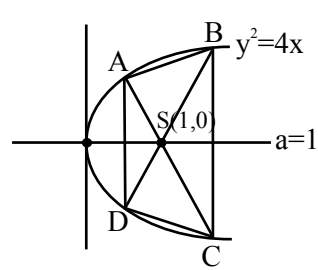


(HELD ON TUESDAY 28th JANUARY 2025)
TIME : 9:00 AM TO 12:00 NOON

MATHEMATICS	TEST PAPER WITH SOLUTION
SECTION-A 1. The number of different 5 digit numbers greater than 50000 that can be formed using the digits 0, 1, 2, 3, 4, 5, 6, 7, such that the sum of their first and last digits should not be more than 8, is (1) 4608 (2) 5720 (3) 5719 (4) 4607 Ans. (4) Sol. Case I 5 _ _ _ 0 Case II 5 _ _ _ 1 5 _ _ _ 2 5 _ _ _ 3 6 _ _ _ 0 6 _ _ _ 1 6 _ _ _ 2 7 _ _ _ 0 7 _ _ _ 1 $9 \times (8 \times 8 \times 8) = 4608$ but 50000 is not included, so total numbers $4608 - 1 = 4607$ 2. Let ABCD be a trapezium whose vertices lie on the parabola $y^2 = 4x$. Let the sides AD and BC of the trapezium be parallel to y-axis. If the diagonal AC is of length $\frac{25}{4}$ and it passes through the point (1, 0), then the area of ABCD is : (1) $\frac{75}{4}$ (2) $\frac{25}{2}$ (3) $\frac{125}{8}$ (4) $\frac{75}{8}$ Ans. (1) Sol. 	$A(at_1^2, 2at_1) \& C\left(\frac{a}{t_1^2}, -\frac{2a}{t_1}\right)$ Length AC = $a\left(t_1 - \frac{1}{t_1}\right)^2 = \frac{25}{4}$, $t_1 + \frac{1}{t_1} = \pm \frac{5}{2}$ $\Rightarrow t_1 = 2$ or $\frac{1}{2}$, $A\left(\frac{1}{2}, 1\right)$, $D\left(\frac{1}{4}, -1\right)$, $B(4, 4)$, $C(4, -4)$ So, area of trapezium = $\frac{1}{2}(8+2)\left(4 - \frac{1}{4}\right) = \frac{75}{4}$ 3. Two number k_1 and k_2 are randomly chosen from the set of natural numbers. Then, the probability that the value of $i^{k_1} + i^{k_2}$, ($i = \sqrt{-1}$) is non-zero, equals (1) $\frac{1}{2}$ (2) $\frac{1}{4}$ (3) $\frac{3}{4}$ (4) $\frac{2}{3}$ Ans. (3) Sol. $i^{-1} + i^k = 0 \Rightarrow i^{k_1} \rightarrow 4$ option for $i, -1, -i, 1$ Total cases $\Rightarrow 4 \times 4 = 16$ Unfavourable cases $\Rightarrow i^{k_1} + i^{k_2} = 0$ $\left\{ \begin{array}{l} 1, -1 \\ -1, 1 \\ i, -i \\ -i, i \end{array} \right\}$ 4 Cases $\Rightarrow$ Probability = $\frac{16-4}{16} = \frac{3}{4}$ 4. If $f(x) = \frac{2^x}{2^x + \sqrt{x}}$, $x \in \mathbb{R}$, then $\sum_{k=1}^{81} f\left(\frac{k}{82}\right)$ is equal to : (1) 41 (2) $\frac{81}{2}$ (3) 82 (4) $81\sqrt{2}$ Ans. (2)

Sol. $f(x) = \frac{2^x}{2^x + \sqrt{2}}$

$$f(x) + f(1-x) = \frac{2^x}{2^x + \sqrt{2}} + \frac{2^{1-x}}{2^{1-x} + \sqrt{2}}$$

$$= \frac{2^x}{2^x + \sqrt{2}} + \frac{2}{2 + \sqrt{2} 2^x} = \frac{2^x + \sqrt{2}}{2^x + \sqrt{2}} = 1$$

Now, $\sum_{k=1}^{81} f\left(\frac{k}{82}\right) = f\left(\frac{1}{82}\right) + f\left(\frac{2}{82}\right) + \dots + f\left(\frac{81}{82}\right)$

$$= f\left(\frac{1}{82}\right) + f\left(\frac{1}{82}\right) + \dots + f\left(1 - \frac{2}{82}\right) + f\left(1 - \frac{1}{82}\right)$$

$$\left[f\left(\frac{1}{82}\right) + f\left(1 - \frac{1}{82}\right) \right] + \left[f\left(\frac{2}{82}\right) + f\left(1 - \frac{2}{82}\right) \right] + \dots + 40 \text{ cases} + f\left(\frac{41}{82}\right)$$

$$(1+1+\dots+1) 40 \text{ times} + \frac{2^{1/2}}{2^{1/2} + 2^{1/2}}$$

$$40 + \frac{1}{2} = \frac{81}{2}$$

5. Let $f : \mathbb{R} \rightarrow \mathbb{R}$ be a function defined by

$$f(x) = (2+3a)x^2 + \left(\frac{a+2}{a-1}\right)x + b, a \neq 1. \text{ If}$$

$$f(x+y) = f(x) + f(y) + 1 - \frac{2}{7}xy, \text{ then the value of}$$

$$28 \sum_{i=1}^5 |f(i)| \text{ is:}$$

(1) 715

(2) 735

(3) 545

(4) 675

Ans. (4)

Sol. $f(x) = (3a+2)x^2 + \left(\frac{a+2}{a-1}\right)x + b$

$$f\left(x + \frac{1}{2}\right) = f(x) + f\left(\frac{1}{2}\right) + 1 - \frac{2}{7}xy \dots (1)$$

In (1) Put $x = y = 0 \Rightarrow f(0) = 2f(0) + 1 \Rightarrow f(0) = -1$

So, $f(0) = 0 + 0 + b = -1 \Rightarrow b = -1$

In (1) Put $y = -x \Rightarrow f(0) = f(x) + f(-x) + 1 + \frac{2}{7}x^2$

$$-1 = 2(3a+2)x^2 + 2b + 1 + \frac{2}{7}x^2$$

$$-1 = \left(2(3a+2) + \frac{2}{7}\right)x^2 + 1 - 2$$

$$\Rightarrow 6a + 4 + \frac{2}{7} = 0$$

$$a = -\frac{5}{7}$$

So $f(x) = -\frac{1}{7}x^2 - \frac{3}{4}x - 1$

$$\Rightarrow |f(x)| = \frac{1}{28} |4x^2 + 21x + 28|$$

Now, $28 \sum_{i=1}^5 |f(6)| = 28(|f(1)| + f(2) + \dots + f(5))$

$$28 \cdot \frac{1}{28} \cdot 675 = 675$$

6. Let $A(x, y, z)$ be a point in xy -plane, which is equidistant from three points $(0, 3, 2)$, $(2, 0, 3)$ and $(0, 0, 1)$.

Let $B = (1, 4, -1)$ and $C = (2, 0, -2)$. Then among the statements

(S1) : ΔABC is an isosceles right angled triangle and

(S2) : the area of ΔABC is $\frac{9\sqrt{2}}{2}$.

(1) both are true

(2) only (S1) is true

(3) only (S2) is true

(4) both are false

Ans. (2)

Sol. $A(x, y, z)$ Let $P(0, 3, 2)$, $Q(2, 0, 3)$, $R(0, 0, 1)$

$$AP = AQ = AR$$

$$x^2 + (y-3)^2 + (z-2)^2 = (x-2)^2 + y^2 + (z-3)^2 = x^2 + y^2 + (z-1)^2$$

In xy plane $z = 0$

$$\text{So, } x^2 - 4x + 4 + y^2 + 9 = x^2 + y^2 + 1$$



$$\Rightarrow a_n = -1 + \left(\frac{3}{2}\right)^n$$

$$\sum_{k=1}^{100} a_k = \sum_{k=1}^{100} (-1) + \left(\frac{3}{2}\right)^k$$

$$= -100 + \frac{\left(\frac{3}{2}\right)\left(\left(\frac{3}{2}\right)^{100} - 1\right)}{\frac{3}{2} - 1}$$

$$= -100 + 3\left(\left(\frac{3}{2}\right)^{100} - 1\right)$$

$$= 3(a_{100}) - 100$$

10. $\cos\left(\sin^{-1}\frac{3}{5} + \sin^{-1}\frac{5}{13} + \sin^{-1}\frac{33}{65}\right)$ is equal to :

- (1) 1 (2) 0
(3) $\frac{33}{65}$ (4) $\frac{32}{65}$

Ans. (2)

Sol. $\cos\left(\sin^{-1}\frac{3}{5} + \sin^{-1}\frac{5}{13} \quad \sin^{-1}\frac{33}{65}\right)$

$$\cos\left(\tan^{-1}\frac{3}{4} + \tan^{-1}\frac{5}{12} \quad \tan^{-1}\frac{33}{56}\right)$$

$$\cos\left(\tan^{-1}\left(\frac{\frac{3}{4} + \frac{5}{12}}{1 + \frac{3}{4} \cdot \frac{5}{12}}\right) + \tan^{-1}\frac{33}{56}\right)$$

$$\cos\left(\tan^{-1}\frac{56}{33} + \cot^{-1}\frac{56}{33}\right)$$

$$\cos\left(\frac{\pi}{2}\right) = 0$$

11. Let T_r be the r^{th} term of an A.P. If for some m , $T_m = \frac{1}{25}$, $T_{25} = \frac{1}{20}$ and $20 \sum_{r=1}^{25} T_r = 13$, then

$$5m \sum_{r=m}^{2m} T_r \text{ is equal to :}$$

- (1) 112 (2) 126
(3) 98 (4) 142

Ans. (2)

Sol. $T_m = \frac{1}{25}$, $T_{25} = \frac{1}{20}$, $20 \sum_{r=1}^{25} T_r = 13$

$$T_m = a + (m-1)d = \frac{1}{25} \quad \dots\dots(1)$$

$$T_{25} = a + 24d = \frac{1}{20}$$

$$20 \cdot \frac{25}{2} \left[a + \frac{1}{20} \right] = 13 \Rightarrow a = \frac{1}{500}$$

$$\text{also, } 20S_{25} = 20 \cdot \frac{25}{2} [2a + 24d] = 13 \Rightarrow d = \frac{1}{500}$$

$$\text{from (1)} \quad \frac{1}{500} + \frac{m-1}{500} = \frac{1}{25} \Rightarrow m = 20$$

Now,

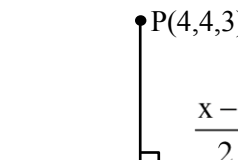
$$5m \sum_{r=m}^{2m} T_r = 100 \sum_{r=20}^{40} T_r = 126$$

12. If the image of the point (4, 4, 3) in the line $\frac{x-1}{2} = \frac{y-2}{1} = \frac{z-1}{3}$ is (α, β, γ) , then $\alpha + \beta + \gamma$ is

equal to

- (1) 9 (2) 12
(3) 8 (4) 7

Ans. (1)

Sol. 
 $\frac{x-1}{2} = \frac{y-2}{1} = \frac{z-1}{3}$
 $Q(2\lambda+1, \lambda+2, 3\lambda+11)$

$$\overrightarrow{PQ} \perp (2\hat{i} + \hat{j} + 3\hat{k})$$

$$\Rightarrow 2(2\lambda - 3) + 1(\lambda - 2) + 3(3\lambda - 2) = 0$$

$$\Rightarrow 14\lambda - 14 = 0, \lambda = 1$$

So, $Q(3, 3, 4)$

Let image in $R(\alpha, \beta, \gamma)$

$$\frac{\alpha + \gamma}{2} = 3, \frac{\beta + \gamma}{2} = 3, \frac{\gamma + 3}{2} = 4$$

$$(\alpha, \beta, \gamma) = (2, 2, 5)$$

$$\Rightarrow \alpha + \beta + \gamma = 9$$

13. If $\int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} \frac{96x^2 \cos^2 x}{(1+e^x)} dx = \pi(\alpha\pi^2 + \beta)$, $\alpha, \beta \in \mathbb{Z}$, then

$(\alpha + \beta)^2$ equals :

- (1) 144 (2) 196
(3) 100 (4) 64

Ans. (3)

Sol. $\int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} \frac{96x^2 \cos^2 x}{(1+e^x)} dx$ (Apply King Property)

$$\int_0^{\frac{\pi}{2}} 96x^2 \cos^2 x = 48 \int_0^{\frac{\pi}{2}} x^2 (1 + \cos 2x) dx$$

$$48 \left[\left(\frac{x^3}{3} \right)_0^{\frac{\pi}{2}} + \int_0^{\frac{\pi}{2}} x^2 \cos 2x dx \right]$$

$\Rightarrow$ On solving $\pi(2\pi^2 - 12)$

$\Rightarrow \alpha = 2, \beta = -12$

$\Rightarrow (\alpha + \beta)^2 = 100$

14. The sum of all local minimum values of the

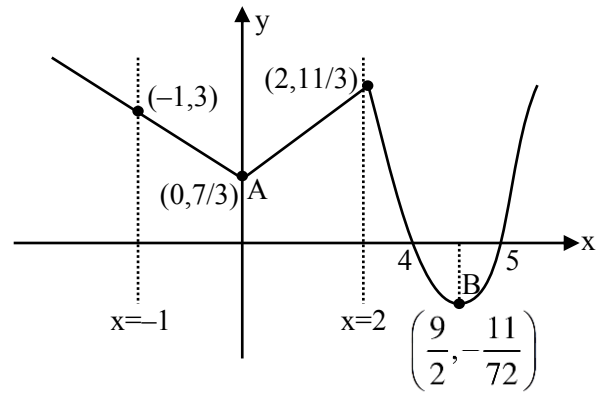
$$\text{function } f(x) = \begin{cases} 1 - 2x, & x < -1 \\ \frac{1}{3}(7 + 2|x|), & -1 \leq x \leq 2 \\ \frac{11}{18}(x-4)(x-5), & x > 2 \end{cases}$$

(1) $\frac{171}{72}$ (2) $\frac{131}{72}$

(3) $\frac{157}{72}$ (4) $\frac{167}{72}$

Ans. (3)

Sol. $f(x) = \begin{cases} 1 - 2x, & x < -1 \\ \frac{1}{3}(7 - 2x), & -1 \leq x \leq 2 \\ \frac{1}{3}(7 + 2x), & 0 \leq x < 2 \\ \frac{11}{18}(x-4)(x-5), & x > 2 \end{cases}$



$\therefore$ Local minimum values at A & B

$$\frac{7}{3} - \frac{11}{72}$$

$$\Rightarrow \frac{168 - 11}{72} \Rightarrow \frac{157}{72}$$

15. The sum, of the squares of all the roots of the equation $x^2 + |2x - 3| - 4 = 0$, is :

(1) $3(3 - \sqrt{2})$ (2) $6(3 - \sqrt{2})$

(3) $6(2 - \sqrt{2})$ (4) $3(2 - \sqrt{2})$

Ans. (3)

Sol. $x^2 + |2x - 3| - 4 = 0$

Case I : $x \geq \frac{3}{2}$

$$x^2 + 2x - 3 - 4 = 0$$

$$x^2 + 2x - 7 = 0$$

$$x = 2\sqrt{2} - 1$$

Case II : $x < \frac{3}{2}$

$$x^2 + 3 - 2x - 4 = 0$$

$$x^2 - 2x - 1 = 0$$

$$x = 1 - \sqrt{2}$$

$$\text{Sum of squares} = (2\sqrt{2} - 1)^2 + (1 - \sqrt{2})^2$$

$$= 8 - 4\sqrt{2} + 1 + 1 - 2\sqrt{2} + 2$$

$$= 6(2 - \sqrt{2}) \quad \therefore (3)$$

16. Let for some function $y = f(x)$, $\int_0^x t f(t) dt = x^2 f(x)$,

$x > 0$ and $f(2) = 3$. Then $f(6)$ is equal to :

- (1) 1 (2) 2
(3) 6 (4) 3

Ans. (1)

Sol. $\int_0^x t f(t) dt = x^2 + (x)$, $x > 0$

Diff. both side w.r. to x

$$x f(x) = x^2 f'(x) + 2x f(x)$$

$$-x f(x) = x^2 f''(x)$$

$$\frac{f'(x)}{f(x)} dx = \int \frac{-}{2} dx$$

$$\log f(x) = -\log x + \log c$$

$$f(x) = \frac{c}{x}$$

$$f(2) = 3 \Rightarrow 3 = \frac{c}{2} \Rightarrow c = 6$$

$$f(x) = \frac{6}{x}$$

$$f(6) = 1 \quad \therefore (1)$$

17. Let ${}^nC_{r-1} = 28$, ${}^nC_r = 56$ and ${}^nC_{r+1} = 70$. Let $A(4\cos t, 4\sin t)$, $B(2\sin t, -2\cos t)$ and $C(3r-n, r^2-n-1)$ be the vertices of a triangle ABC, where t is a parameter. If $(3x-1)^2 + (3y)^2 = \alpha$, is the locus of the centroid of triangle ABC, then α equals :

- (1) 20 (2) 8
(3) 6 (4) 18

Ans. (1)

Sol. ${}^nC_{r-1} = 28$, ${}^nC_r = 56$

$$\frac{{}^nC_{r-1}}{{}^nC_r} = \frac{28}{56}$$

$$\frac{\frac{n!}{(r-1)!(n-r+1)!}}{\frac{n!}{r!(n-r)!}} = \frac{1}{2}$$

$$\frac{r}{(n-r+1)} = \frac{1}{2}$$

$$3r = n+1 \quad \text{---(i)}$$

$$\frac{{}^nC_r}{{}^nC_{r+1}} = \frac{56}{70}$$

$$\frac{(r+1)}{(n-r)} = \frac{56}{70} \Rightarrow 9r = 4n-5 \quad \text{---(ii)}$$

By (i) & (ii)

$$(r=3), (n=8)$$

$$A(4\cos t, 4\sin t) \quad B(2\sin t, -2\cos t) \quad C(3r-n, r^2-n-1)$$

$$A(4\cos t, 4\sin t) \quad B(2\sin t, -2\cos t) \quad C(1, 0)$$

$$(3x-1)^2 + (3y)^2 = (4\cos t + 2\sin t)^2 + (4\sin t - \cos t)^2$$

$$(3x-1)^2 + (3y)^2 = 20 \quad \therefore (1)$$

18. Let O be the origin, the point A be $z_1 = \sqrt{3} + 2\sqrt{2}i$, the point $B(z_2)$ be such that

$$\sqrt{3}|z_2| = |z_1| \quad \text{and} \quad \arg(z_2) = \arg(z_1) + \frac{\pi}{6}. \text{ Then}$$

$$(1) \text{ area of triangle } ABO \text{ is } \frac{11}{\sqrt{3}}$$

$$(2) ABO \text{ is a scalene triangle}$$

$$(3) \text{ area of triangle } ABO \text{ is } \frac{11}{4}$$

$$(4) ABO \text{ is an obtuse angled isosceles triangle}$$

Ans. (4)

Sol. $z_1 = \sqrt{3} + 2\sqrt{2}i$ & $\frac{|z_2|}{|z_1|} = \frac{1}{\sqrt{3}}$

$$\text{given } \arg\left(\frac{z_2}{z_1}\right) = \frac{\pi}{6}$$

$$z_2 = \frac{|z_2|}{|z_1|} \cdot z_1 \cdot e^{i\left(\frac{\pi}{6}\right)}$$

$$z_2 = \frac{1}{\sqrt{3}} \cdot \frac{(\sqrt{3} + 2\sqrt{2}i)(\sqrt{3} + i)}{2}$$

$$z_2 = \frac{(3 - 2\sqrt{2}) + i(2\sqrt{6} + \sqrt{3})}{2\sqrt{3}}$$

Now,

$$z_1 - z_2 = \frac{(3 + 2\sqrt{2}) + i(2\sqrt{6} - \sqrt{3})}{2\sqrt{3}}$$

$|z_1 - z_2| = |z_2| \Rightarrow \Delta ABO$ is isosceles with angles

$$\frac{\pi}{6}, \frac{\pi}{6} \& \frac{2\pi}{3}$$

19. Three defective oranges are accidentally mixed with seven good ones and on looking at them, it is not possible to differentiate between them. Two oranges are drawn at random from the lot. If x denote the number of defective oranges, then the variance of x is :

- (1) $28/75$ (2) $14/25$
(3) $26/75$ (4) $18/25$

Ans. (1)

Sol. 10 oranges $\begin{cases} 7 \text{ good} \\ 3 \text{ defected} \end{cases}$

Probability distribution

x_i	p_i
$x = 0$	$\frac{7C_2}{10C_2} = \frac{42}{90}$
$x = 1$	$\frac{7C_1 \times 3C_1}{10C_2} = \frac{42}{90}$
$x = 2$	$\frac{3C_2}{10C_2} = \frac{6}{90}$

Now,

$$\mu = \sum x_i p_i = \frac{42}{90} + \frac{12}{90} = \frac{54}{90}$$

$$\sigma^2 = \sum p_i x_i^2 - \mu^2 = \frac{42}{90} + \frac{24}{90} - \left(\frac{54}{90}\right)^2$$

$$\Rightarrow \frac{66}{90} - \left(\frac{54}{90}\right)^2$$

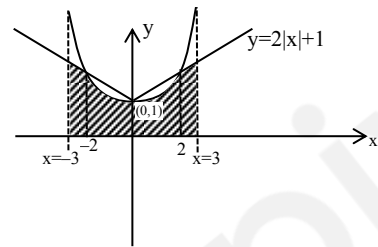
$$\sigma^2 \Rightarrow \frac{28}{75} \therefore (1)$$

20. The area (in sq. units) of the region

$$\{(x, y): 0 \leq y \leq 2|x| + 1, 0 \leq y \leq x^2 + 1, |x| \leq 3\}$$

- (1) $\frac{80}{3}$ (2) $\frac{64}{3}$
(3) $\frac{17}{3}$ (4) $\frac{32}{3}$

Ans. (2)



Sol.

$$\text{Area} = 2 \left[\int_0^3 (x^2 + 1) dx - \int_0^3 (2x + 1) dx \right]$$

$$\Rightarrow \frac{64}{3} \therefore (2)$$

SECTION-B

21. Let M denote the set of all real matrices of order 3×3 and let $S = \{-3, -2, -1, 1, 2\}$. Let
 $S_1 = \{A = [a_{ij}] \in M : A = A^T \text{ and } a_{ij} \in S, \forall i, j\}$
 $S_2 = \{A = [a_{ij}] \in M : A = -A^T \text{ and } a_{ij} \in S, \forall i, j\}$
 $S_3 = \{A = [a_{ij}] \in M : a_{11} + a_{22} + a_{33} = 0 \text{ and } a_{ij} \in S, \forall i, j\}$
 If $n(S_1 \cup S_2 \cup S_3) = 125\alpha$, then α equals.

Ans. (1613)

$$\text{Sol. } \begin{bmatrix} a_{11} & a_{12} & a_{13} \\ a_{21} & a_{22} & a_{23} \\ a_{31} & a_{32} & a_{33} \end{bmatrix}$$

No. of elements in $S_1 : A = A^T \Rightarrow 5^3 \times 5^3$

No. of elements in $A = -A^T \Rightarrow 0$
 since no. zero in S

No. of elements in $S_3 \Rightarrow$

$$\left. \begin{aligned} a_{11} + a_{22} + a_{33} = 0 &\Rightarrow (1, 2, -3) \Rightarrow 31 \\ &\text{or} \\ &(1, 1, -2) \Rightarrow 3 \\ &\text{or} \\ &(-1, -1, 2) \Rightarrow 3 \end{aligned} \right\} \Rightarrow 12 \times 5^6$$

$$n(S_1 \cap S_3) = 12 \times 5^3$$

$$n(S_1 \cup S_2 \cup S_3) = 5^6(1+12) - 12 \times 5^3$$

$$\Rightarrow 5^3 \times [13 \times 5^3 - 12] = 125\alpha$$

$$\alpha = 1613$$

22. If $\alpha = 1 + \sum_{r=1}^6 (-3)^{r-1} {}^{12}C_{2r-1}$, then the distance of the point $(12, \sqrt{3})$ from the line $\alpha x - \sqrt{3}y + 1 = 0$ is

Ans. (5)

Sol. $\alpha = 1 + \sum_{r=1}^6 (-)^{r-1} {}^{12}C_{2r-1} 3^{r-1}$

$$\alpha = 1 + \sum_{r=1}^6 {}^{12}C_{2r-1} \frac{(\sqrt{3}i)^{2r-1}}{\sqrt{3}i} \quad i = \text{iota, let } \sqrt{3}i = x$$

$$\alpha = 1 + \frac{1}{\sqrt{3}i} ({}^{12}C_1 x + {}^{12}C_3 x^3 + \dots + {}^{12}C_{11} x^{11})$$

$$= 1 + \frac{1}{\sqrt{3}i} \left(\frac{(1 + \sqrt{3}i)^{12} - (1 - \sqrt{3}i)^{12}}{2} \right)$$

$$= 1 + \frac{1}{\sqrt{3}i} \left(\frac{(-2w^2)^{12} - (2w)^{12}}{2} \right) = 1$$

so distance of $(12, \sqrt{3})$ from $x - \sqrt{3}y + 1 = 0$ is

$$\frac{12 - 3 + 1}{2} = 5$$

23. Let $\vec{a} = \hat{i} + \hat{j} + \hat{k}$, $\vec{b} = 2\hat{i} + 2\hat{j} + \hat{k}$ and $\vec{d} = \vec{a} \times \vec{b}$. If $\vec{c}$ is a vector such that $\vec{a} \cdot \vec{c} = |\vec{c}|$, $|\vec{c} - 2\vec{a}|^2 = 8$ and the angle between $\vec{d}$ and $\vec{c}$ is $\frac{\pi}{4}$, then

$|10 - 3\vec{b} \cdot \vec{c}| + |\vec{d} \times \vec{c}|^2$ is equal to

Ans. (6)

Sol. $\vec{a} = \hat{i} + \hat{j} + \hat{k}$
 $\vec{b} = 2\hat{i} + 2\hat{j} + \hat{k}$
 $\vec{d} = \vec{a} \times \vec{b}$
 $= -\hat{i} + \hat{j}$
 $|\vec{c} - 2\vec{a}|^2 = 8$
 $|\vec{c}|^2 + 4|\vec{a}|^2 - 4(\vec{a} \cdot \vec{c}) = 8$
 $c^2 + 12 - 4c = 8$
 $c^2 - 4c + 4 = 0$
 $|c| = 2$
 $\vec{d} = \vec{a} \times \vec{b}$
 $\vec{d} \times \vec{c} = (\vec{a} \times \vec{b}) \times \vec{c}$

$$\left(|a||c| \sin \frac{\pi}{4} \right)^2 = ((a \cdot c) \cdot b - (b \cdot c) \cdot a)^2$$

$$4 = 4b^2 + (b \cdot c)2(a^2) - 2(b \cdot c)(a \cdot b)$$

$$4 = 36 + 3x^2 - 20x$$

$$\text{Let } b \cdot c = x$$

$$3x^2 - 20x + 32 = 0$$

$$3x^2 - 12x - 8x + 32 = 0$$

$$x = \frac{8}{3}, 4; \quad b \cdot c = \frac{8}{3}, 4; \quad b \cdot c = \frac{8}{3}$$

$$\text{Now } |10 - 3b \cdot c| + |d \times c|^2; |10 - 8| + (2)^2 \Rightarrow 6 \text{ Ans.}$$

24. Let

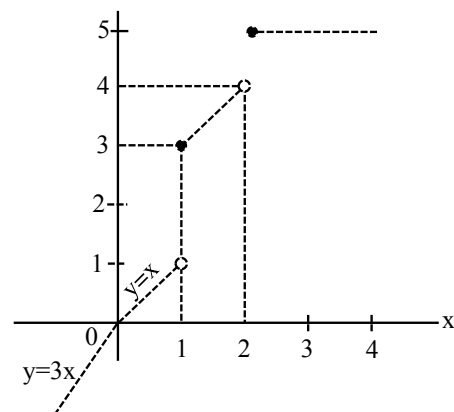
$$f(x) = \begin{cases} 3x, & x < 0 \\ \min\{1 + x + [x], x + 2[x]\}, & 0 \leq x \leq 2 \\ 5, & x > 2 \end{cases}$$

where $[.]$ denotes greatest integer function. If α and β are the number of points, where f is not continuous and is not differentiable, respectively, then $\alpha + \beta$ equals.....

Ans. (5)

Sol. $f(x) = \begin{cases} 3x & ; \quad x < 0 \\ \min\{1 + x, x\} & ; \quad 0 \leq x < 1 \\ \min\{2 + x, x + 2\} & ; \quad 1 \leq x < 2 \\ 5 & ; \quad x > 2 \end{cases}$

$$f(x) = \begin{cases} 3x & ; \quad x < 0 \\ x & ; \quad 0 \leq x < 1 \\ x + 2 & ; \quad 1 \leq x < 2 \\ 5 & ; \quad x > 2 \end{cases}$$



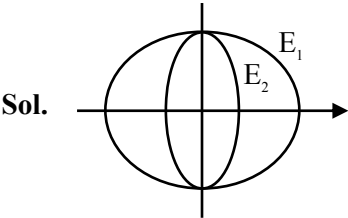
Not continuous at $x \in \{1, 2\} \Rightarrow \alpha = 2$

Not diff. at $x \in \{0, 1, 2\} \Rightarrow \beta = 3$

$$\alpha + \beta = 5$$

25. Let $E_1 : \frac{x^2}{9} + \frac{y^2}{4} = 1$ be an ellipse. Ellipses E_i 's are constructed such that their centres and eccentricities are same as that of E_1 , and the length of minor axis of E_i is the length of major axis of E_{i+1} ($i \geq 1$). If A_i is the area of the ellipse E_i , then $\frac{5}{\pi} \left(\sum_{i=1}^{\infty} A_i \right)$, is equal to

Ans. (54)



$$E_1 : \frac{x^2}{9} + \frac{y^2}{4} = 1 \Rightarrow e = \sqrt{1 - \frac{4}{9}} = \frac{\sqrt{5}}{3}$$

$$E_2 : \frac{x^2}{a^2} + \frac{y^2}{4} = 1$$

$$= \frac{\sqrt{5}}{3} = \sqrt{1 - \frac{a^2}{4}} \Rightarrow \frac{5}{9} = 1 - \frac{a^2}{4}$$

$$a^2 = \frac{16}{9}$$

$$E_2 : \frac{x^2}{16} + \frac{y^2}{4} = 1$$

$$E_3 : \frac{x^2}{16} + \frac{y^2}{b^2} = 1$$

$$e = \frac{\sqrt{5}}{3} = \sqrt{1 - \frac{b^2}{16}} \Rightarrow b^2 = \frac{64}{81}$$

$$E_3 : \frac{x^2}{16} + \frac{y^2}{64} = 1$$

$$A_1 = \pi \times 3 \times 2 \Rightarrow 6\pi$$

$$A_2 = \pi \times \frac{4}{3} \times 2 = \frac{8\pi}{3}$$

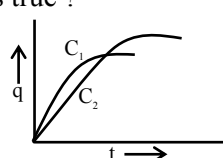
$$A_3 = \pi \times \frac{4}{3} \times \frac{8}{9} = \frac{32\pi}{81}$$

$$\sum_{i=1}^{\infty} A_i = 6\pi + \frac{8\pi}{3} + \frac{32\pi}{81} + \dots \infty$$

$$\Rightarrow \frac{6\pi}{1 - \frac{4}{9}} = \frac{54\pi}{5}$$

$$\therefore \frac{5}{\pi} \sum_{i=1}^{\infty} A_i \Rightarrow \frac{5}{\pi} \times \frac{54\pi}{5} = 54$$

(HELD ON TUESDAY 28th JANUARY 2025)
TIME : 9:00 AM TO 12:00 NOON

PHYSICS	TEST PAPER WITH SOLUTION
SECTION-A 26. Two capacitors C_1 and C_2 are connected in parallel to a battery. Charge-time graph is shown below for the two capacitors. The energy stored with them are U_1 and U_2, respectively. Which of the given statements is true ?  (1) $C_1 > C_2$, $U_1 > U_2$ (2) $C_2 > C_1$, $U_2 < U_1$ (3) $C_1 > C_2$, $U_1 < U_2$ (4) $C_2 > C_1$, $U_2 > U_1$ Ans. (4) Sol. potential difference, $v \rightarrow$ same $U = \frac{1}{2} cv^2$ as $q_1 < q_2$ $\therefore c_1 < c_2$ & $U_1 < U_2$ 27. In the experiment for measurement of viscosity 'η' of given liquid with a ball having radius R, consider following statements. A. Graph between terminal velocity V and R will be a parabola B. The terminal velocities of different diameter balls are constant for a given liquid. C. Measurement of terminal velocity is dependent on the temperature. D. This experiment can be utilized to assess the density of a given liquid. E. If balls are dropped with some initial speed, the value of η will change. Choose the correct answer from the options given below: (1) B, D and E only (2) A, C and D only (3) C, D and E only (4) A, B and E only Ans. (2)	Sol. $V_T = \frac{2}{9} R^2 \frac{g}{\eta} (d - \rho)$ 28. Consider following statements: A. Surface tension arises due to extra energy of the molecules at the interior as compared to the molecules at the surface, of a liquid. B. As the temperature of liquid rises, the coefficient of viscosity increases. C. As the temperature of gas increases, the coefficient of viscosity increases. D. The onset of turbulence is determined by Reynold's number. E. In a steady flow two stream lines never intersect. Choose the correct answer from the options given below : (1) A, D, E only (2) C, D, E only (3) B, C, D only (4) A, B, C only Ans. (2) 29. Three infinitely long wires with linear charge density λ are placed along the x-axis, y-axis and z-axis respectively. Which of the following denotes an equipotential surface ? (1) $xy + yz + zx = \text{constant}$ (2) $(x + y)(y + z)(z + x) = \text{constant}$ (3) $(x^2 + y^2)(y^2 + z^2)(z^2 + x^2) = \text{constant}$ (4) $xyz = \text{constant}$ Ans. (3)

Sol. $\oint \vec{E} \cdot d\vec{r} = -\int \frac{2k\lambda}{r} dr = 2k\lambda \ln r + c$

Net potential due to all wire

$$v = 2k\lambda \ln \sqrt{x^2 + y^2} + 2k\lambda \ln \sqrt{y^2 + z^2} + 2k\lambda \ln \sqrt{z^2 + x^2} + c$$

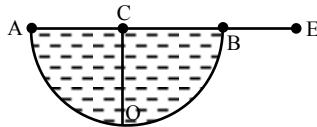
for $v = c$

$$\sqrt{(x^2 + y^2)(y^2 + z^2)(z^2 + x^2)} = c$$

$$\therefore (x^2 + y^2)(y^2 + z^2)(z^2 + x^2) = c$$

where $c = \text{constant}$

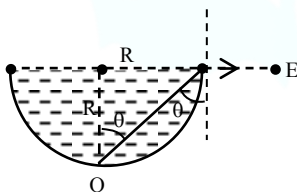
- 30.** A hemispherical vessel is completely filled with a liquid of refractive index μ . A small coin is kept at the lowest point (O) of the vessel as shown in figure. The minimum value of the refractive index of the liquid so that a person can see the coin from point E (at the level of the vessel) is _____.



- (1) $\sqrt{3}$ (2) $\frac{3}{2}$
(3) $\sqrt{2}$ (4) $\frac{\sqrt{3}}{2}$

Ans. (3)

Sol.



$$\sin c = \frac{1}{\mu}$$

for $\mu \rightarrow \text{least}$, $c \rightarrow \text{maximum}$

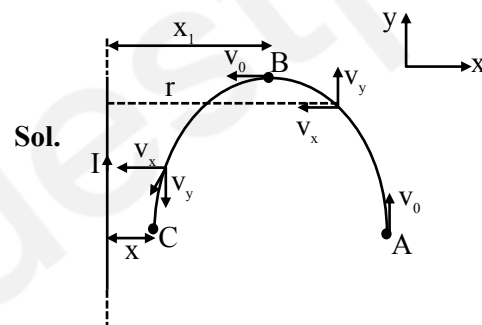
$$\theta = c = 45^\circ$$

$$\mu = \frac{1}{\sin 45^\circ} = \sqrt{2}$$

- 31.** Consider a long thin conducting wire carrying a uniform current I . A particle having mass " M " and charge " q " is released at a distance " a " from the wire with a speed v_0 along the direction of current in the wire. The particle gets attracted to the wire due to magnetic force. The particle turns round when it is at distance x from the wire. The value of x is [μ_0 is vacuum permeability]

- (1) $a \left[1 - \frac{mv_0}{2q\mu_0 I} \right]$ (2) $\frac{a}{2}$
(3) $a \left[1 - \frac{mv_0}{q\mu_0 I} \right]$ (4) $ae^{\frac{-4\pi mv_0}{q\mu_0 I}}$

Ans. (4)



Sol.

$A \rightarrow B$

$$\vec{V} = -v_x \hat{i} + v_y \hat{j}$$

$$\vec{B} = \frac{\mu_0 I}{2\pi r} (-\hat{k})$$

$$\vec{F} = q(\vec{V} \times \vec{B}) = \frac{\mu_0 I q}{2\pi r} [-v_x \hat{j} - v_y \hat{i}]$$

$$a_x = -\frac{\mu_0 I q}{2\pi m} \cdot \frac{v_y}{r}$$

$$a_y = -\frac{\mu_0 I q}{2\pi m} \cdot \frac{v_x}{r}$$

$$\frac{v_x dv_x}{dr} = -\frac{\mu_0 I q}{2\pi m} \frac{v_y}{r}$$

$$\frac{v_x dv_x}{v_y} = -\frac{\mu_0 I q}{2\pi m} \frac{dr}{r}$$

$$\int_0^{v_0} \frac{v_x dv_x}{\sqrt{v_0^2 - v_x^2}} = -\frac{\mu_0 I q}{2\pi m} \int_a^{x_1} \frac{dr}{r}$$

$$\text{Let, } z^2 = v_0^2 - v_x^2$$

$$2zdz = -2v_x dv_x$$

$$zdz = -v_x dv_x$$

$$\frac{v dv_x}{\sqrt{v_0^2 - v_x^2}} = \frac{-zdz}{z} \quad -dz$$

then integral becomes

$$-\int_{v_0}^0 dz = -\frac{\mu_0 I q}{2\pi m} \ln \frac{x}{a}$$

$$v_0 = -\frac{\mu_0 I q}{2\pi m} \ln \frac{x_1}{a}$$

$$x_1 = a e^{\frac{2\pi m v_0}{\mu_0 I q}} \dots\dots(1)$$

For B $\rightarrow$ C

$$\vec{v} = -v_x \hat{i} - v_y \hat{j}$$

$$\vec{B} = \frac{\mu_0 I}{2\pi r} (-\hat{k})$$

$$\vec{F} = q(\vec{v} \times \vec{B}) = \frac{\mu_0 I q}{2\pi r} (-v_x \hat{j} + v_y \hat{i})$$

$$a_x = +\frac{\mu_0 I q}{2\pi m} \frac{v_y}{r} \quad a_y = -\frac{\mu_0 I q}{2\pi m} \frac{v_x}{r}$$

$$\frac{v_x dv_x}{dr} = \frac{\mu_0 I q}{2\pi m} \frac{v_y}{r}$$

$$\int_{v_0}^0 \frac{v_x dv_x}{\sqrt{v_0^2 - v_x^2}} = \frac{\mu_0 I q}{2\pi m} \frac{dr}{r}$$

$$\frac{\mu_0 I q}{2\pi m} \ln \frac{x}{x_1} = -\int_0^{v_0} dz \quad -v_0$$

$$x = x_1 e^{\frac{2\pi m v_0}{\mu_0 I q}} \dots\dots(2)$$

From equation 1 and 2

$$X = a e^{\frac{4\pi m v_0}{\mu_0 I q}}$$

- 32.** A Carnot engine (E) is working between two temperatures 473K and 273K. In a new system two engines – engine E_1 works between 473K to 373K and engine E_2 works between 373K to 273K. If η_{12} , η_1 and η_2 are the efficiencies of the engines E, E_1 and E_2 , respectively, then

(1) $\eta_{12} < \eta_1 + \eta_2$

(2) $\eta_{12} = \eta_1 \eta_2$

(3) $\eta_{12} = \eta_1 + \eta_2$

(4) $\eta_{12} \geq \eta_1 + \eta_2$

Ans. (1)

Sol. $\eta_{12} = 1 - \frac{273}{473} = \frac{200}{473} = 0.423$

$$\eta_1 = 1 - \frac{373}{473} = \frac{100}{473} = 0.211$$

$$\eta_2 = 1 - \frac{273}{373} = \frac{100}{373} = 0.268$$

- 33.** Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R**
Assertion A: A sound wave has higher speed in solids than gases.

Reason R: Gases have higher value of Bulk modulus than solids.

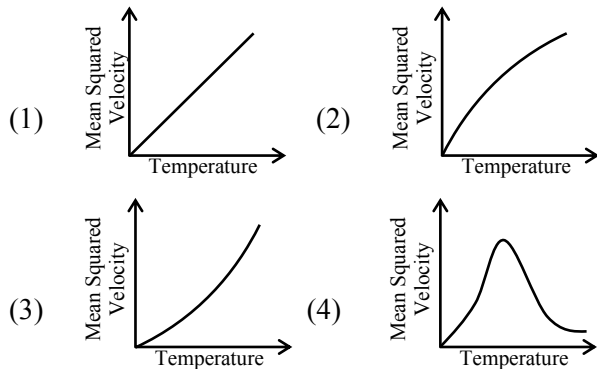
In the light of the above statements, choose the **correct** answer from the options given below

- (1) Both **A** and **R** are true and **R** is the correct explanation of **A**
 (2) **A** is false but **R** is true
 (3) Both **A** and **R** are true but **R** is **NOT** the correct explanation of **A**
 (4) **A** is true but **R** is false.

Ans. (4)

Sol. Solids have higher value of bulk modulus than gases.

34. For a particular ideal gas which of the following graphs represents the variation of mean squared velocity of the gas molecules with temperature ?



Ans. (1)

Sol. $V_{rms} = \sqrt{\frac{3RT}{M}}$

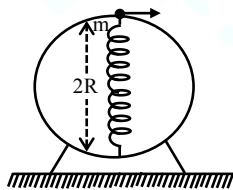
$$V_{rms}^2 = 3RT/M$$

Hence we can conclude that V_{rms}^2 is directly proportional to temperature

$$y = mx$$

⇒ Graph will be straight line

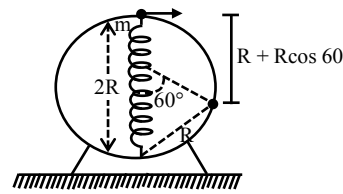
35. A bead of mass 'm' slides without friction on the wall of a vertical circular hoop of radius 'R' as shown in figure. The bead moves under the combined action of gravity and a massless spring (k) attached to the bottom of the hoop. The equilibrium length of the spring is 'R'. If the bead is released from top of the hoop with (negligible) zero initial speed, velocity of bead, when the length of spring becomes 'R', would be (spring constant is 'k', g is acceleration due to gravity)



(1) $2\sqrt{gR + \frac{kR^2}{m}}$ (2) $\sqrt{2Rg + \frac{4kR^2}{m}}$
 (3) $\sqrt{2Rg + \frac{kR^2}{m}}$ (4) $\sqrt{3Rg + \frac{kR^2}{m}}$

Ans. (4)

Sol.



Work energy theorem

$$Mg(R + R\cos 60) + \frac{1}{2}k(R^2 - 0^2) = \frac{1}{2}mv^2$$

$$Mg \frac{3R}{2} + \frac{KR^2}{2} = \frac{1}{2}mv^2$$

$$V = \sqrt{3gR + \frac{KR^2}{m}}$$

36. Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R**

Assertion A: In a central force field, the work done is independent of the path chosen

Reason R: Every force encountered in mechanics does not have an associated potential energy.

In the light of the above statements, choose the **most appropriate** answer from the options given below

- (1) **A** is true but **R** is false
 (2) Both **A** and **R** are true but **R** is **NOT** the correct explanation of **A**
 (3) Both **A** and **R** are true and **R** is the correct explanation of **A**
 (4) **A** is false but **R** is true

Ans. (2)

Sol. Both statement are correct but Reason is not the correct explanation of Assertion.

37. Choose the correct nuclear process from the below options

[p: proton, n: neutron, e^- : electron, e^+ : positron, ν : neutrino, $\bar{\nu}$: antineutrino]

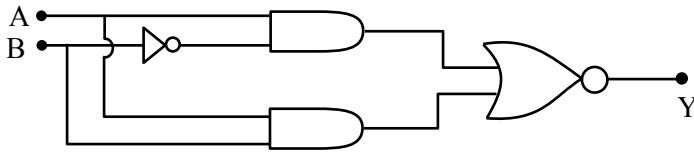
- (1) $n \rightarrow p + e^- + \bar{\nu}$ (2) $n \rightarrow p + e^- + \nu$
 (3) $n \rightarrow p + e^+ + \bar{\nu}$ (4) $n \rightarrow p + e^+ + \nu$

Ans. (1)

Sol. Theoretical equation for β^- decay

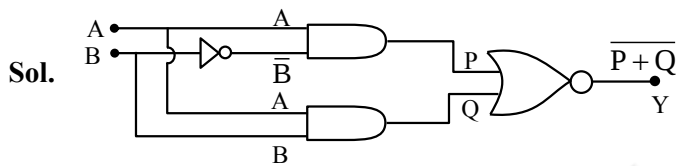
$$n_0^1 \rightarrow p_1^1 + e_{-1}^0 + \bar{\nu}$$

38. Which of the following circuits has the same output as that of the given circuit?



- (1)
- (2)
- (3)
- (4)

Ans. (1)



$$P = A \cdot \bar{B}$$

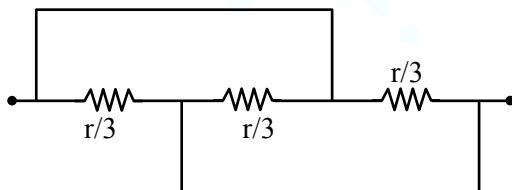
$$Q = A \cdot B$$

$$Y = \overline{P + Q} = \overline{A \cdot \bar{B} + A \cdot B}$$

$$= \overline{A \cdot (B + \bar{B})} = \overline{A \cdot 1}$$

$$Y = \bar{A}$$

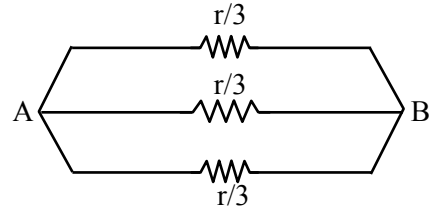
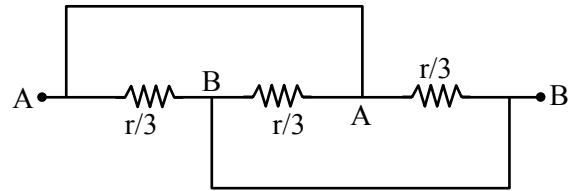
39. Find the equivalent resistance between two ends of the following circuit.



- (1) r (2) $\frac{r}{6}$
- (3) $\frac{r}{9}$ (4) $\frac{r}{3}$

Ans. (3)

Sol.



All are in parallel

$$R_{eq} = \frac{r/3}{3} = r/9$$

40. A wire of resistance R is bent into an equilateral triangle and an identical wire is bent into a square. The ratio of resistance between the two end points of an edge of the triangle to that of the square is

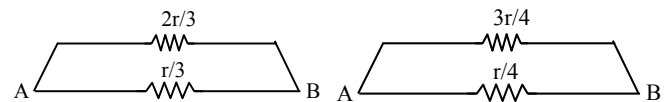
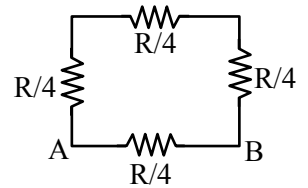
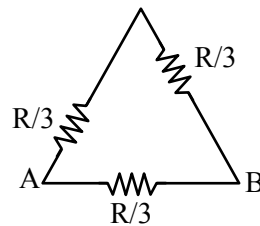
- (1) $9/8$ (2) $8/9$
- (3) $27/32$ (4) $32/27$

Ans. (4)

Sol. $R = \frac{\rho \ell}{A}$

So, $R \propto \ell$

Side length of triangle is $1/3$ of total length.



$$(R_{eq})_1 = \frac{2r/3 \times r/3}{2r/3 + r/3}$$

$$(R_{eq})_2 = \frac{3r/4 \times r/4}{3r/4 + r/4}$$

$$(R_{eq})_1 = 2r/9$$

$$(R_{eq})_2 = 3r/16$$

$$\frac{(R_{eq})_1}{(R_{eq})_2} = \frac{2r/9}{3r/16} = \frac{32}{27}$$

41. Due to presence of an em-wave whose electric component is given by $E = 100 \sin(\omega t - kx) \text{ NC}^{-1}$, a cylinder of length 200 cm holds certain amount of em-energy inside it. If another cylinder of same length but half diameter than previous one holds same amount of em-energy, the magnitude of the electric field of the corresponding em-wave should be modified as

- (1) $25 \sin(\omega t - kx) \text{ NC}^{-1}$
 (2) $200 \sin(\omega t - kx) \text{ NC}^{-1}$
 (3) $400 \sin(\omega t - kx) \text{ NC}^{-1}$
 (4) $50 \sin(\omega t - kx) \text{ NC}^{-1}$

Ans. (2)

Sol. Energy density = $\frac{1}{2} \epsilon_0 E^2 \times c$

$$\text{Energy} = \frac{1}{2} \epsilon_0 E^2 \times c \times \text{volume}$$

$$(\text{Energy})_1 = (\text{Energy})_2 \quad (\text{Given})$$

$$\frac{1}{2} \epsilon_0 E_1 c \pi R_1^2 \times L_1 = \frac{1}{2} \epsilon_0 E_2 c \pi R_2^2 \times L_2$$

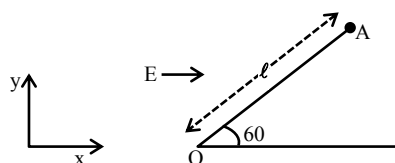
$$E_1^2 R_1^2 = E_2^2 R_2^2$$

$$E_1 R_1 = E_2 R_2$$

$$100 \times R_1 = E_2 \times \frac{R_1}{2}$$

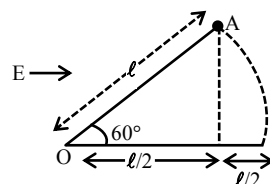
$$E_2 = 200 \text{ N/C}$$

42. A particle of mass 'm' and charge 'q' is fastened to one end 'A' of a massless string having equilibrium length ℓ , whose other end is fixed at point 'O'. The whole system is placed on a frictionless horizontal plane and is initially at rest. If uniform electric field is switched on along the direction as shown in figure, then the speed of the particle when it crosses the x-axis is



- (1) $\sqrt{\frac{2qE\ell}{m}}$ (2) $\sqrt{\frac{qE\ell}{4m}}$
 (3) $\sqrt{\frac{qE\ell}{m}}$ (4) $\sqrt{\frac{qE\ell}{2m}}$

Ans. (3)



Sol.

$$W_{\text{all}} = \Delta k$$

$$W_e = k_f - k_i$$

$$qE \frac{\ell}{2} = \frac{1}{2} mv^2 - 0$$

$$v = \sqrt{\frac{qE\ell}{m}}$$

43. A proton of mass ' m_p ' has same energy as that of a photon of wavelength ' λ '. If the proton is moving at non-relativistic speed, then ratio of its de Broglie wavelength to the wavelength of photon is.

- (1) $\frac{1}{c} \sqrt{\frac{2E}{m_p}}$ (2) $\frac{1}{c} \sqrt{\frac{E}{m_p}}$
 (3) $\frac{1}{c} \sqrt{\frac{E}{2m_p}}$ (4) $\frac{1}{2c} \sqrt{\frac{E}{m_p}}$

Ans. (3)

Sol. E is missing in the question but considering E as energy, the solution will be

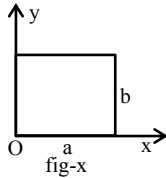
$$E_{\text{photon}} = \frac{hc}{\lambda} = E ; E_{\text{proton}} = \frac{1}{2} m_p v^2 = E$$

$$\frac{\lambda_{\text{proton}}}{\lambda_{\text{photon}}} = \frac{h/p}{hc/E} = \frac{h/\sqrt{2m_p E}}{hc/E}$$

$$= \frac{E}{c\sqrt{2m_p E}}$$

$$\frac{\lambda_{\text{proton}}}{\lambda_{\text{photon}}} = \frac{1}{c} \sqrt{\frac{E}{2m_p}}$$

44. The centre of mass of a thin rectangular plate (fig - x) with sides of length a and b , whose mass per unit area (σ) varies as $\sigma = \frac{\sigma_0 x}{ab}$ (where σ_0 is a constant), would be _____



- (1) $\left(\frac{2}{3}a, \frac{b}{2}\right)$

(3) $\left(\frac{a}{2}, \frac{b}{2}\right)$

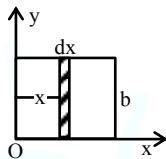
(2) $\left(\frac{2}{3}a, \frac{2}{3}b\right)$

(4) $\left(\frac{1}{3}a, \frac{b}{2}\right)$

Ans. (1)

Sol. σ is constant in y-direction

So, $y_{cm} = b/2$



$$x_{cm} = \frac{\int_0^a x dm}{\int_0^a dm}$$

$$= \frac{\int_0^a x \sigma_x dA}{\int_0^a \sigma_x dA}$$

$$= \frac{\int_0^a x \frac{\sigma_0 x}{ab} b dx}{\int_0^a \frac{\sigma_0 x}{ab} b dx}$$

$$x_{cm} = \frac{\int_0^a x^2 dx}{\int_0^a x dx}$$

$$= \frac{\left(\frac{x^3}{3}\right)_0^a}{\left(\frac{x^2}{2}\right)_0^a} = \frac{a^3/3}{a^2/2}$$

$$= \frac{2a}{3}$$

45. A thin prism P_1 with angle 4° made of glass having refractive index 1.54, is combined with another thin prism P_2 made of glass having refractive index 1.72 to get dispersion without deviation. The angle of the prism P_2 in degrees is

- (1) 4

(2) 3
- (3) 16/3

(4) 1.5

Ans. (2)

Sol. $\delta_{net} = 0$

$$(\mu_1 - 1)A_1 - (\mu_2 - 1)A_2 = 0$$

$$(1.54 - 1)4 - (1.72 - 1)A_2 = 0$$

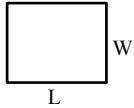
$$A_2 = 3^\circ$$

46. A tiny metallic rectangular sheet has length and breadth of 5 mm and 2.5mm, respectively. Using a specially designed screw gauge which has pitch of 0.75 mm and 15 divisions in the circular scale, you are asked to find the area of the sheet. In this measurement, the maximum fractional error will

be $\frac{x}{100}$ where x is _____

Ans. (3)

Sol.



Since least count of the instrument can be calculated as

$$\text{Least count} = \frac{\text{pitch length}}{\text{No. of division on circular scale}}$$

$$= \frac{0.75}{15} = 0.05 \text{ mm.}$$

Here we are provided $L = 5 \text{ mm}$ & $W = 2.5 \text{ mm}$

$L = 5 \text{ mm}$ & $W = 2.5 \text{ mm}$

$\therefore$ We know that

$$A = L.W$$

For calculating fractional error, we can write

$$\frac{dA}{A} = \frac{dL}{L} + \frac{dW}{W}$$

Here $dL = dW = 0.05 \text{ mm}$

$$\frac{dA}{A} = \frac{0.05}{5} + \frac{0.05}{2.5}$$

$$\Rightarrow \frac{dA}{A} = \frac{1}{100} + \frac{2}{100} = \frac{3}{100},$$

So, $x = 3$

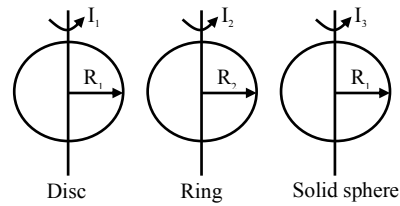
47. The moment of inertia of a solid disc rotating along its diameter is 2.5 times higher than the moment of inertia of a ring rotating in similar way.

The moment of inertia of a solid sphere which has same radius as the disc and rotating in similar way, is n times higher than the moment of inertia of the given ring. Here, $n = \underline{\hspace{2cm}}$.

Consider all the bodies have equal masses.

Ans. (4)

Sol.



$$I_1 = \frac{MR_1^2}{4}, I_2 = \frac{MR_2^2}{2}, I_3 = \frac{2MR_1^2}{5}$$

According to problem

$$\frac{I_1}{I_2} = 2.5 \Rightarrow \frac{\frac{MR_1^2}{4}}{\frac{MR_2^2}{2}} = 2.5 \Rightarrow \frac{R_1^2}{R_2^2} = 5 \dots (1)$$

Now we are provided with information that

$$\frac{I_3}{I_2} = n$$

$$\Rightarrow \frac{\frac{2MR_1^2}{5}}{\frac{MR_2^2}{2}} = n \Rightarrow \frac{4R_1^2}{5R_2^2} = \dots (2)$$

From Eq', (1) and (2)

$$\Rightarrow n = 4$$

48. In a measurement, it is asked to find modulus of elasticity per unit torque applied on the system. The measured quantity has dimension of $[M^a L^b T^c]$. If $b = 3$, the value of c is _____

NTA Ans. (4)

Ans. (0)

Sol. $\frac{\text{modulus of elasticity}}{\text{Torque}} = \frac{\text{Stress}}{\text{strain} \times \text{torque}}$

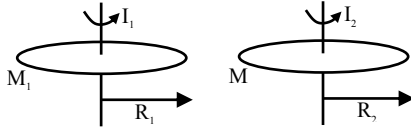
$$= \frac{[\text{Force}]}{[\text{Area}] \times [\text{Force} \times \text{length}]}$$

$$= \frac{1}{[\text{Area} \times \text{length}]} = [L^{-2}]$$

49. Two iron solid discs of negligible thickness have radii R_1 and R_2 and moment of inertia I_1 and I_2 , respectively. For $R_2 = 2R_1$, the ratio of I_1 and I_2 would be $1/x$, where $x =$ _____

Ans. (16)

Sol.



$$\text{Given } R_2 = 2R_1$$

$$M_1 = \sigma \times \pi R_1^2 = M_0$$

$$M_2 = \sigma \times \pi R_2^2 = M_0$$

$$M_2 = \sigma \times \pi R_2^2 = \sigma \times \pi [2R_1]^2 = \sigma \times 4\pi R_1^2 = 4M_0$$

$$\frac{I_1}{I_2} = \frac{\frac{M_1 R_1^2}{2}}{\frac{M_2 R_2^2}{2}} = \frac{M_1 R_1^2}{M_2 R_2^2} = \frac{1}{4} \times \frac{1}{4} = \frac{1}{16}$$

50. A double slit interference experiment performed with a light of wavelength 600 nm forms an interference fringe pattern on a screen with 10th bright fringe having its centre at a distance of 10 mm from the central maximum. Distance of the centre of the same 10th bright fringe from the central maximum when the source of light is replaced by another source of wavelength 660 nm would be _____ mm.

Ans. (11)

- Sol. In case of YDSE the distance of n^{th} maxima from central maxima is given by

$$Y = \frac{n\lambda D}{d}$$

Here n , D & d are same

So, $y \propto \lambda$

$$\Rightarrow \frac{y_2}{y_1} = \frac{\lambda_2}{\lambda_1} \Rightarrow \frac{y_2}{10 \text{ mm}} = \frac{660 \text{ nm}}{600 \text{ nm}}$$

$$\Rightarrow y_2 = 11 \text{ mm}$$

(HELD ON WEDNESDAY 28th JANUARY 2025)
TIME : 9:00 AM TO 12:00 NOON
CHEMISTRY
SECTION-A

51. The incorrect decreasing order of atomic radii is :
 (1) $Mg > Al > C > O$ (2) $Al > B > N > F$
 (3) $Be > Mg > Al > Si$ (4) $Si > P > Cl > F$

Ans. (3)
Sol. Correct order of atomic radii : $Be < Mg > Al > Si$

52. Given below are two statements :

Statement I : In the oxalic acid vs $KMnO_4$ (in the presence of dil H_2SO_4) titration the solution needs to be heated initially to $60^\circ C$, but no heating is required in Ferrous ammonium sulphate (FAS) vs $KMnO_4$ titration (in the presence of dil H_2SO_4)

Statement II : In oxalic acid vs $KMnO_4$ titration, the initial formation of $MnSO_4$ takes place at high temperature, which then acts as catalyst for further reaction. In the case of FAS vs $KMnO_4$, heating oxidizes Fe^{2+} into Fe^{3+} by oxygen of air and error may be introduced in the experiment.

In the light of the above statements, choose the **correct** answer from the options given below :

- (1) Statement I is false but Statement II is true
 (2) Both Statement I and Statement II are true
 (3) Statement I is true but Statement II is false
 (4) Both Statement I and Statement II are false

Ans. (2)
Sol. $2MnO_4^- + 5(COO)^{2-} + 16H^+ \rightarrow$


This reaction is slow at room temperature, but becomes fast at $60^\circ C$. Manganese(II) ions catalyse the reaction; thus, the reaction is autocatalytic; once manganese(II) ions are formed, it becomes faster and faster.

The titration of FAS v/s $KMnO_4$ do not require heating because at higher temperature the oxidation of Fe^{+2} to Fe^{+3} by atmospheric O_2 will be prominent.

TEST PAPER WITH SOLUTIONS

53. Match the List-I with List-II

List-I (Redox Reaction)		List-II (Type of Redox Reaction)	
A	$CH_{4(g)} + 2O_{2(g)} \xrightarrow{\Delta} CO_{2(g)} + 2H_2O_{(l)}$	(I)	Disproportionation reaction
B	$2NaH_{(s)} \xrightarrow{\Delta} 2Na_{(s)} + H_{2(g)}$	(II)	Combination reaction
C	$V_2O_{5(s)} + 5Ca_{(s)} \xrightarrow{\Delta} 2V_{(s)} + 5CaO_{(s)}$	(III)	Decomposition reaction
D	$2H_2O_{2(aq)} \xrightarrow{\Delta} 2H_2O_{(l)} + O_{2(g)}$	(IV)	Displacement reaction

Choose the **correct** answer from the options given below :

- (1) A-II, B-III, C-IV, D-I
 (2) A-II, B-III, C-I, D-IV
 (3) A-III, B-IV, C-I, D-II
 (4) A-IV, B-I, C-II, D-III

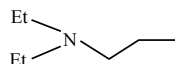
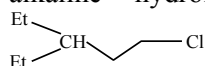
Ans. (1)
Sol. (A) Combustion of hydrocarbon

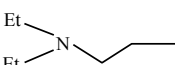
(B) Decomposition into gaseous product.

(C) Displacement of 'V' by 'Ca' atom.

 (D) Disproportionation of $H_2O_2^{-1}$ into O^{-2} and O^0 oxidation states.

54. Given below are two statements :

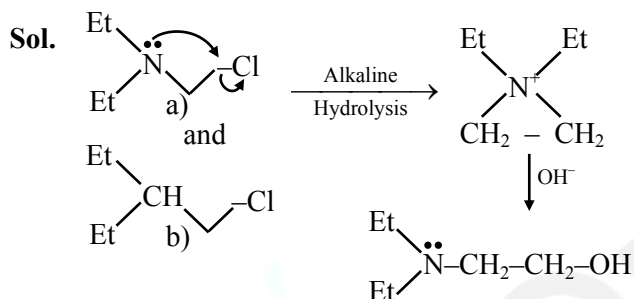
Statement I :  will undergo alkaline hydrolysis at a faster rate than 

Statement II :  intramolecular substitution takes place first by involving lone pair of electrons on nitrogen.

In the light of the above statements, choose the **most appropriate** answer from the options given below :

- (1) Both Statement I and Statement II are incorrect
- (2) Statement I is incorrect but statement II is correct
- (3) Both Statement I and Statement II are correct
- (4) Statement I is correct but Statement II is incorrect

Ans. (3)



Rate of (a) is faster than rate of (b) because it is an intramolecular substitution.

55. A weak acid HA has degree of dissociation x . Which option gives the correct expression of $\text{pH} = \text{pK}_a$?

- (1) $\log(1 + 2x)$
- (2) $\log\left(\frac{1-x}{x}\right)$
- (3) 0
- (4) $\log\left(\frac{x}{1-x}\right)$

Ans. (4)



$t=0$ a

$t=t$ $a(1-x)$ ax ax

$K_a = (ax) \frac{(x)}{1-x}$; $[\text{H}^+] = ax$

$-\log(K_a) = -\log(ax) - \log\left(\frac{x}{1-x}\right)$

$\text{pK}_a = \text{pH} - \log\left(\frac{x}{1-x}\right)$

$\text{pH} - \text{pK}_a = \log\left(\frac{x}{1-x}\right)$

56. Consider 'n' is the number of lone pair of electrons present in the equatorial position of the most stable structure of ClF_3 . The ions from the following with 'n' number of unpaired electrons are :

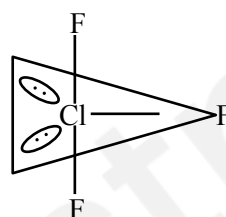
- A. V^{3+}
- B. Ti^{3+}
- C. Cu^{2+}
- D. Ni^{2+}
- E. Ti^{2+}

Choose the **correct** answer from the options given below :

- (1) A and C only
- (2) A, D and E only
- (3) B and C only
- (4) B and D only

Ans. (2)

Sol. ClF_3



$n = 2$ (No of lone pair present in equatorial plane)
(Unpaired e^-)

- (A) $\text{V}^{+3} : [\text{Ar}]3d^2$ 2
- (B) $\text{Ti}^{+3} : [\text{Ar}]3d^1$ 1
- (C) $\text{Cu}^{+2} : [\text{Ar}]3d^9$ 1
- (D) $\text{Ni}^{+2} : [\text{Ar}]3d^8$ 2
- (E) $\text{Ti}^{+2} : [\text{Ar}]3d^2$ 2

57.

$[\text{A}]_0 / \text{molL}^{-1}$	$t_{1/2} / \text{min}$
0.100	200
0.025	100

For a given reaction $\text{R} \rightarrow \text{P}$, $t_{1/2}$ is related to $[\text{A}]_0$ as given in table :

Given : $\log 2 = 0.30$

Which of the following is **true**?

A. The order of the reaction is $\frac{1}{2}$.

B. If $[\text{A}]_0$ is 1M, then $t_{1/2}$ is $200\sqrt{10}$ min

C. The order of the reaction changes to 1 if the concentration of reactant changes from 0.100 M to 0.500 M.

D. $t_{1/2}$ is 800 min for $[\text{A}]_0 = 1.6$ M

Choose the **correct** answer from the options given below :

- (1) A and C only
- (2) A and B only
- (3) A, B and D only
- (4) C and D only

Ans. (3)

Sol. $t_{1/2} \propto \frac{1}{A_0^{n-1}}$

$$\frac{(t_{1/2})_1}{(t_{1/2})_2} = \frac{(A_0)_2^{n-1}}{(A_0)_1^{n-1}}$$

$$\frac{200}{100} = \left(\frac{0.025}{0.100} \right)^{n-1}$$

$$2 = \left(\frac{1}{4} \right)^{n-1}$$

$$n - 1 = -\frac{1}{2}$$

$$n = \frac{1}{2} \text{ (order)}$$

$$\Rightarrow t_{1/2} \propto \sqrt{A_0}$$

$$\frac{200}{t_{1/2}} = \frac{(0.1)^{1/2}}{(1)^{1/2}}$$

$$\text{when } A_0 = 1 \text{ M}$$

$$t_{1/2} = 200\sqrt{10} \text{ min}$$

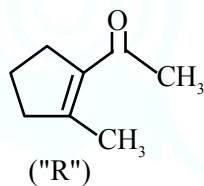
* 1st order kinetics have $t_{1/2}$ independent of their concentration. So upon changing the concentration $t_{1/2}$ should not change for first order reaction.

$$\frac{200}{t_{1/2}} = \frac{(0.1)^{1/2}}{(1.6)^{1/2}}$$

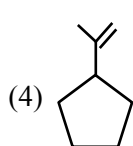
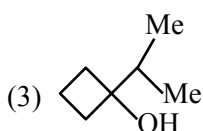
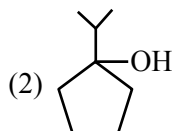
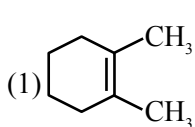
$$\text{when } A_0 = 1.6 \text{ M}$$

$$t_{1/2} = 800 \text{ min}$$

58. A molecule ("P") on treatment with acid undergoes rearrangement and gives ("Q") ("Q") on ozonolysis followed by reflux under alkaline condition gives ("R"). The structure of ("R") is given below :

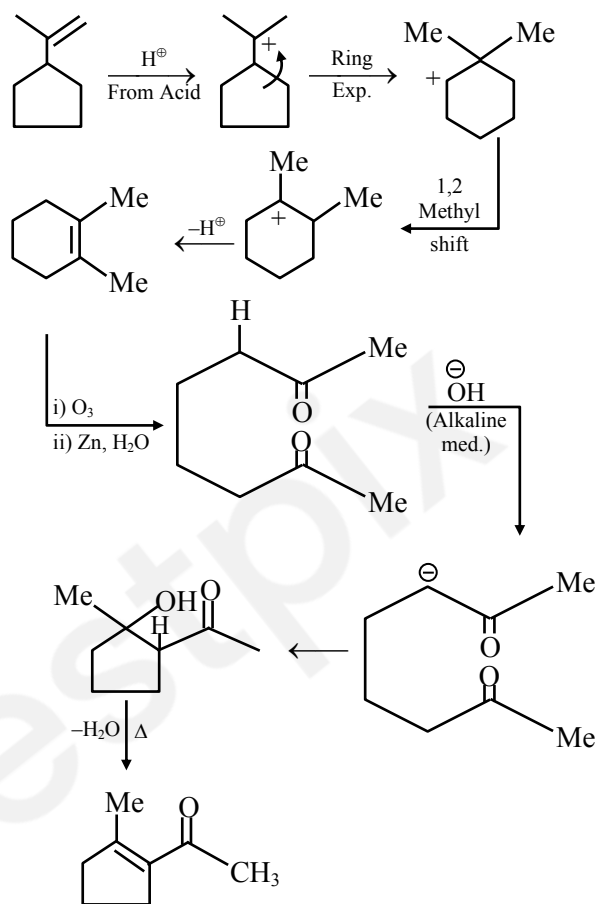


The structure of ("P") is

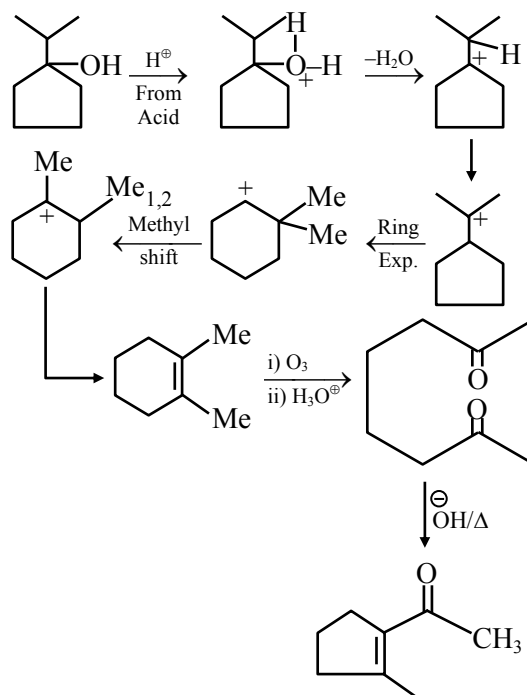


Ans. (2)

Sol.



OR



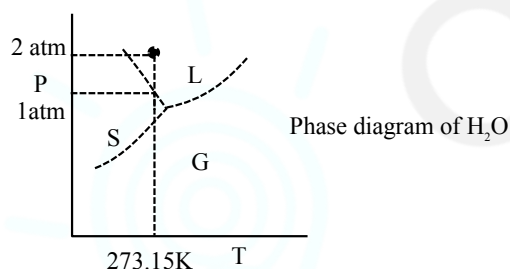
Note : In question about molecule "P" is not clarified, whether it is alcohol or alkene and as in question language rearrangement product is asking hence according to question language ans. is either (2) or (4). As alkene also undergoes rearrangement in presence of acid but option (2) also fulfil all conditions.

59. Ice and water are placed in a closed container at a pressure of 1 atm and temperature 273.15 K. If pressure of the system is increased 2 times, keeping temperature constant, then identify correct observation from following :

- (1) Volume of system increases.
- (2) Liquid phase disappears completely.
- (3) The amount of ice decreases.
- (4) The solid phase (ice) disappears completely.

Ans. (4)

Sol.

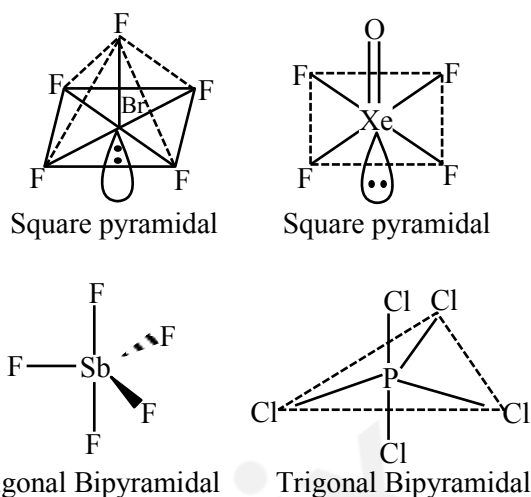


If pressure is made two time then mixture of ice and water will completely convert into water (liquid) form.

60. The molecules having square pyramidal geometry are
- (1) BrF_5 & XeOF_4
 - (2) SbF_5 & XeOF_4
 - (3) SbF_5 & PCl_5
 - (4) BrF_5 & PCl_5

Ans. (1)

Sol.



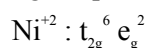
BrF_5 : Square pyramidal
 XeOF_4 : Square pyramidal
 SbF_5 : Trigonal bipyramidal
 PCl_5 : Trigonal bipyramidal

61. The metal ion whose electronic configuration is not affected by the nature of the ligand and which gives a violet colour in non-luminous flame under hot condition in borax bead test is

- (1) Ti^{3+}
- (2) Ni^{2+}
- (3) Mn^{2+}
- (4) Cr^{3+}

Ans. (2)

Sol. Ni^{+2} gives violet colored bead in non-luminous flame under hot conditions. Ni^{+2} has d^8 configuration which does not depend on nature of ligand present in octahedral complex.



62. Both acetaldehyde and acetone (individually) undergo which of the following reactions?

- A. Iodoform Reaction
- B. Cannizzaro Reaction
- C. Aldol condensation
- D. Tollen's Test
- E. Clemmensen Reduction

Choose the **correct** answer from the options given below :

- (1) A, B and D only
- (2) A, C and E only
- (3) C and E only
- (4) B, C and D only

Ans. (2)

Sol.

S. No.	Name of Reaction	Acetaldehyde $\text{CH}_3\text{--}\overset{\text{O}}{\underset{\text{O}}{\parallel}}\text{C--H}$	Acetone $\text{CH}_3\text{--}\overset{\text{O}}{\underset{\text{O}}{\parallel}}\text{C--CH}_3$
1	Iodoform reaction	⊕ve	⊕ve
2	Cannizzaro	⊖ve	⊖ve
3	Aldol reaction	⊕ve	⊕ve
4	Tollen's test	⊕ve	⊖ve
5	Clemmensen reduction	⊕ve	⊕ve

Ans. (2) A, C and E only

63. In a multielectron atom, which of the following orbitals described by three quantum numbers with have same energy in absence of electric and magnetic fields?

A. $n = 1, l = 0, m_l = 0$

B. $n = 2, l = 0, m_l = 0$

C. $n = 2, l = 1, m_l = 1$

D. $n = 3, l = 2, m_l = 1$

E. $n = 3, l = 2, m_l = 0$

Choose the **correct** answer from the options given below :

(1) A and B only

(2) B and C only

(3) C and D only

(4) D and E only

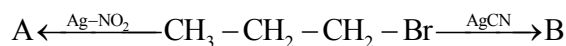
Ans. (4)

Sol.

	orbital
A : $n = 1, l = 0, m_l = 0$	1s
B : $n = 2, l = 0, m_l = 0$	2s
C : $n = 3, l = 1, m_l = 1$	3p
D : $n = 3, l = 2, m_l = 1$	3d
E : $n = 3, l = 2, m_l = 0$	3d

In absence of electric and magnetic fields, all orbitals of 3d are degenerate

64. The products A and B in the following reactions, respectively are



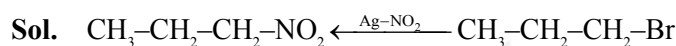
(1) $\text{CH}_3\text{--CH}_2\text{--CH}_2\text{--ONO}$, $\text{CH}_3\text{--CH}_2\text{--CH}_2\text{--NC}$

(2) $\text{CH}_3\text{--CH}_2\text{--CH}_2\text{--ONO}$, $\text{CH}_3\text{--CH}_2\text{--CH}_2\text{--CN}$

(3) $\text{CH}_3\text{--CH}_2\text{--CH}_2\text{--NO}_2$, $\text{CH}_3\text{--CH}_2\text{--CH}_2\text{--CN}$

(4) $\text{CH}_3\text{--CH}_2\text{--CH}_2\text{--NO}_2$, $\text{CH}_3\text{--CH}_2\text{--CH}_2\text{--NC}$

Ans. (4)



(A)



(B)

65. What is the freezing point depression constant of a solvent, 50 g of which contain 1 g non volatile solute (molar mass 256 g mol^{-1}) and the decrease in freezing point is 0.40 K ?

(1) $5.12 \text{ K kg mol}^{-1}$ (2) $4.43 \text{ K kg mol}^{-1}$

(3) $1.86 \text{ K kg mol}^{-1}$ (4) $3.72 \text{ K kg mol}^{-1}$

Ans. (1)



$$0.4 = K_b \cdot \frac{1}{50 \times 10^{-3}}$$

$$K_b = 5.12 \text{ K kg / mol}$$

66. Consider the following elements In, Tl, Al, Pb, Sn and Ge.

The most stable oxidation states of elements with highest and lowest first ionisation enthalpies, respectively, are

(1) +2 and +3 (2) +4 and +3

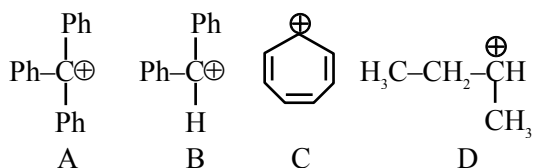
(3) +4 and +1 (4) +1 and +4

Ans. (3)

Sol. Among Al, In, Tl, Ge, Sn, Pb, the metal having highest IE_1 is Ge and lowest IE_1 is In.

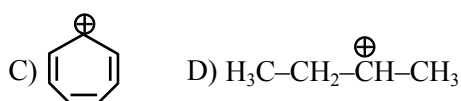
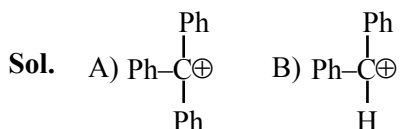
Most stable oxidation state of Ge is +4 and In is +3.

67. The correct order of stability of following carbocations is :



- (1) A > B > C > D (2) B > C > A > D
(3) C > B > A > D (4) C > A > B > D

Ans. (4)



Solution :-

C is aromatic due to $\oplus$ ve charge hence it is most stable

A have more resonance structure

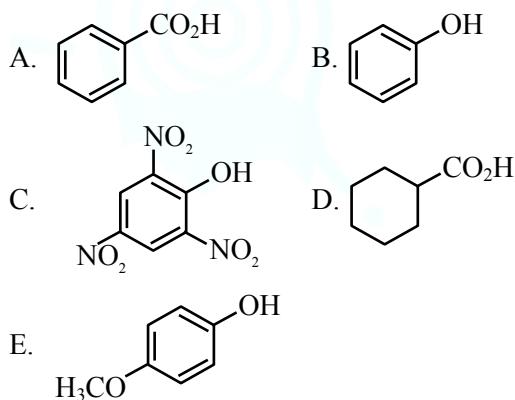
B have less resonance structure

D have only hyper conjugation

Consider First Aromaticity > Resonance > Hyper conjugation

Ans. D < B < A < C

68. The compounds that produce CO_2 with aqueous NaHCO_3 solution are :



Choose the **correct** answer from the options given below :

- (1) A and C only (2) A, B and E only
(3) A, C and D only (4) A and B only

Ans. (3)

Sol. A, C, D produce CO_2 with aqueous NaHCO_3 solution.
A, C, D acids are stronger acid than H_2CO_3 (Carbonic acid)

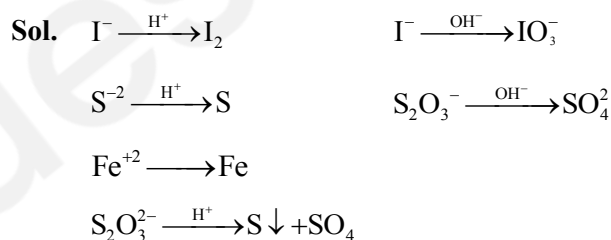
69. Which of the following oxidation reactions are carried out by both $\text{K}_2\text{Cr}_2\text{O}_7$ and KMnO_4 in acidic medium ?

- A. $\text{I}^- \rightarrow \text{I}_2$
B. $\text{S}^{2-} \rightarrow \text{S}$
C. $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+}$
D. $\text{I}^- \rightarrow \text{IO}_3^-$
E. $\text{S}_2\text{O}_3^{2-} \rightarrow \text{SO}_4^{2-}$

Choose the **correct** answer from the options given below :

- (1) B, C and D only (2) A, D and E only
(3) A, B and C only (4) C, D and E only

Ans. (3)



70. Given below are two statements :

Statement I : D-glucose pentaacetate reacts with 2, 4-dinitrophenylhydrazine.

Statement II : Starch, on heating with concentrated sulfuric acid at 100°C and 2-3 atmosphere pressure produces glucose.

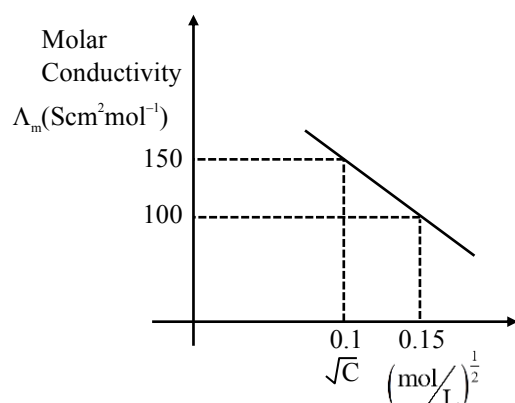
In the light of the above statements, choose the **correct** answer from the options given below

- (1) Both Statement I and Statement II are false
(2) Statement I is false but Statement II is true
(3) Statement I is true but Statement II is false
(4) Both Statement I and Statement II are true

Ans. (2)

SECTION-B

71. Given below is the plot of the molar conductivity vs $\sqrt{\text{concentration}}$ for KCl in aqueous solution.



If, for the higher concentration of KCl solution, the resistance of the conductivity cell is 100Ω , then the resistance of the same cell with the dilute solution is ' x ' Ω .

The value of x is _____ (Nearest integer)

Ans. 150

Sol. $R = \rho \frac{\ell}{A}$

$\kappa = G \cdot G^*$ $G = \frac{1}{R}$; $\kappa = \frac{1}{\rho}$

$G^* = \frac{\ell}{A}$

R = Resistance

ρ = Resistivity

$\frac{\ell}{A}$ = cell constant (G^*)

$\frac{\kappa_c}{\kappa_d} = \frac{R}{R} ; \lambda_m = \frac{\kappa \times 1000}{R}$

$\frac{\kappa_c}{\kappa_d} = \frac{(\lambda_m \cdot C)}{(\lambda_m \cdot C)_d} \cdot \frac{R}{R_c}$ c = concentrated sol.
 d = diluted solution

$\frac{100 \cdot (0.15)^2}{150 \cdot (0.1)^2} = \frac{R_d}{100}$

$R_d = 150\Omega$

72. Quantitative analysis of an organic compound (X) shows following % composition.

C : 14.5% Cl : 64.46%

H : 1.8%

(Empirical formula mass of the compound (X) is _____ $\times 10^{-1}$)

(Given molar mass in g mol^{-1} of C : 12, H : 1, O : 16, Cl : 35.5)

Ans. 1655

Sol.	C	Cl	H	O
%mass	14.5	64.46	1.8	19.24
Molar ratio	$\frac{14.5}{12}$	$\frac{64.46}{35.5}$	$\frac{1.8}{1}$	$\frac{19.24}{16}$
	1.2	1.8	1.8	1.2
Minimum	2	3	3	2
integral ratio				

Empirical formula = $\text{C}_2\text{H}_3\text{Cl}_3\text{O}_2$

Mass = 165.5

Mass = 1655×10^{-1}

73. The molarity of a 70% (mass/mass) aqueous solution of a monobasic acid (X) is _____ M (Nearest integer)

[Given : Density of aqueous solution of (X) is 1.25 g mL^{-1}]

Molar mass of the acid is 70 g mol^{-1}]

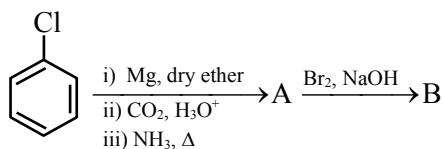
Ans. 125

Sol. Assuming 100 gm solution contain 70 gm solute.

Volume of 100 gm solution will be $\frac{100}{1.25} \text{ ml}$.

Molarity = $\frac{70 / 70}{100 / 1.25} \times 1000 = 12.5$ or 125×10^{-1}

74. Consider the following sequence of reactions :



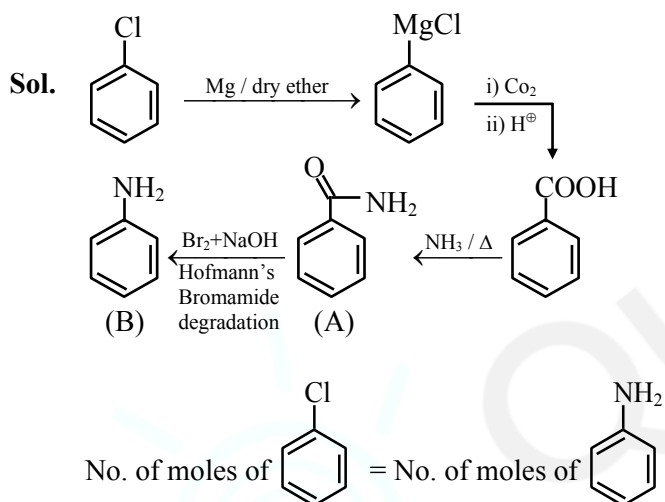
Chlorobenzene

11.25 mg of chlorobenzene will produce $\times 10^{-1}$ mg of product B.

(Consider the reactions result in complete conversion.)

[Given molar mass of C, H, O, N and Cl as 12, 1, 16, 14 and 35.5 g mol⁻¹ respectively]

Ans. 93



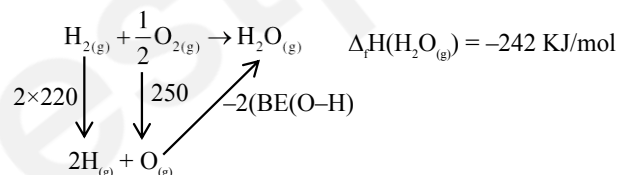
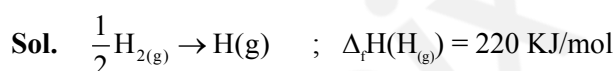
$$\frac{11.25 \times 10^{-3}}{112.5} = \frac{x \times 10^{-1} \times 10^{-3}}{93}$$

$$x \times 10^{-1} = 93 \times 0.1$$

$$x = 93 \text{ mg}$$

75. The formation enthalpies, $\Delta H_f^\ominus$ for $\text{H}_{(\text{g})}$ and $\text{O}_{(\text{g})}$ are 220.0 and 250.0 kJ mol⁻¹, respectively, at 298.15 K, and $\Delta H_f^\ominus$ for $\text{H}_2\text{O}_{(\text{g})}$ is -242.0 kJ mol⁻¹ at the same temperature. The average bond enthalpy of the O-H bond in water at 298.15 K is _____ kJ mol⁻¹ (nearest integer).

Ans. 466



$$\Delta_f H(\text{H}_2\text{O}_{(\text{l})}) = -242 = 440 + 250 - 2(\text{B.E.}(\text{O}-\text{H}))$$

$$\boxed{\text{BE}(\text{O}-\text{H}) = 466 \text{ KJ/mol}}$$