high value of BOD (Biological oxygen demand), since chemical and organic matter requires dissolved oxygen to get decompose.

72. Match List-I with List-II.

	List-I		List-II
(A)		(I)	Spiro compound
(B)		(II)	Aromatic compound
(C)		(III)	Non-planar Heterocyclic compound
(D)		(IV)	Bicyclo compound

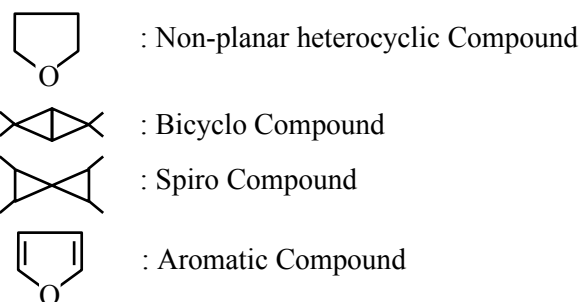
Choose the correct answer from the options given below :

- (A) (A) – (II), (B) – (I), (C) – (IV), (D) – (III)
(B) (A) – (IV), (B) – (III), (C) – (I), (D) – (II)
(C) (A) – (III), (B) – (IV), (C) – (I), (D) – (II)
(D) (A) – (IV), (B) – (III), (C) – (II), (D) – (I)

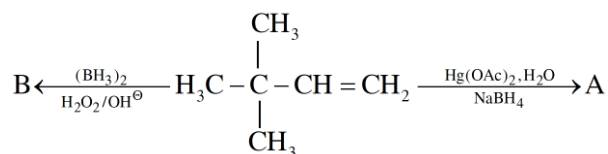
Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol.



73. Choose the correct option for the following reactions.



(A) 'A' and 'B' are both Markovnikov addition products.

(B) 'A' is Markovnikov product and 'B' is anti-Markovnikov product.

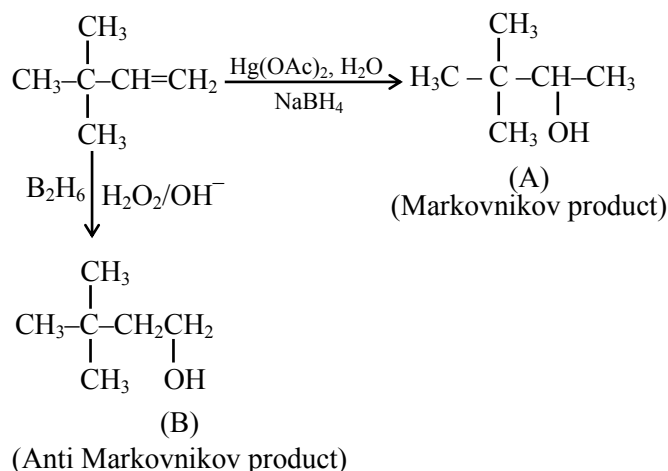
(C) 'A' and 'B' are both anti-Markovnikov products.

(D) 'B' is Markovnikov and 'A' is anti-Markovnikov product.

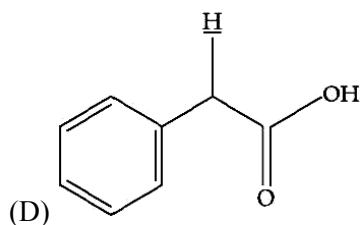
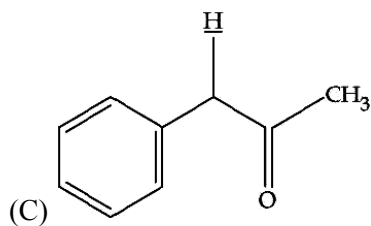
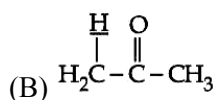
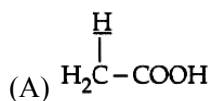
Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol.



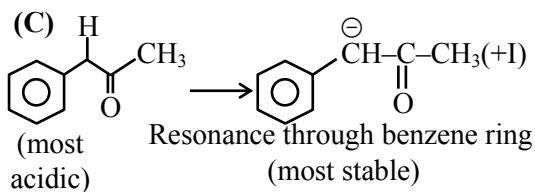
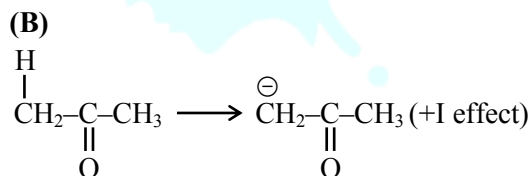
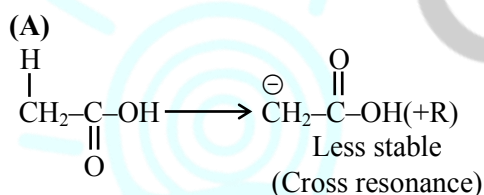
74. Among the following marked proton of which compound shows lowest pK_a value ?



Official Ans. by NTA (C)

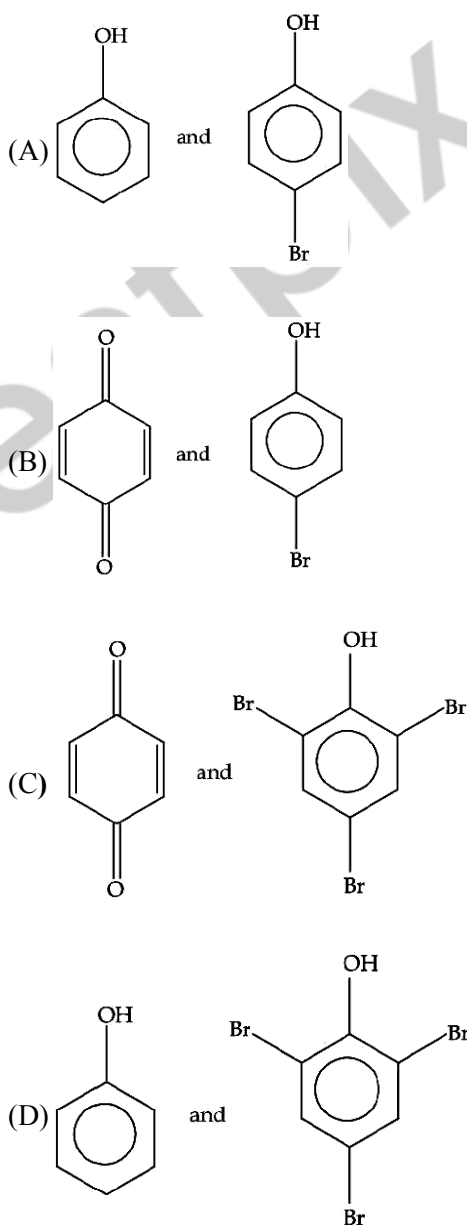
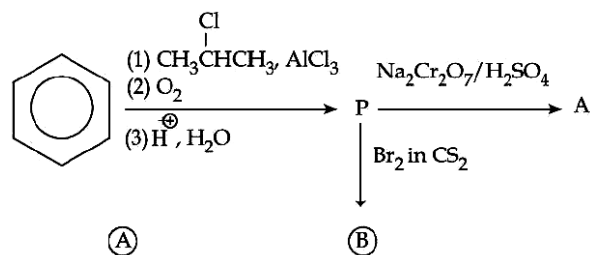
QuestPix Ans. (C)

Sol.



So it has least pK_a value.

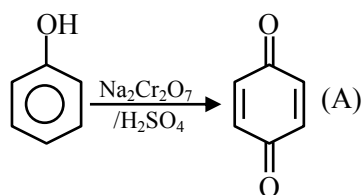
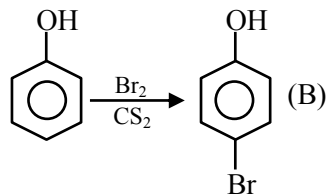
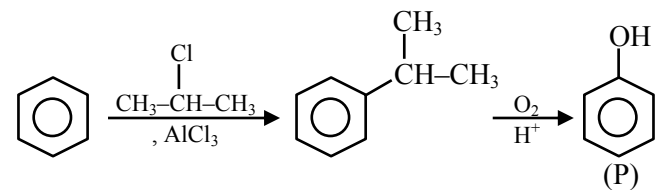
75. Identify the major product A and B for the below given reaction sequence.



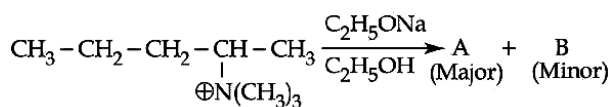
Official Ans. by NTA (B)

QuestPix Ans. (B)

Sol.



76. Identify the correct statement for the below given transformation.



(A) A - $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_3$,
B - $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}=\text{CH}_2$,
Saytzeff products

(B) A - $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_3$,
B - $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}=\text{CH}_2$,
Hafmann products

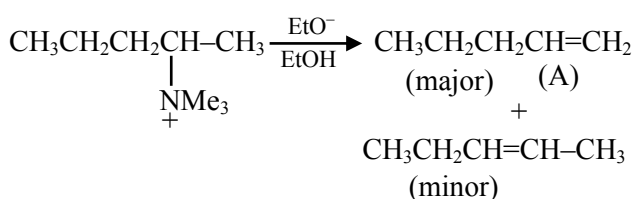
(C) A - $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}=\text{CH}_2$,
B - $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_3$,
Hofmann products

(D) A - $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}=\text{CH}_2$,
B - $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_3$,
Saytzeff products

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol.

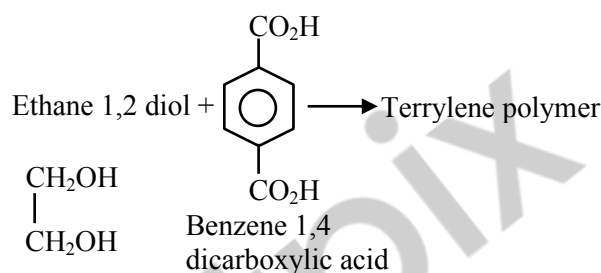


77. Terylene polymer is obtained by condensation of :
(A) Ethane-1, 2-diol and Benzene-1, 3 dicarboxylic acid
(B) Propane-1, 2-diol and Benzene-1, 4 dicarboxylic acid
(C) Ethane-1, 2-diol and Benzene-1, 4 dicarboxylic acid
(D) Ethane-1, 2-diol and Benzene-1, 2 dicarboxylic acid

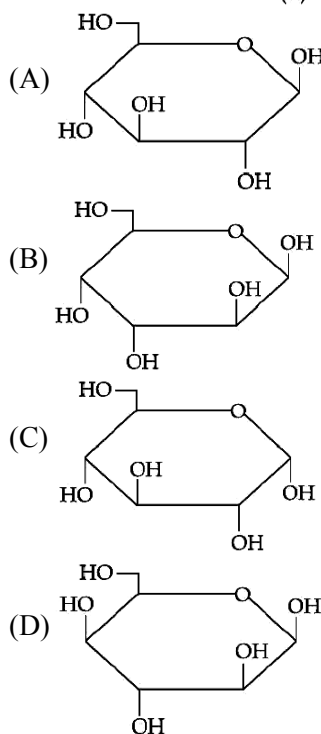
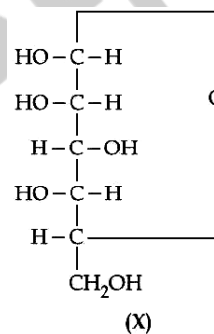
Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol.



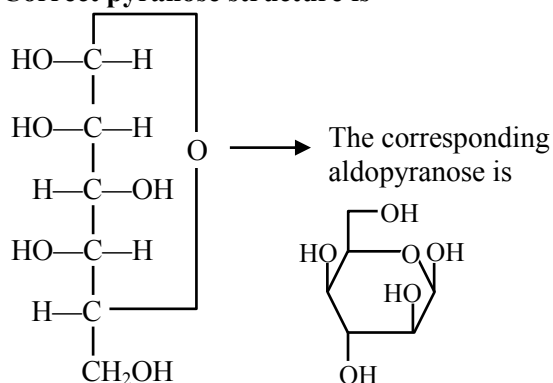
78. For the below given cyclic hemiacetal (X), the correct pyranose structure is :



Official Ans. by NTA (D)

QuestPix Ans. (D)

Sol. Correct pyranose structure is



X(Hemiacetal)

79. Statements about Enzyme Inhibitor Drugs are given below :

- (A) There are Competitive and Non-competitive inhibitor drugs.
- (B) These can bind at the active sites and allosteric sites.
- (C) Competitive Drugs are allosteric site blocking drugs.
- (D) Non-competitive Drugs are active site blocking drugs.

Choose the correct answer from the options given below :

- (A) (A), (D) only
- (B) (A), (C) only
- (C) (A), (B) only
- (D) (A), (B), (C) only

Official Ans. by NTA (C)

QuestPix Ans. (C)

Sol. Enzyme inhibitors can be competitive inhibitors (inhibit the attachment of substrate on active site of enzyme) and non-competitive inhibitor (changes the active site of enzyme after binding at allosteric site.)

80. For kinetic study of the reaction of iodide ion with H_2O_2 at room temperature :

- (A) Always use freshly prepared starch solution.
- (B) Always keep the concentration of sodium thiosulphate solution less than that of KI solution.
- (C) Record the time immediately after the appearance of blue colour.
- (D) Record the time immediately before the appearance of blue colour.
- (E) Always keep the concentration of sodium thiosulphate solution more than that of KI solution.

Choose the correct answer from the options given below :

- (A) (A), (B), (C) only
- (B) (A), (D), (E) only
- (C) (D), (E) only
- (D) (A), (B), (E) only

Official Ans. by NTA (A)

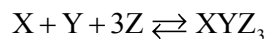
QuestPix Ans. (A)

Sol. The is recorded immediately after the blue colour appears.

$\text{Na}_2\text{S}_2\text{O}_3$ is kept in limited amount.

SECTION-B

81. In the given reaction,



if one mole of each of X and Y with 0.05 mol of Z gives compound XYZ_3 . (Given : Atomic masses of X, Y and Z are 10, 20 and 30 amu, respectively). The yield of XYZ_3 is _____ g.

(Nearest integer)

Official Ans. by NTA (2)

QuestPix Ans. (2)

Sol. $\text{X} + \text{Y} + 3\text{Z} \rightleftharpoons \text{XYZ}_3$

1mol 1mol 0.05mol

Z is L.R.

$$\frac{0.05}{3} = 1 \text{ mole of } \text{XYZ}_3$$

$$\text{Mass of } \text{XYZ}_3 = \frac{0.05}{3} \times (10 + 20 + 30 \times 3) = 2\text{g}$$

82. An element M crystallises in a body centred cubic unit cell with a cell edge of 300 pm. The density of the element is 6.0 g cm^{-3} . The number of atoms present in 180 g of the element is _____ $\times 10^{23}$.

(Nearest integer)

Official Ans. by NTA (22)

QuestPix Ans. (22)

Sol. M is body centred cubic, $\therefore Z = 2$

Let mass of 1 atom of M is A

Edge length = 300 pm

Density = 6 g/cm^3

$$\therefore 6 \text{ g/cm}^3 = \frac{Z \times A}{(300 \times 10^{-10})^3} = \frac{2 \times A}{27 \times 10^{-24}}$$

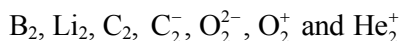
$$A = 81 \times 10^{-24} \text{ g}$$

$$\therefore \text{Atomic mass} = 48.6 \text{ g}$$

$$\therefore \text{Mole in } 180 \text{ g} = \frac{180}{48.6} = 3.7 \text{ moles}$$

$$\begin{aligned} \text{Atoms of M} &= 3.7 \times 6 \times 10^{23} \\ &= 22.22 \times 10^{23} \text{ atoms} \end{aligned}$$

83. The number of paramagnetic species among the following is _____.



Official Ans. by NTA (4)

QuestPix Ans. (4)

Sol. Paramagnetic $\text{B}_2, \text{C}_2^-, \text{O}_2^+, \text{He}_2^+$

84. 150 g of acetic acid was contaminated with 10.2 g ascorbic acid ($\text{C}_6\text{H}_8\text{O}_6$) to lower down its freezing point by $(x \times 10^{-1})^\circ\text{C}$. The value of x is _____.

(Nearest integer) [Given $K_f = 3.9 \text{ K kg mol}^{-1}$;

Molar mass of ascorbic acid = 176 g mol^{-1}]

Official Ans. by NTA (15)

QuestPix Ans. (15)

Sol. 150g CH_3COOH

10.2g ascorbic acid $\Rightarrow$ 0.058 moles

$$\Delta T_f = (x \times 10^{-1})^\circ\text{C}$$

$$\Delta T_f = k_f \cdot \text{molality}$$

$$= 3.9 \times \frac{0.058}{150} \times 1000$$

$$= 1.5^\circ\text{C}$$

$$= 15 \times 10^{-1}^\circ\text{C}$$

85. K_a for butyric acid ($\text{C}_3\text{H}_7\text{COOH}$) is 2×10^{-5} . The pH of 0.2 M solution of butyric acid is $___ \times 10^{-1}$.

(Nearest integer) [Given $\log 2 = 0.30$]

Official Ans. by NTA (27)

QuestPix Ans. (27)

Sol. K_a of Butyric acid $\Rightarrow 2 \times 10^{-5}$ $\text{PK}_a = 4.7$

pH of 0.2 M solution

$$\text{pH} = \frac{1}{2} \text{pK}_a - \frac{1}{2} \log C$$

$$= \frac{1}{2} (4.7) - \frac{1}{2} \log (0.2)$$

$$= 2.35 + 0.35 = 2.7$$

$$\text{pH} = 27 \times 10^{-1}$$

86. For the given first order reaction



the half life of the reaction is 0.3010 min. The ratio of the initial concentration of reactant to the concentration of reactant at time 2.0 min will be equal to _____. (Nearest integer)

Official Ans. by NTA (100)

QuestPix Ans. (100)

Sol. $\text{A} \rightarrow \text{B}$ $t_{1/2} = 0.3010 \text{ min}$

$$A_0/A_t \text{ at time 2 min} = ?$$

$$K = \frac{2.303}{t} \log \left[\frac{A_0}{A_t} \right]$$

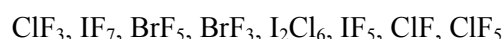
$$\Rightarrow \frac{0.693}{t_{1/2}} = \frac{2.303}{2} \log \left(\frac{A_0}{A_t} \right)$$

$$\text{Or } \frac{2.303 \times 0.3010}{0.3010} = \frac{2.303}{2} \log \frac{A_0}{A_t}$$

$$\log \frac{A_0}{A_t} = 2$$

$$\therefore \frac{A_0}{A_t} = 10^2 = 100$$

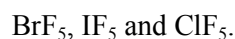
87. The number of interhalogens from the following having square pyramidal structure is :



Official Ans. by NTA (3)

QuestPix Ans. (3)

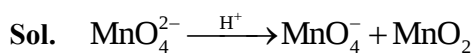
Sol. Square pyramidal structures are



88. The disproportionation of MnO_4^{2-} in acidic medium resulted in the formation of two manganese compounds A and B. If the oxidation state of Mn in B is smaller than that of A, then the spin-only magnetic moment (μ) value of B in BM is _____. (Nearest integer)

Official Ans. by NTA (4)

QuestPix Ans. (4)



No. of unpaired $\bar{e} = 3$

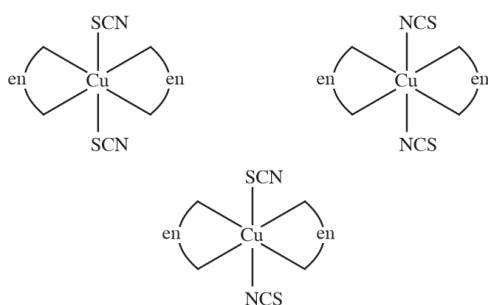
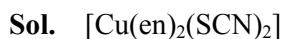
$$\therefore \mu = \sqrt{15} = 3.877$$

Nearest Integer = 4

- 89.** Total number of relatively more stable isomer(s) possible for octahedral complex $[\text{Cu}(\text{en})_2(\text{SCN})_2]$ will be _____.

Official Ans. by NTA (3)

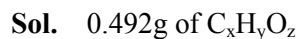
QuestPix Ans. (3)



- 90.** On complete combustion of 0.492 g of an organic compound containing C, H and O, 0.7938 g of CO_2 and 0.4428 g of H_2O was produced. The % composition of oxygen in the compound is _____.

Official Ans. by NTA (46)

QuestPix Ans. (46)



Gives 0.7938 g $\text{CO}_2 = 0.018$ moles

0.4428g $\text{H}_2\text{O} = 0.0246$ moles

So moles of C = 0.018 $\Rightarrow$ 0.216 g

Moles of H = 0.049 $\Rightarrow$ 0.049g

$$\therefore \text{wt. of Oxygen} = 0.492 - 0.216 - 0.049 = 0.227\text{g}$$

$$\% \text{ of Oxygen} = \frac{0.227}{0.492} \times 100 = 46 \text{ (approx.)}$$