

(Held On Tuesday 09th April, 2024)

TIME : 3 : 00 PM to 6 : 00 PM



$$\frac{dy}{\sqrt{1-y^2}} = dx \text{ OR } \frac{dy}{\sqrt{1-y^2}} = -dx$$

$$\Rightarrow \sin^{-1}y = x + c, \sin^{-1}y = -x + c$$

$$x = 0, y = 0 \Rightarrow c = 0$$

$$\sin^{-1}y = x, \text{ as } y \geq 0$$

$$\sin x = y$$

$$\Rightarrow \frac{dy}{dx} = \cos x$$

$$\frac{d^2y}{dx^2} = -\sin x$$

$$\Rightarrow -\sin x + \sin x + 1 = 1$$

4. Let z be a complex number such that the real part of $\frac{z-2i}{z+2i}$ is zero. Then, the maximum value of $|z-(6+8i)|$ is equal to :

$$(1) 12 \quad (2) \infty$$

$$(3) 10 \quad (4) 8$$

Ans. (1)

$$\text{Sol. } \frac{z-2i}{z+2i} \cdot \frac{\bar{z}-2i}{\bar{z}-2i} = 0$$

$$z\bar{z} - 2i\bar{z} - 2iz + 4(-1) = 0$$

$$+z\bar{z} + 2zi + 2\bar{z}i + 4(-1) = 0$$

$$\Rightarrow 2|z|^2 = 8 \Rightarrow |z| = 2$$

$$|z-(6+8i)|_{\text{maximum}} = 10 + 2 = 12$$

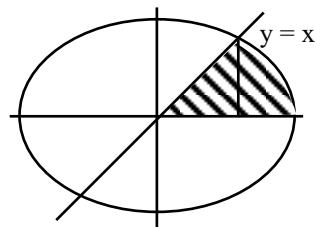
5. The area (in square units) of the region enclosed by the ellipse $x^2 + 3y^2 = 18$ in the first quadrant below the line $y = x$ is :

$$(1) \sqrt{3}\pi \quad (2) \frac{3}{4}\sqrt{3}\pi$$

$$(3) \sqrt{3}\pi \quad (4) \sqrt{3}\pi + 1$$

Ans. (2)

$$\text{Sol. } \frac{x^2}{18} + \frac{y^2}{6} = 1$$



$$\frac{x^2}{18} + \frac{3x^2}{18} = 1 \Rightarrow 4x^2 = 18 \Rightarrow x^2 = \frac{9}{2}$$

$$\int_{\frac{3}{\sqrt{2}}}^{3\sqrt{2}} \frac{\sqrt{18-x^2}}{\sqrt{3}} dx$$

$$= \frac{1}{\sqrt{3}} \left[\frac{x\sqrt{18-x^2}}{2} + \frac{18}{2} \sin^{-1} \frac{x}{3\sqrt{2}} \right]_{\frac{3}{\sqrt{2}}}^{3\sqrt{2}}$$

$$= \frac{1}{\sqrt{3}} \left(9 \times \frac{\pi}{6} - \frac{3}{2\sqrt{2}} \times \frac{3\sqrt{3}}{\sqrt{2}} - 9 \times \frac{\pi}{6} \right)$$

Required Area

$$= \frac{1}{2} \times \frac{9}{2} + \left(\frac{18\pi}{6} - \frac{9\sqrt{3}}{2} \right) \frac{1}{\sqrt{3}}$$

$$= \sqrt{3}\pi$$

6. Let the foci of a hyperbola H coincide with the foci

$$\text{of the ellipse E : } \frac{(x-1)^2}{100} - \frac{(y-1)^2}{75} = 1 \text{ and the}$$

eccentricity of the hyperbola H be the reciprocal of the eccentricity of the ellipse E. If the length of the transverse axis of H is α and the length of its conjugate axis is β , then $3\alpha^2 + 2\beta^2$ is equal to :

$$(1) 242$$

$$(2) 225$$

$$(3) 237$$

$$(4) 205$$

Ans. (2)



11. $\lim_{x \rightarrow \frac{\pi}{2}} \left| \frac{\int_{x^3}^{(\pi/2)^3} (\sin(2t^{1/3}) + \cos(t^{1/3})) dt}{\left(x - \frac{\pi}{2}\right)^2} \right|$ is equal

to :

- (1) $\frac{9\pi^2}{8}$ (2) $\frac{11\pi^2}{10}$
 (3) $\frac{3\pi^2}{2}$ (4) $\frac{5\pi^2}{9}$

Ans. (1)

Sol. $\lim_{x \rightarrow \frac{\pi}{2}} \frac{0 - \{\sin(2x) + \cos(x)\} \cdot 3x^2}{2\left(x - \frac{\pi}{2}\right)}$

$$\begin{aligned} &= \lim_{x \rightarrow \frac{\pi}{2}} \frac{-\{2\sin x \cos x + \cos x\} 3x^2}{2\left(x - \frac{\pi}{2}\right)} \\ &= \lim_{x \rightarrow \frac{\pi}{2}} \left[\frac{2\sin x \sin\left(\frac{\pi}{2} - x\right)}{\left(x - \frac{\pi}{2}\right)} + \frac{\sin\left(\frac{\pi}{2} - x\right)}{2\left(x - \frac{\pi}{2}\right)} \right] 3x^2 \\ &= \left(1(1) + \frac{1}{2}\right) 3 \left(\frac{\pi}{2}\right)^2 \\ &= \frac{9\pi^2}{8} \end{aligned}$$

12. The sum of the coefficient of $x^{2/3}$ and $x^{-2/5}$ in the binomial expansion of $\left(x^{2/3} + \frac{1}{2}\right)^9$ is :
 (1) 21/4 (2) 69/16
 (3) 63/16 (4) 19/4

Ans. (1)

Sol. $T_{r+1} = {}^9C_r (x^{2/3})^{9-r} \left(\frac{x^{-2/5}}{2}\right)^r$
 $= {}^9C_r \left(\frac{1}{2}\right)^r (r)^{\binom{6-2r-2r}{3}}$

for coefficient of $x^{2/3}$, put $6 - \frac{2r}{3} - \frac{2r}{5} = \frac{2}{3}$

$$\Rightarrow r = 5$$

$$\therefore \text{Coefficient of } x^{2/3} \text{ is } {}^9C_5 \left(\frac{1}{5}\right)^5$$

For coefficient of $x^{-2/5}$, put $6 - \frac{2r}{3} - \frac{2r}{5} = -\frac{2}{5}$

$$\Rightarrow r = 6$$

$$\text{Coefficient of } x^{-2/5} \text{ is } {}^9C_6 \left(\frac{1}{2}\right)^6$$

$$\text{Sum} = {}^9C_5 \left(\frac{1}{2}\right)^5 + {}^9C_6 \left(\frac{1}{2}\right)^6 = \frac{21}{4}$$

13. Let $B = \begin{vmatrix} 1 & 3 \\ 1 & 5 \end{vmatrix}$ and A be a 2×2 matrix such that

$AB^{-1} = A^{-1}$. If $BCB^{-1} = A$ and $C^4 + \alpha C^2 + \beta I = O$, then $2\beta - \alpha$ is equal to :

- (1) 16 (2) 2
 (3) 8 (4) 10

Ans. (4)

Sol. $BCB^{-1} = A$

$$\Rightarrow (BCB^{-1})(BCB^{-1}) = A \cdot A$$

$$\Rightarrow BCI CB^{-1} = A^2$$

$$\Rightarrow BC^2B^{-1} = A^2$$

$$\Rightarrow B^{-1}(BC^2B^{-1})B = B^{-1}(A \cdot A)B$$

From equation (1)

$$C^2 = A^{-1} \cdot A \cdot B$$

$$C^2 = B$$

$$\text{Also } AB^{-1} = A^{-1}$$

$$\Rightarrow AB^{-1} \cdot A = A^{-1} \cdot A = I$$

$$\Rightarrow A^{-1}(AB^{-1}A) = A^{-1}I$$

$$B^{-1}A = A^{-1}$$

Now characteristics equation of C^2 is

$$|C^2 - \lambda I| = 0$$

$$|B - \lambda I| = 0$$



$$\Rightarrow \begin{vmatrix} 1-\lambda & 3 \\ 1 & 5-\lambda \end{vmatrix} = 0$$

$$\Rightarrow (1-\lambda)(5-\lambda) - 3 = 0 \Rightarrow (\lambda^2 - 6\lambda + 5) - 3 = 0$$

$$\Rightarrow \lambda^2 - 6\lambda + 2 = 0$$

$$\Rightarrow \beta^2 - 6B + 2I = 0$$

$$\Rightarrow C^4 - 6C^2 + 2I = 0$$

$$\alpha = -6$$

$$\beta = 2$$

$$\therefore 2\beta - \alpha = 4 + 6 = 10$$

- 14.** If $\log_e y = 3 \sin^{-1} x$, then $(1-x^2)y'' - xy'$ at $x = \frac{1}{2}$

is equal to :

$$(1) 9e^{\pi/6} \quad (2) 3e^{\pi/6}$$

$$(3) 3e^{\pi/2} \quad (4) 9e^{\pi/2}$$

Ans. (4)

Sol. $\ln(y) = 3 \sin^{-1} x$

$$\frac{1}{y} \cdot y' = 3 \frac{1}{\sqrt{1-x^2}}$$

$$\Rightarrow y' = \frac{3y}{\sqrt{1-x^2}} \text{ at } x = \frac{1}{2}$$

$$\Rightarrow y' = \frac{3e^{3(\frac{\pi}{6})}}{\sqrt{3}} - 2\sqrt{3}e^{\frac{\pi}{2}}$$

$$\Rightarrow y'' = 3 \left| \frac{\sqrt{1-x^2}y' - y \frac{1}{2\sqrt{1-x^2}}(-2x)}{(1-x^2)^2} \right|$$

$$\Rightarrow (1-x^2)y'' = 3 \left| 3y + \frac{xy}{\sqrt{1-x^2}} \right|$$

$$\downarrow \text{at } x = \frac{1}{2}, y = e^{3\sin^{-1}(\frac{1}{2})} = e^{3(\frac{\pi}{6})} = e^{\frac{\pi}{2}}$$

$$(1-x^2)y'' \Big|_{x=\frac{1}{2}} = 3 \left(3e^{\frac{\pi}{2}} + \frac{\frac{1}{2}e^{\frac{\pi}{2}}}{\sqrt{3}} \right)$$

$$= 3e^{\frac{\pi}{2}} \left(3 + \frac{1}{\sqrt{3}} \right)$$

$$(1-x^2)y'' - xy' \Big|_{x=\frac{1}{2}}$$

$$= 3e^{\frac{\pi}{2}} \left(3 + \frac{1}{\sqrt{3}} \right) - \frac{1}{2} \left(2\sqrt{3}e^{\frac{\pi}{2}} \right) = 9e^{\frac{\pi}{2}}$$

- 15.** The integral $\int_{1/4}^{3/4} \cos \left(2 \cot^{-1} \sqrt{\frac{1-x}{1+x}} \right) dx$ is equal

to:

$$(1) -1/2 \quad (2) 1/4$$

$$(3) 1/2 \quad (4) -1/4$$

Ans. (4)

Sol. $I = \int_{1/4}^{3/4} \cos \left(2 \cot^{-1} \sqrt{\frac{-x}{1+x}} \right) dx$

$$\int_{1/4}^{3/4} \cos \left(2 \left| \tan^{-1} \sqrt{\frac{1+x}{1-x}} \right| \right) dx$$

$$\int_{1/4}^{3/4} \frac{1 - \tan^2 \left| \tan^{-1} \sqrt{\frac{1+x}{1-x}} \right|}{1 + \tan^2 \left| \tan^{-1} \sqrt{\frac{1+x}{1-x}} \right|} dx$$

$$= \int_{1/4}^{3/4} \frac{1 - \left(\frac{1+x}{1-x} \right)}{1 + \left(\frac{1+x}{1-x} \right)} dx = \int_{1/4}^{3/4} \frac{-2x}{2} dx$$

$$= \int_{1/4}^{3/4} (-x) dx = - \left(\frac{x^2}{2} \right) \Big|_{1/4}^{3/4}$$

$$= -\frac{1}{2} \left| \frac{9}{16} - \frac{1}{16} \right|$$

$$= -\frac{1}{4}$$



$$\text{Sol. } \sum_{n=0}^{\infty} ar^n = 57$$

$$a + ar + ar^2 + \dots = 57$$

$$\frac{a}{1-r} = 57 \dots\dots\dots (I)$$

$$\sum_{n=0}^{\infty} a^3 r^{3n} = 9747$$

$$a^3 + a^3 \cdot r^3 + a^3 \cdot r^6 + \dots = 9746$$

$$\frac{a^3}{1-r^3} = 9746 \dots \dots \dots \text{(II)}$$

$$\frac{(I)^3}{(II)} \Rightarrow \frac{\frac{a^3}{(1-r)^3}}{\frac{a^3}{1-r^3}} = \frac{57}{9717} = 19$$

On solving, $r = \frac{2}{3}$ and $r = -$ (rejected)

a = 19

$$\therefore a + 18r = 19 + 18 \times \frac{2}{3} = 31$$

Ans. (3)

Sol. Favourable cases = 6C_3

Total out comes = 6^3

Probability of getting greater number than previous

$$\text{one} = \frac{^6C_3}{r^3} = \frac{20}{216} = \frac{5}{54}$$

- 18.** The value of the integral $\int_{-1}^2 \log_e(x + \sqrt{x^2 - 1}) dx$ is :

$$(1) \sqrt{5} - \sqrt{2} + \log_e \left(\frac{9+4\sqrt{5}}{1+\sqrt{2}} \right)$$

$$(2) \sqrt{2} - \sqrt{5} \quad \log_e \left(\frac{9 + 4\sqrt{5}}{1 + \sqrt{5}} \right)$$

$$(3) \sqrt{5} - \sqrt{2} + \log_e \left(\frac{7+4\sqrt{5}}{1+\sqrt{2}} \right)$$

$$(4) \sqrt{2} - \sqrt{5} + \log_e \left(\frac{7+4\sqrt{5}}{1+\sqrt{2}} \right)$$

Ans. (2)

$$\text{Sol. } I = \int_{-1}^2 1 \cdot \log_e \left(x + \sqrt{x^2 + 1} \right) dx$$

$$= x \log_e \left(x + \sqrt{x^2 + 1} \right) - \int_{-1}^2 \left| \frac{1 + \frac{x}{\sqrt{x^2 + 1}}}{x + \sqrt{x^2 + 1}} \right| dx$$

$$= x \log_e \left(x + \sqrt{x^2 + 1} \right) - \int_{-1}^2 \frac{x}{\sqrt{x^2 + 1}} dx$$

$$= x \log_e \left(x + \sqrt{x^2 + 1} \right) - \sqrt{x^2 + 1}$$

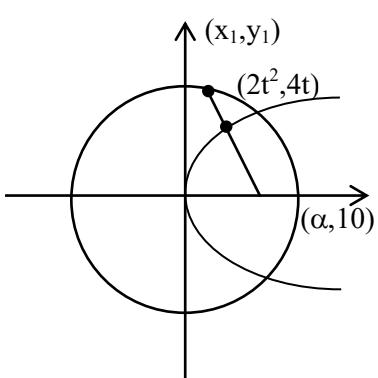
$$= \left(2 \log_e (2 + \sqrt{5}) - \sqrt{5} \right) \\ - \left(-\log_e (-1 + \sqrt{2}) - \sqrt{2} \right)$$

$$= \log \left(2 + \sqrt{5} \right)^2 - \sqrt{5} + \log_e \left(\sqrt{2} - 1 \right) + \sqrt{2}$$

$$= \log \left(2 + \sqrt{5} \right)^2 - \sqrt{5} + \log_e \left(\sqrt{2} - 1 \right) + \sqrt{2}$$

SECTION-B

21. Consider the circle $C : x^2 + y^2 = 4$ and the parabola $P : y^2 = 8x$. If the set of all values of α , for which three chords of the circle C on three distinct lines passing through the point $(\alpha, 0)$ are bisected by the parabola P is the interval (p, q) , then $(2q - p)^2$ is equal to _____.
Ans. (80)

Sol.


$$T = S_1$$

$$xx_1 + yy_1 = x_1^2 + y_1^2$$

$$\alpha x_1 = x_1^2 + y_1^2$$

$$\alpha(2t^2) = 4t^4 + 16t^2$$

$$\alpha = 2t^2 + 8$$

$$\frac{\alpha - 8}{2} = t^2$$

$$\text{Also, } 4t^4 + 16t^2 - 4 < 0$$

$$t^2 = -2 + \sqrt{5}$$

$$\alpha = 4 + 2\sqrt{5}$$

$$\therefore \alpha \in (8, 4 + 2\sqrt{5})$$

$$\therefore (2q - p)^2 = 80$$

22. Let the set of all values of p , for which

$f(x) = (p^2 - 6p + 8)(\sin^2 2x - \cos^2 2x) + 2(2-p)x + 7$ does not have any critical point, be the interval (a, b) . Then $16ab$ is equal to _____.
Ans. (252)

Sol. $f(x) = -(p^2 - 6p + 8) \cos 4x + 2(2-p)x + 7$

$$f'(x) = +4(p^2 - 6p + 8) \sin 4x + (4-2p) \neq 0$$

$$\sin 4x \neq \frac{2p-4}{4(p-4)(p-2)}$$

$$\sin 4x \neq \frac{2(p-2)}{4(p-4)(p-2)}$$

$$p \neq 2$$

$$\sin 4x \neq \frac{1}{2(p-4)}$$

$$\Rightarrow \left| \frac{1}{2(p-4)} \right| > 1$$

on solving we get

$$\therefore p \in \left(\frac{7}{2}, \frac{9}{2} \right)$$

$$\text{Hence } a = \frac{7}{2}, b = \frac{9}{2}$$

$$\therefore 16ab = 252$$

23. For a differentiable function $f : IR \rightarrow IR$, suppose

$$f'(x) = 3f(x) + \alpha, \text{ where } \alpha \in IR, f(0) = 1 \text{ and}$$

$$\lim_{x \rightarrow -\infty} f(x) = 7. \text{ Then } 9f(-\log_e 3) \text{ is equal to _____.}$$

Ans. (61)

Sol. $\frac{dy}{dx} - 3y = \alpha$

$$\text{If } y = e^{\int -3dx} = e^{-3x}$$

$$\therefore y - e^{-3x} = \int e^{-3x} \cdot \alpha dx$$

$$y e^{-3x} = \frac{\alpha e^{-3x}}{-3} + c$$

$$(* e^{3x})$$

$$y = \frac{\alpha}{-3} + C \cdot e^{3x}$$

$$\text{on substituting } x = 0, y = 1$$

$$x \rightarrow -\infty, y = 7$$

$$\text{we get } y = 7 - 6e^{3x}$$

$$9f(-\log_e 3) = 61$$

- 24.** The number of integers, between 100 and 1000 having the sum of their digits equals to 14, is _____.

Ans. (70)

Sol. $N = a b c$

(i) All distinct digits

$$a + b + c = 14$$

$$a \geq 1$$

$$b, c \in \{0 \text{ to } 9\}$$

by hit & trial : 8 cases

$$(6, 5, 3) \quad (8, 6, 0) \quad (9, 4, 1)$$

$$(7, 6, 1) \quad (8, 5, 1) \quad (9, 3, 2)$$

$$(7, 5, 2) \quad (8, 4, 2)$$

$$(7, 4, 3) \quad (9, 5, 0)$$

(ii) 2 same, 1 diff $a = b ; c$

$$2a + c = 14$$

by values :

$$\begin{array}{l} (3,8) \\ (4,6) \\ (5,4) \\ (6,2) \\ (7,0) \end{array} \left| \begin{array}{l} \text{Total} \\ \frac{3!}{2!} \times 5 - \end{array} \right.$$

$$= 14 \text{ cases}$$

(iii) all same :

$$3a = 14$$

$$a = \frac{14}{3} \times \text{rejected}$$

$$0 \text{ cases}$$

Hence, Total cases :

$$8 \times 3! + 2 \times (4) + 14$$

$$= 48 + 22$$

$$= 70$$

- 25.** Let $A = \{(x, y) : 2x + 3y = 23, x, y \in \mathbb{N}\}$ and $B = \{x : (x, y) \in A\}$. Then the number of one-one functions from A to B is equal to _____.

Ans. (24)

Sol. $2x + 3y = 23$

$$x = 1 \quad y = 7$$

$$x = 4 \quad y = 5$$

$$x = 7 \quad y = 3$$

$$x = 10 \quad y = 1$$

$$\begin{matrix} A & B \end{matrix}$$

$$(1, 7) \quad 1$$

$$(4, 5) \quad 4$$

$$(7, 3) \quad 7$$

$$(10, 1) \quad 10$$

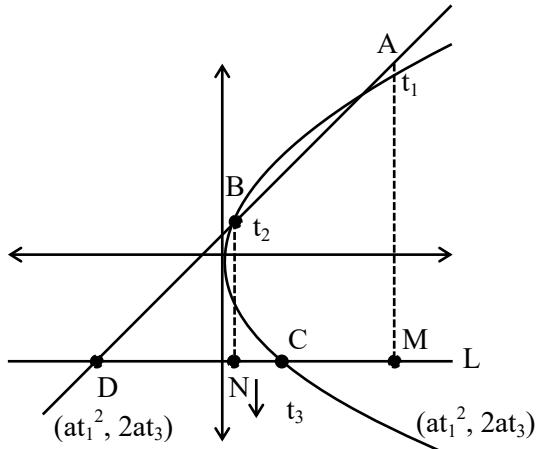
The number of one-one functions from A to B is equal to 4!

- 26.** Let A, B and C be three points on the parabola $y^2 = 6x$ and let the line segment AB meet the line L through C parallel to the x-axis at the point D. Let M and N respectively be the feet of the perpendiculars from A and B on L.

Then $\left(\frac{AM \cdot BN}{CD} \right)^2$ is equal to _____.

Ans. (36)

Sol.



Sol.

$$m_{AB} = m_{AD}$$

$$\Rightarrow \frac{2}{t_1 + t} = \frac{2a(t_1 - t)}{at_1^2 - \alpha}$$

$$\Rightarrow at_1^2 - \alpha = a\{t_1 - t_1 t_3 + t_1 t_2 - t_2 t_3\}$$

$$\Rightarrow \alpha = a(t_1 t_3 + t_2 t_3 - t_1 t_2)$$

$$AM = |2a(t_1 - t)|, BN = |2a(t_2 - t)|,$$

$$CD = |at_3^2 - \alpha|$$



$$\begin{aligned} CD &= \left| at_3^2 - a(t_1t_3 + t_2t_3 - t_1t_2) \right| \\ &= a \left| t_3^2 - t_1t_3 - t_2t_3 + t_1t_2 \right| \\ &= a \left| t_3(t_3 - t_1) - t_2(t_3 - t_1) \right| \end{aligned}$$

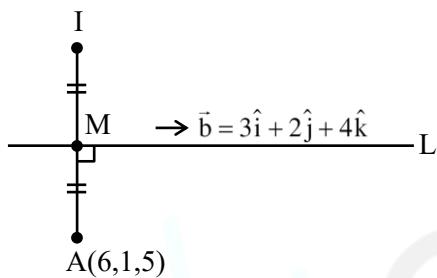
$$CD = a |(t_3 - t_2)(t_3 - t_1)|$$

$$\left(\frac{AM \cdot BN}{CD} \right)^2 = \left\{ \frac{2a(t_1 - t_3) \cdot 2a(t_2 - t_3)}{a(t_3 - t_2)(t_3 - t_1)} \right\}^2$$

$$16a^2 = 16 \times \frac{9}{4} = 36$$

27. The square of the distance of the image of the point $(6, 1, 5)$ in the line $\frac{x-1}{3} = \frac{y}{2} = \frac{z-2}{4}$, from the origin is _____.

Ans. (62)



Sol.

$$\text{Let } M(3\lambda + 1, 2\lambda, 4\lambda + 2)$$

$$\vec{AM} \cdot \vec{b} = 0$$

$$\Rightarrow 9\lambda - 15 + 4\lambda - 2 + 16\lambda - 12 = 0$$

$$\Rightarrow 29\lambda = 29$$

$$\Rightarrow \lambda = 1$$

$$M(4, 2, 6), I(2, 3, 7)$$

$$\text{Required Distance} = \sqrt{4+9+49} = \sqrt{62}$$

Ans. 62

28. If $\left(\frac{1}{\alpha+1} + \frac{1}{\alpha+2} + \dots + \frac{1}{\alpha+1012} \right)$
 $- \left(\frac{1}{2 \cdot 1} + \frac{1}{4 \cdot 3} + \frac{1}{6 \cdot 5} + \dots + \frac{1}{2024 \cdot 2023} \right)$
 $= \frac{1}{2024}$, then α is equal to-

Ans. (1011)

$$\begin{aligned} \text{Sol. } & \left(\frac{1}{\alpha+1} + \frac{1}{\alpha+2} + \dots + \frac{1}{\alpha+2012} \right) \\ & - \left\{ \left(\frac{1}{1} - \frac{1}{2} \right) + \left(\frac{1}{3} - \frac{1}{4} \right) + \dots + \left(\frac{1}{2023} - \frac{1}{2024} \right) \right\} = \frac{1}{2024} \\ & \Rightarrow \left(\frac{1}{\alpha+1} + \frac{1}{\alpha+2} + \dots + \frac{1}{\alpha+2012} \right) \\ & - \left\{ \left(\frac{1}{1} + \frac{1}{2} + \frac{1}{3} + \frac{1}{4} \right) + \dots + \frac{1}{2023} \right. \\ & \quad \left. - \frac{1}{2024} - 2 \left(\frac{1}{2} + \frac{1}{4} + \dots + \frac{1}{2022} \right) \right\} = \frac{1}{2024} \\ & \Rightarrow \left(\frac{1}{\alpha+1} + \frac{1}{\alpha+2} + \dots + \frac{1}{\alpha+2012} \right) \\ & - \left(\frac{1}{1} + \frac{1}{2} + \dots + \frac{1}{2023} \right) \\ & + \frac{1}{2024} + \left(\frac{1}{1} + \frac{1}{2} + \dots + \frac{1}{1011} \right) = \frac{1}{2024} \\ & \Rightarrow \frac{1}{\alpha+1} + \frac{1}{\alpha+2} + \dots + \frac{1}{\alpha+2012} \\ & = \frac{1}{1012} + \frac{1}{1013} + \dots + \frac{1}{2023} \\ & \Rightarrow \alpha = 1011 \end{aligned}$$

29. Let the inverse trigonometric functions take principal values. The number of real solutions of the equation $2 \sin^{-1} x + 3 \cos^{-1} x = \frac{2\pi}{5}$, is _____.

Ans. (0)

$$\begin{aligned} \text{Sol. } & 2 \sin^{-1} x + 3 \cos^{-1} x = \frac{2\pi}{5} \\ & \Rightarrow \pi \cos^{-1} x = \frac{2\pi}{5} \\ & \Rightarrow \cos^{-1} x = \frac{-3\pi}{5} \end{aligned}$$

Not possible

Ans. 0

30. Consider the matrices : $A = \begin{bmatrix} 2 & -5 \\ 3 & m \end{bmatrix}$, $B = \begin{bmatrix} 20 \\ m \end{bmatrix}$

and $X = \begin{bmatrix} x \\ y \end{bmatrix}$. Let the set of all m , for which the

system of equations $AX = B$ has a negative solution
(i.e., $x < 0$ and $y < 0$), be the interval (a, b) .

Then $8 \int_a^b |A| dm$ is equal to _____.

Ans. (450)

Sol. $A = \begin{bmatrix} 2 & -5 \\ 3 & m \end{bmatrix}$, $B = \begin{bmatrix} 20 \\ m \end{bmatrix}$

$$X = \begin{pmatrix} x \\ y \end{pmatrix}$$

$$2x - 5y = 20 \quad \dots(1)$$

$$3x + my = m \quad \dots(2)$$

$$\Rightarrow y = \frac{2m - 60}{2m + 15}$$

$$y < 0 \Rightarrow m \in \left(\frac{-15}{2}, 30 \right)$$

$$x = \frac{25m}{2m + 15}$$

$$x < 0 \Rightarrow m \in \left(\frac{-15}{2}, 0 \right]$$

$$\Rightarrow m \in \left(\frac{-15}{2}, 0 \right)$$

$$|A| = 2m + 15$$

Now,

$$8 \int_{\frac{-15}{2}}^0 (2m + 15) dm = 8 \left\{ m^2 + 15m \right\}_{\frac{-15}{2}}^0$$

$$\Rightarrow 8 \left[\left(\frac{225}{4} - \frac{225}{4} \right) \right]$$

$$= 8 \times \frac{225}{4} = 450$$

(Held On Tuesday 09th April, 2024)

TIME : 3 : 00 PM to 6 : 00 PM

40. The de-Broglie wavelength associated with a particle of mass m and energy E is $\hbar / \sqrt{2mE}$. The dimensional formula for Planck's constant is :
- $[ML^{-1}T^{-2}]$
 - $[ML^2T^{-1}]$
 - $[MLT^{-2}]$
 - $[M^2L^2T^{-2}]$

Ans. (2)

Sol. $\lambda = \frac{\hbar}{\sqrt{2mE}}$ or $E = \hbar\nu$

$$[ML^2T^{-2}] = h[T^{-1}]$$

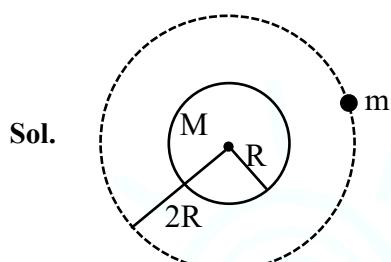
$$h = [ML^2T^{-1}]$$

41. A satellite of 10^3 kg mass is revolving in circular orbit of radius $2R$. If $\frac{10^4 R}{6} J$ energy is supplied to the satellite, it would revolve in a new circular orbit of radius :

(use $g = 10 \text{ m/s}^2$, R = radius of earth)

- $2.5 R$
- $3 R$
- $4 R$
- $6 R$

Ans. (4)



$$\text{Total energy} = \frac{-GMm}{2(2R)}$$

if energy $= \frac{10^4 R}{6}$ is added then

$$\frac{-GMm}{4R} + \frac{10^4 R}{6} = \frac{-GMm}{2r}$$

where r is new radius of revolving and $g = \frac{GM}{R^2}$

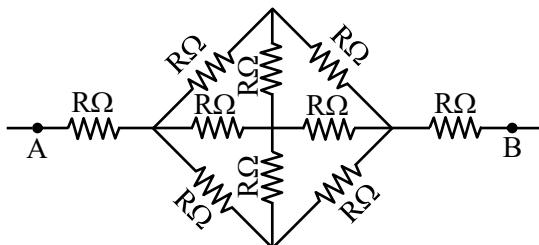
$$-\frac{mgR}{4} + \frac{10^4 R}{6} = -\frac{mgR^2}{2r} \quad (m = 10^3 \text{ kg})$$

$$-\frac{10^3 \times 10 \times R}{4} + \frac{10^4 R}{6} = -\frac{10^3 \times 10 \times R^2}{2r}$$

$$-\frac{1}{4} + \frac{1}{6} = -\frac{R}{2r}$$

$$r = 6R$$

42. The effective resistance between A and B , if resistance of each resistor is R , will be



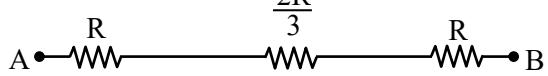
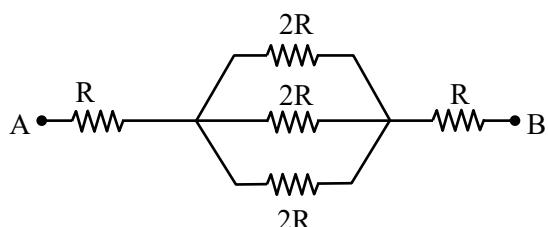
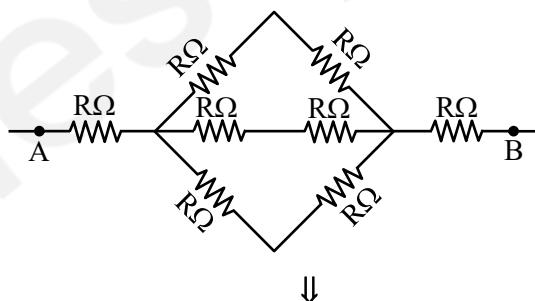
- $\frac{2}{3} R$
- $\frac{8R}{3}$

- $\frac{5R}{3}$
- $\frac{4R}{3}$

Ans. (2)

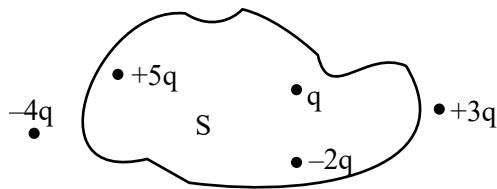
Sol. From symmetry we can remove two middle resistance.

New circuit is



$$A \xrightarrow{8R/3} B$$

43. Five charges $+q$, $+5q$, $-2q$, $+3q$ and $-4q$ are situated as shown in the figure. The electric flux due to this configuration through the surface S is :



- (1) $\frac{5q}{\epsilon_0}$ (2) $\frac{4q}{\epsilon_0}$
 (3) $\frac{3q}{\epsilon_0}$ (4) $\frac{q}{\epsilon_0}$

Ans. (2)

Sol. As per gauss theorem,

$$\phi = \frac{q_{in}}{\epsilon_0} = \frac{q + (-2q) + 5q}{\epsilon_0} = \frac{4q}{\epsilon_0}$$

44. A proton and a deuteron ($q = +e$, $m = 2.0u$) having same kinetic energies enter a region of uniform magnetic field \vec{B} , moving perpendicular to \vec{B} . The ratio of the radius r_d of deuteron path to the radius r_p of the proton path is :

- (1) $1 : 1$ (2) $1 : \sqrt{2}$
 (3) $\sqrt{2} : 1$ (4) $1 : 2$

Ans. (3)

Sol. In uniform magnetic field,

$$R = \frac{mv}{qB} = \frac{\sqrt{2m(K.E)}}{qB}$$

Since same K.E

$$R \propto \frac{\sqrt{m}}{q}$$

$$\therefore \frac{R_{deuteron}}{R_{proton}} = \sqrt{\frac{m_d}{m_p}} \times \frac{q_p}{q_d} = \sqrt{2} \times 1$$

$$\therefore \gamma_d : \gamma_p = \sqrt{2} : 1$$

45. UV light of 4.13 eV is incident on a photosensitive metal surface having work function 3.13 eV. The maximum kinetic energy of ejected photoelectrons will be :

- (1) 4.13 eV (2) 1 eV
 (3) 3.13 eV (4) 7.26 eV

Ans. (2)

Sol. $E_{photon} = (\text{work function}) + K.E_{max}$
 $\therefore 4.13 = 3.13 + K.E_{max}$
 $\therefore K.E_{max} = 1 \text{ eV}$

46. The energy released in the fusion of 2 kg of hydrogen deep in the sun is E_H and the energy released in the fission of 2 kg of ^{235}U is E_U . The ratio $\frac{E_H}{E_U}$ is approximately :

(Consider the fusion reaction as ${}^1_1\text{H} + 2e^- \rightarrow {}^4_2\text{He} + 2\nu + 6\gamma + 26.7 \text{ MeV}$, energy released in the fission reaction of ^{235}U is 200 MeV per fission nucleus and $N_A = 6.023 \times 10^{23}$)

- (1) 9.13 (2) 15.04
 (3) 7.62 (4) 25.6

Ans. (3)

Sol. In each fusion reaction, ${}^1_1\text{H}$ nucleus are used.

$$\text{Energy released per Nuclei of } {}^1_1\text{H} = \frac{26.7}{4} \text{ MeV}$$

\therefore Energy released by 2 kg hydrogen (E_H)

$$= \frac{2000}{1} \times N_A \times \frac{26.7}{4} \text{ MeV}$$

&

\therefore Energy released by 2 kg Vranium (E_V)

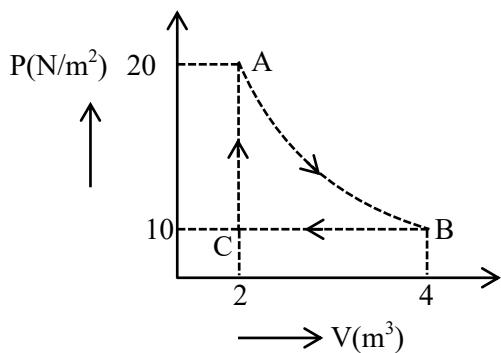
$$= \frac{2000}{235} \times N_A \times 200 \text{ MeV}$$

So,

$$\frac{E_H}{E_V} = 235 \times \frac{26.7}{4 \times 200} = 7.84$$

\therefore Approximately close to 7.62

47. A real gas within a closed chamber at 27°C undergoes the cyclic process as shown in figure. The gas obeys $PV^3 = \text{RT}$ equation for the path A to B . The net work done in the complete cycle is (assuming $R = 8\text{J/molK}$):



- (1) 225 J (2) 205 J
 (3) 20 J (4) -20 J

Ans. (2)

Sol. $W_{AB} = \int PdV$ (Assuming T to be constant)

$$= \int \frac{RTdV}{V^3}$$

$$= RT \int_2^4 V^{-3} dV$$

$$= 8 \times 300 \times \left(-\frac{1}{2} \left[\frac{1}{4^2} - \frac{1}{2^2} \right] \right)$$

$$= 225\text{ J}$$

$$W_{BC} = P \int_4^2 dV = 10(2 - 4) = -20\text{ J}$$

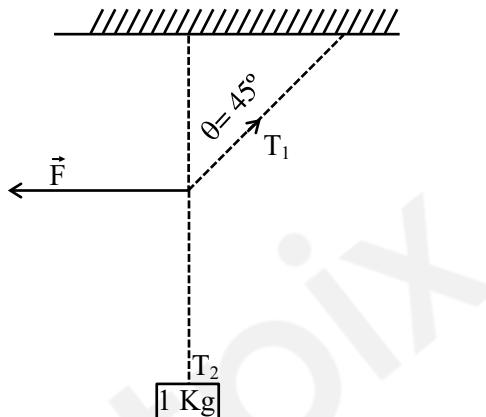
$$W_{CA} = 0$$

$$\therefore W_{\text{cycle}} = 205\text{ J}$$

Note : Data is inconsistent in process AB.

So needs to be challenged.

48. A 1 kg mass is suspended from the ceiling by a rope of length 4m . A horizontal force ' F ' is applied at the mid point of the rope so that the rope makes an angle of 45° with respect to the vertical axis as shown in figure. The magnitude of F is :



- (1) $\frac{10}{\sqrt{2}}\text{ N}$ (2) 1 N
 (3) $\frac{1}{10 \times \sqrt{2}}\text{ N}$ (4) 10 N

Ans. (4)

Sol. $T_1 \sin 45^\circ = F$

$$T_1 \cos 45^\circ = T_2 = 1 \times g$$

$$\therefore \tan 45^\circ = \frac{F}{g}$$

$$\therefore F = 10\text{ N}$$

49. A spherical ball of radius $1 \times 10^{-4}\text{ m}$ and density 10^5 kg/m^3 falls freely under gravity through a distance h before entering a tank of water, If after entering in water the velocity of the ball does not change, then the value of h is approximately :

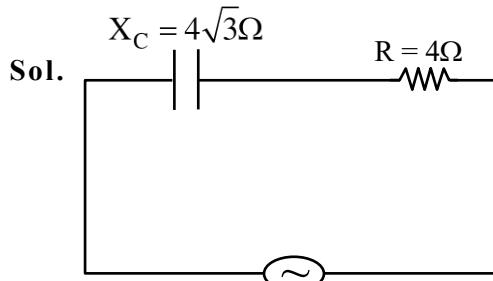
(The coefficient of viscosity of water is $9.8 \times 10^{-6}\text{ N s/m}^2$)

- (1) 2296 m (2) 2249 m
 (3) 2518 m (4) 2396 m

Ans. (3)

53. A capacitor of reactance $4\sqrt{3}\Omega$ and a resistor of resistance 4Ω are connected in series with an ac source of peak value $8\sqrt{2}V$. The power dissipation in the circuit isW.

Ans. (4)



$$Z = \sqrt{R^2 + X^2}$$

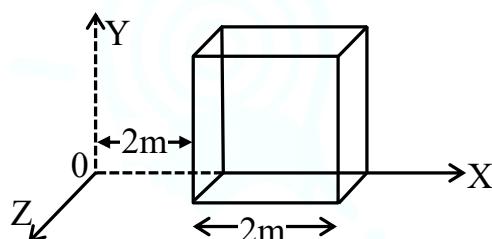
$$Z = \sqrt{4^2 + (4\sqrt{3})^2} = 8\Omega$$

$$V_{rms} = \frac{V}{\sqrt{2}} = \frac{8\sqrt{2}}{\sqrt{2}} = (8V)$$

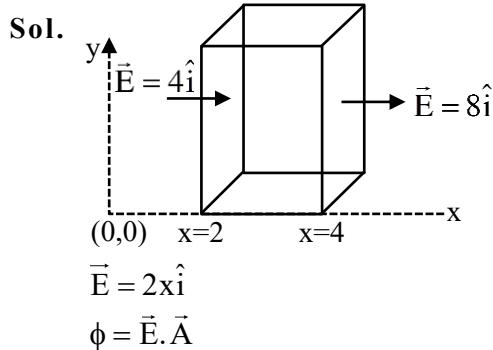
$$I_{rms} = \frac{V_{rms}}{Z} = \frac{8}{8} = 1A$$

$$\text{Power dissipated} = I_{rms}^2 \times R = 1 \times 4 = (4W)$$

54. An electric field $\vec{E} = (2x\hat{i})NC^{-1}$ exists in space. A cube of side 2m is placed in the space as per figure given below. The electric flux through the cube is Nm²/C.



Ans. (16)



$$\phi_{in} = -4 \times 4 = -16 \text{ Nm}^2 / \text{C}$$

$$\phi_{out} = 8 \times 4 = 32 \text{ Nm}^2 / \text{C}$$

$$d_{net} = \phi_{in} + \phi_{out} = -16 + 32 = 16 \text{ Nm}^2 / \text{C}$$

55. A circular disc reaches from top to bottom of an inclined plane of length l . When it slips down the plane, if takes t s. When it rolls down the plane then it takes $\left(\frac{\alpha}{2}\right)^{1/2} t$ s, where α is

Ans. (3)

Sol. For slipping

$$a = g \sin \theta$$

$$l = \frac{1}{2} a t^2 \Rightarrow t = \sqrt{\frac{2l}{g \sin \theta}}$$

For rolling

$$a' = \frac{g \sin \theta}{1 + \frac{k^2}{R^2}} \left[k = \frac{R}{\sqrt{2}} \right]$$

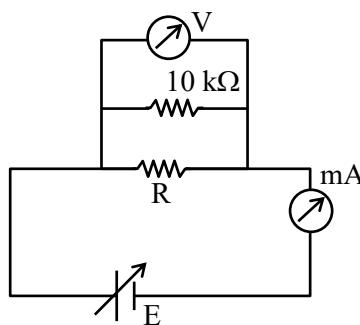
$$\Rightarrow a' = \frac{2g \sin \theta}{3}$$

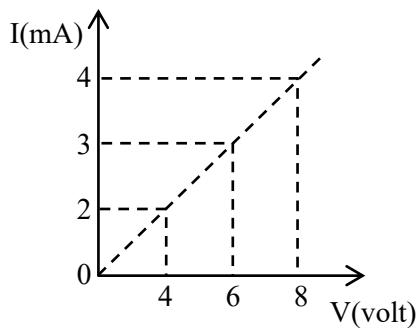
$$l = \frac{1}{2} a'(t')^2$$

$$\Rightarrow t' = \sqrt{\frac{6l}{2g \sin \theta}} = \sqrt{\frac{\alpha}{2}} \sqrt{\frac{2l}{g \sin \theta}}$$

$$\Rightarrow \boxed{\alpha = 3}$$

56. To determine the resistance (R) of a wire, a circuit is designed below. The V-I characteristic curve for this circuit is plotted for the voltmeter and the ammeter readings as shown in figure. The value of R is Ω .




Ans. (2500)

$$\text{Sol. } R_{\text{eq}} = \frac{10^4 R}{10^4 + R}$$

$$E = 4V, I = 2\text{mA}$$

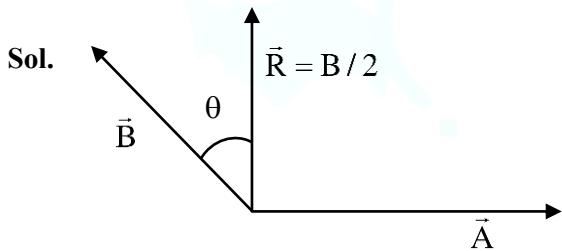
$$I = \frac{E}{R_{\text{eq}}} \Rightarrow 2 \times 10^{-3} = \frac{4(10^4 + R)}{10^4 R}$$

$$\Rightarrow 20R = 40000 + 4R$$

$$16R = 40000$$

$$R = 2500\Omega$$

57. The resultant of two vectors \vec{A} and \vec{B} is perpendicular to \vec{A} and its magnitude is half that of \vec{B} . The angle between vectors \vec{A} and \vec{B} is
.....

Ans. (150)


$$B \cos \theta = \frac{B}{2}$$

$$\Rightarrow \theta = 60^\circ$$

So, angle between \vec{A} & \vec{B} is $90^\circ + 60^\circ = 150^\circ$

58. Monochromatic light of wavelength 500 nm is used in Young's double slit experiment. An interference pattern is obtained on a screen. When one of the slits is covered with a very thin glass plate (refractive index = 1.5), the central maximum is shifted to a position previously occupied by the 4th bright fringe. The thickness of the glass-plate is μm.

Ans. (4)

$$\text{Sol. } (\mu - 1)t = n\lambda$$

$$(1.5 - 1)t = 4 \times 500 \times 10^{-9} \text{ m}$$

$$t = 4000 \times 10^{-9} \text{ m}$$

$$t = 4 \mu\text{m}$$

59. A force $(3x^2 + 2x - 5)$ N displaces a body from $x = 2 \text{ m}$ to $x = 4 \text{ m}$. Work done by this force is J.

Ans. (58)

$$\text{Sol. } W = \int_{x_1}^{x_2} F dx$$

$$W = \int_2^4 (3x^2 + 2x - 5) dx$$

$$W = \left[x^3 + x^2 - 5x \right]_2^4$$

$$W = [60 - 2] \text{ J} = 58 \text{ J}$$

60. At room temperature (27°C), the resistance of a heating element is 50Ω . The temperature coefficient of the material is $2.4 \times 10^{-4} \text{ }^\circ\text{C}^{-1}$. The temperature of the element, when its resistance is 62Ω , is $^\circ\text{C}$.

Ans. (1027)

$$\text{Sol. } R = R_0(1 + \alpha \Delta T)$$

$$62 = 50 [1 + 2.4 \times 10^{-4} \Delta T]$$

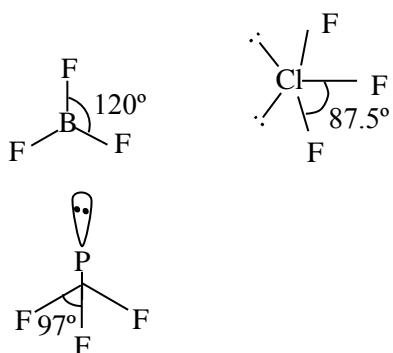
$$\Delta T = 1000^\circ\text{C}$$

$$\Rightarrow T - 27^\circ = 1000^\circ\text{C}$$

$$\boxed{T = 1027^\circ\text{C}}$$

(Held On Tuesday 09th April, 2024)

TIME : 3 : 00 PM to 6 : 00 PM

Sol.


Order of bond angle is


65. Match List I with List II

LIST-I (Test)		LIST-II (Observation)	
A.	Br_2 water test	I.	Yellow orange or orange red precipitate formed
B.	Ceric ammonium nitrate test	II.	Reddish orange colour disappears
C.	Ferric chloride test	III.	Red colour appears
D.	2, 4-DNP test	IV.	Blue, Green, Violet or Red colour appear

Choose the correct answer from the options given below:

- (1) A-I, B-II, C-III, D-IV
- (2) A-II, B-III, C-IV, D-I
- (3) A-III, B-IV, C-I, D-II
- (4) A-IV, B-I, C-II, D-III

Ans. (2)

- Sol.** (A) Br_2 water test is test of unsaturation in which reddish orange colour of bromine water disappears.
- (B) Alcohols give Red colour with ceric ammonium nitrate.
- (C) Phenol gives Violet colour with natural ferric chloride.
- (D) Aldehyde & Ketone give Yellow/Orange/Red Colour compounds with 2, 4-DNP i.e., 2, 4-Dinitrophenyl hydrazine.

66. Match List I with List II

LIST-I (Cell)		LIST-II (Use/Property/Reaction)	
A.	Leclanche cell	I.	Converts energy of combustion into electrical energy
B.	Ni-Cd cell	II.	Does not involve any ion in solution and is used in hearing aids
C.	Fuel cell	III.	Rechargeable
D.	Mercury cell	IV.	Reaction at anode $\text{Zn} \rightarrow \text{Zn}^{2+} + 2\text{e}^-$

Choose the correct answer from the options given below:

- (1) A-I, B-II, C-III, D-IV
- (2) A-III, B-I, C-IV, D-II
- (3) A-IV, B-III, C-I, D-II
- (4) A-II, B-III, C-IV, D-I

Ans. (3)
Sol. A-IV, B-III, C-I, D-II

67. Match List I with List II

LIST-I		LIST-II	
A.	$\text{K}_2[\text{Ni}(\text{CN})_4]$	I.	sp^3
B.	$[\text{Ni}(\text{CO})_4]$	II.	sp^3d^2
C.	$[\text{Co}(\text{NH}_3)_6]\text{Cl}_3$	III.	dsp^2
D.	$\text{Na}_3[\text{CoF}_6]$	IV.	d^2sp^3

Choose the correct answer from the options given below:

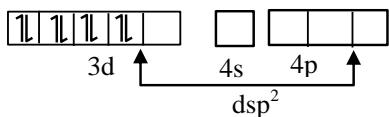
- (1) A-III, B-I, C-II, D-IV
- (2) A-III, B-II, C-IV, D-I
- (3) A-I, B-III, C-II, D-IV
- (4) A-III, B-I, C-IV, D-II

Ans. (4)

Sol. (A) $K_2 [Ni(CN)_4]^{+2}$

$Ni^{2+} : [Ar]3d^8 4s^0$, (CN^- is S.F.L)

Pre hybridization state of Ni^{2+}

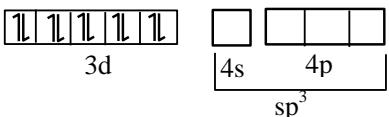


(B) $[Ni(CO)_4]$

$Ni : [Ar] 3d^8 4s^2$

CO is S.F.L, so pairing occur

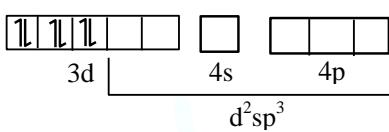
Pre hybridization state of Ni



(C) $[Co(NH_3)_6]Cl_3$

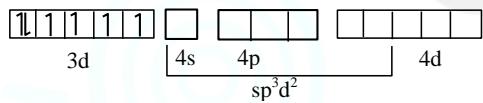
$Co^{+3} : [Ar]3d^6 4s^0$

With Co^{3+} , NH_3 act as S.F.L



(d) $Na_3 [CoF_6]$

$Co^{3+} : [Ar] 3d^6 (F^- : W.F.L)$



68. The coordination environment of Ca^{2+} ion in its complex with $EDTA^{4-}$ is :

- tetrahedral
- octahedral
- square planar
- trigonal prismatic

Ans. (2)

Sol. $EDTA^{4-} \rightarrow$ Hexadentate ligand

$[Ca(EDTA)]^{2-}$

So Coordination environment is octahedral

69. The incorrect statement about Glucose is :

- Glucose is soluble in water because of having aldehyde functional group
- Glucose remains in multiple isomeric form in its aqueous solution
- Glucose is an aldohexose
- Glucose is one of the monomer unit in sucrose

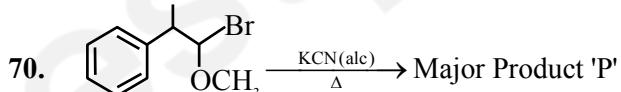
Ans. (1)

Sol. Glucose is soluble in water due to presence of alcohol functional group and extensive hydrogen bonding.

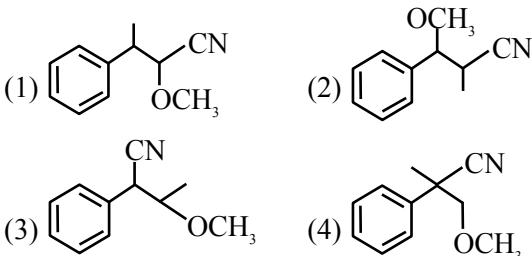
Glucose exist is open chain as well as cyclic forms in its aqueous solution.

Glucose having 6C atoms so it is hexose and having aldehyde functional group so it is aldose. Thus, aldohexose.

Glucose is monomer unit in sucrose with fructose.

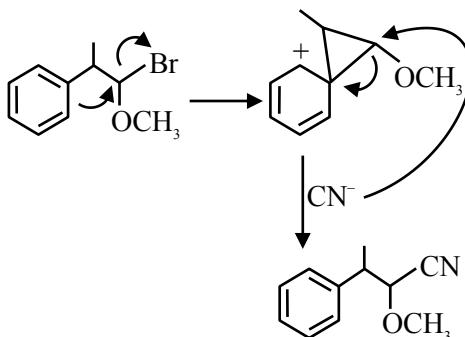


In the above reaction product 'P' is



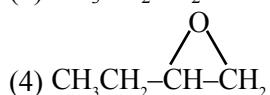
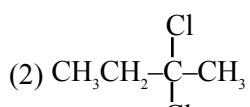
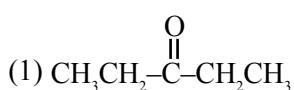
Ans. (1)

Sol.



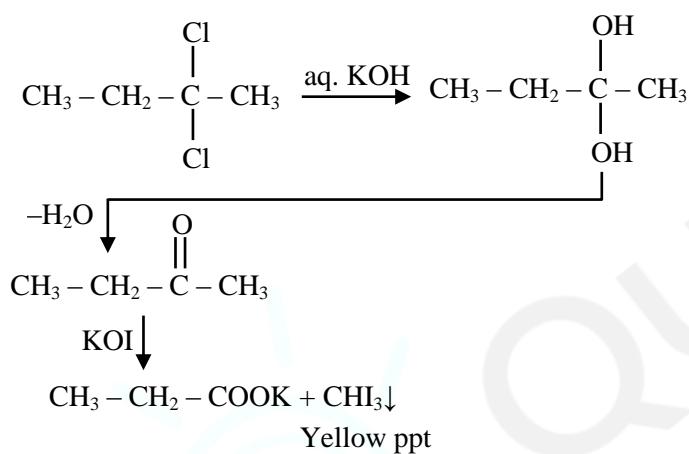
Due to NGP effect of phenyl ring Nucleophilic substitution of Br will occurs.

71. Which of the following compound can give positive iodoform test when treated with aqueous KOH solution followed by potassium hypoiodite.



Ans. (2)

Sol.



72. For a sparingly soluble salt AB_2 , the equilibrium concentrations of A^{2+} ions and B^- ions are $1.2 \times 10^{-4} \text{ M}$ and $0.24 \times 10^{-3} \text{ M}$, respectively. The solubility product of AB_2 is :

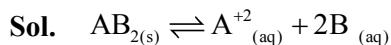
(1) 0.069×10^{-12}

(2) 6.91×10^{-12}

(3) 0.276×10^{-12}

(4) 27.65×10^{-12}

Ans. (2)

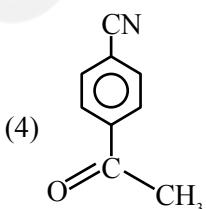
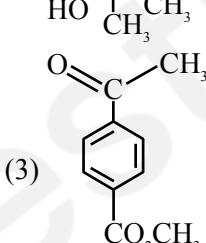
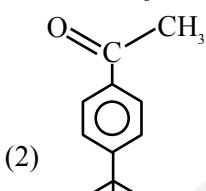
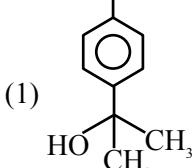
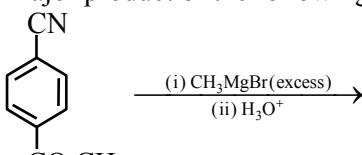


$$K_{\text{sp}} = [\text{A}^{2+}][\text{B}^-]^2$$

$$= 1.2 \times 10^{-4} \times (2.4 \times 10^{-4})^2$$

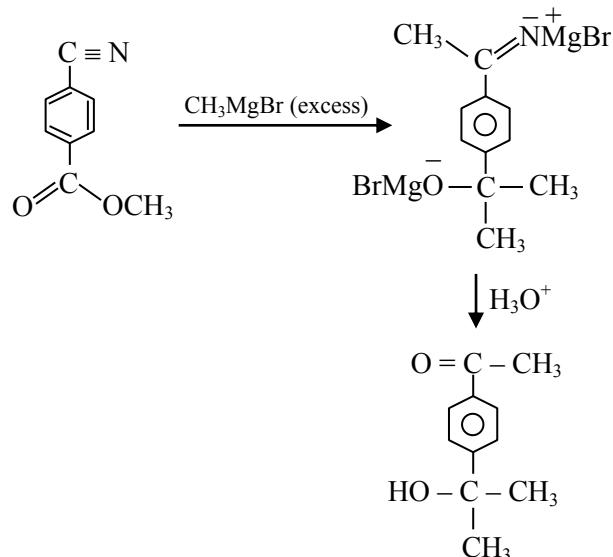
$$= 6.91 \times 10^{-12} \text{ M}^3$$

73. Major product of the following reaction is



Ans. (2)

Sol.



- 74.** Given below are two statements :

Statement I : The higher oxidation states are more stable down the group among transition elements unlike p-block elements.

Statement II : Copper can not liberate hydrogen from weak acids.

In the light of the above statements, choose the correct answer from the options given below :

- (1) Both Statement I and Statement II are false
- (2) Statement I is false but Statement II is true
- (3) Both Statement I and Statement II are true
- (4) Statement I is true but Statement II is false

Ans. (3)

Sol. On moving down the group in transition elements, stability of higher oxidation state increases, due to increase in effective nuclear charge.

$$\Rightarrow E^{\circ}_{\text{Cu}^{+2}/\text{Cu}} = 0.34 \text{ V}$$

$$\Rightarrow E^{\circ}_{\text{H}^+/\text{H}_2} = 0$$



Cu can't liberate hydrogen gas from weak acid.

- 75.** The incorrect statement regarding ethyne is

- (1) The C–C bonds in ethyne is shorter than that in ethene
- (2) Both carbons are sp hybridised
- (3) Ethyne is linear
- (4) The carbon-carbon bonds in ethyne is weaker than that in ethene

Ans. (4)

Sol. The carbon-carbon bonds in ethyne is stronger than that in ethene.

(H–C≡C–H) Ethyne is linear and carbon atoms are SP hybridised.

- 76.** Match List I with List II

List-I (Element)		List-II (Electronic Configuration)	
A.	N	I.	[Ar] $3d^{10} 4s^2 4p^5$
B.	S	II.	[Ne] $3s^2 3p^4$
C.	Br	III.	[He] $2s^2 2p^3$
D.	Kr	IV.	[Ar] $3d^{10} 4s^2 4p^6$

Choose the correct answer from the options given below :

- (1) A-IV, B-III, C-II, D-I
- (2) A-III, B-II, C-I, D-IV
- (3) A-I, B-IV, C-III, D-II
- (4) A-II, B-I, C-IV, D-III

Ans. (2)

Sol. (A) ${}_7\text{N} : [\text{He}] 2s^2 2p^3$
(B) ${}_{16}\text{S} : [\text{Ne}] 2s^2 3p^4$
(C) ${}_{35}\text{Br} : [\text{Ar}] 3d^{10} 4s^2 4p^5$
(D) ${}_{36}\text{Kr} : [\text{Ar}] 3d^{10} 4s^2 4p^6$

- 77.** Match List I with List II

List-I		List-II	
A.	Melting point [K]	I.	$\text{Tl} > \text{In} > \text{Ga} > \text{Al} > \text{B}$
B.	Ionic Radius $[\text{M}^{+3}/\text{pm}]$	II.	$\text{B} > \text{Tl} > \text{Al} \approx \text{Ga} > \text{In}$
C.	$\Delta_i H_1$ $[\text{kJ mol}^{-1}]$	III.	$\text{Tl} > \text{In} > \text{Al} > \text{Ga} > \text{B}$
D.	Atomic Radius [pm]	IV.	$\text{B} > \text{Al} > \text{Tl} > \text{In} > \text{Ga}$

Choose the correct answer from the options given below :

- (1) A-III, B-IV, C-I, D-II
- (2) A-II, B-III, C-IV, D-I
- (3) A-IV, B-I, C-II, D-III
- (4) A-I, B-II, C-III, D-IV

Ans. (3)

Sol. Melting point : B > Al > Tl > In > Ga

Ionic radius (M^{+3}/pm) : Tl > In > Ga > Al > B

$$(\Delta_{\text{IE}} H)_1 \left[\frac{\text{kJ}}{\text{mol}} \right] : B > Tl > Al \approx Ga > In$$

Atomic radius (in pm) : Tl > In > Al > Ga > B

78. Which of the following compounds will give silver mirror with ammoniacal silver nitrate?

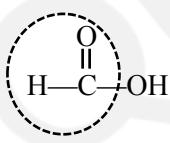
- (A) Formic acid
- (B) Formaldehyde
- (C) Benzaldehyde
- (D) Acetone

Choose the correct answer from the options given below :

- (1) C and D only
- (2) A, B and C only
- (3) A only
- (4) B and C only

Ans. (2)

Sol. Apart from aldehyde, Formic acid



also gives silver mirror test with ammonical silver nitrate.

79. Which out of the following is a correct equation to show change in molar conductivity with respect to concentration for a weak electrolyte, if the symbols carry their usual meaning :

$$(1) \Lambda_m^2 C - K_a \Lambda_m^2 + K_a \Lambda_m \Lambda_m^\circ = 0$$

$$(2) \Lambda - \Lambda_m^\circ + AC^{\frac{1}{2}} = 0$$

$$(3) \Lambda - \Lambda_m^\circ - AC^{\frac{1}{2}} = 0$$

$$(4) \Lambda_m^2 C + K_a \Lambda_m^2 - K_a \Lambda_m \Lambda_m^\circ = 0$$

Ans. (1)

Sol. $\text{HA(aq)} \rightleftharpoons \text{H}^+(\text{aq}) + \text{A}^-(\text{aq})$

$$K_a = \frac{\alpha^2 C}{1 - \alpha}$$

$$\alpha^2 C + K_a \alpha - K_a = 0$$

$$\left(\frac{\infty}{m} \right)^2 C + K_a \frac{\lambda_m}{\lambda} - K_a = 0$$

$$\lambda_m^2 C + K_a \lambda_m \lambda_m^\circ - K_a (\lambda_m^\circ)^2 = 0$$

80. The electronic configuration of Einsteinium is :

(Given atomic number of Einsteinium = 99)

- (1) [Rn] 5f¹² 6d⁰ 7s²
- (2) [Rn] 5f¹¹ 6d⁰ 7s²
- (3) [Rn] 5f¹³ 6d⁰ 7s²
- (4) [Rn] 5f¹⁰ 6d⁰ 7s²

Ans. (2)

Sol. Einsteinium (atomic No = 99) : [Rn] 5f¹¹ 6d⁰ 7s²

SECTION-B

81. Number of oxygen atoms present in chemical formula of fuming sulphuric acid is _____.

Ans. (7)

Sol. Fuming sulphuric acid is a mixture of conc. $\text{H}_2\text{SO}_4 + \text{SO}_3$ Or $\text{H}_2\text{S}_2\text{O}_7$
So, Number of Oxygen atoms = 7

82. A transition metal 'M' among Sc, Ti, V, Cr, Mn and Fe has the highest second ionisation enthalpy. The spin only magnetic moment value of M^+ ion is _____ BM (Near integer)
(Given atomic number Sc : 21, Ti : 22, V : 23, Cr : 24, Mn : 25, Fe : 26)

Ans. (6)

Sol. Among given metals, Cr has maximum IE₂ because Second electron is removed from stable configuration $3d^5$



\therefore No of unpaired e⁻ in Cr^+ is 5, n = 5

So, Magnetic moment = $\sqrt{n(n+2)}$ B.M

$$= \sqrt{5(5+2)} = 5.92 \text{ BM} \approx 6$$

83. The vapour pressure of pure benzene and methyl benzene at 27°C is given as 80 Torr and 24 Torr, respectively. The mole fraction of methyl benzene in vapour phase, in equilibrium with an equimolar mixture of those two liquids (ideal solution) at the same temperature is _____ $\times 10^{-2}$ (nearest integer)

Ans. (23)

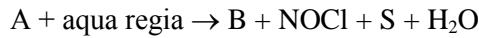
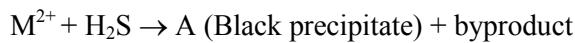
Sol. $X_{\text{methylbenzene}} = 0.5$

$$Y_{\text{methylbenzene}} = \frac{P_{\text{methylbenzene}}}{P_{\text{total}}}$$

$$Y_{\text{methylbenzene}} = \frac{0.5 \times 24}{0.5 \times 80 + 0.5 \times 24}$$

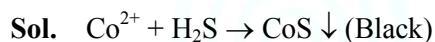
$$= \frac{12}{40+12} = 0.23 = 23 \times 10^{-2}$$

84. Consider the following test for a group-IV cation.

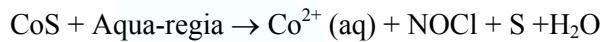


The spin only magnetic moment value of the metal complex C is _____ BM.
(Nearest integer)

Ans. (0)



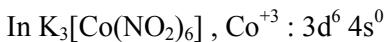
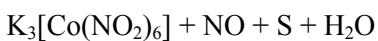
(A)



(A) (B)



↓



$Co^{3+} : d^2 sp^3$ Hybridisation

Number of unpaired $e^- = 0$

$$\text{Magnetic moment} = \sqrt{n(n+2)} = 0 \text{ B.M}$$

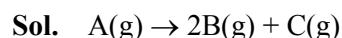
85. Consider the following first order gas phase reaction at constant temperature



If the total pressure of the gases is found to be 200 torr after 23 sec. and 300 torr upon the complete decomposition of A after a very long time, then the rate constant of the given reaction is _____ $\times 10^{-2} \text{ s}^{-1}$ (nearest integer)

[Given : $\log_{10}(2) = 0.301$]

Ans. (3)



$$P_{23} = P_0 + 2x = 200$$

$$P_\infty = 3P_0 = 300$$

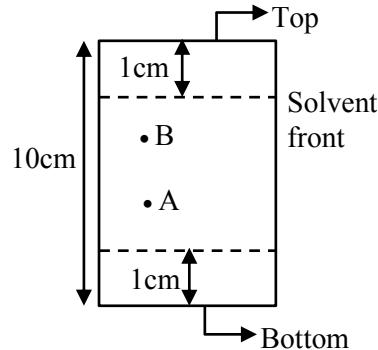
$$P_0 = 100$$

$$K = \frac{1}{t} \ln \frac{P_\infty - P_0}{P_\infty - P_t}$$

$$K = \frac{2.3}{23} \log \frac{300 - 100}{300 - 200}$$

$$= \frac{2.3 \times 0.301}{23} = 0.0301 = 3.01 \times 10^{-2} \text{ sec}^{-1}$$

86.



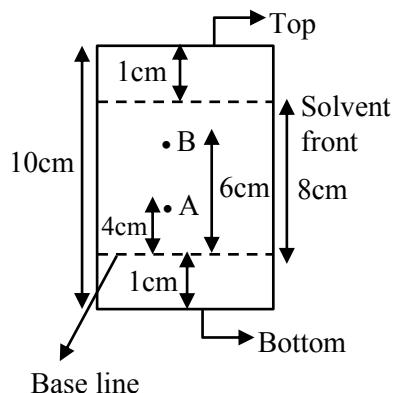
In the given TLC, the distance of spot A & B are 5 cm & 7 cm, from the bottom of TLC plate, respectively.

R_f value of B is $x \times 10^{-1}$ times more than A. The value of x is _____.

Ans. (15)

Sol.

$$R_f = \frac{\text{Distance moved by substance from base line}}{\text{Distance moved by solvent from base line}}$$



$$(R_f)_A = \frac{4}{8} \quad (R_f)_B = \frac{6}{8}$$

$$\frac{(R_f)_B}{(R_f)_A} = \frac{6}{8} \times \frac{8}{4}$$

$$(R_f)_B = 1.5 (R_f)_A$$

$$x = 15$$

- 87.** Based on Heisenberg's uncertainty principle, the uncertainty in the velocity of the electron to be found within an atomic nucleus of diameter 10^{-15} m is _____ $\times 10^9$ ms $^{-1}$ (nearest integer)
 [Given : mass of electron = 9.1×10^{-31} kg,
 Plank's constant (h) = 6.626×10^{-34} Js]
 (Value of π = 3.14)

Ans. (58)

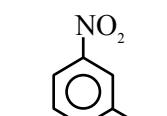
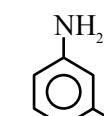
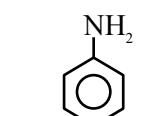
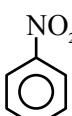
$$\text{Sol. } m\Delta V \cdot \Delta x = \frac{h}{4\pi}$$

$$\Delta V = \frac{6.626 \times 10^{-34}}{9.1 \times 10^{-31} \times 10^{-15} \times 4 \times 3.14} \\ = 57.97 \times 10^9 \text{ m/sec}$$

- 88.** Number of compounds from the following which **cannot** undergo Friedel-Crafts reactions is : _____
 toluene, nitrobenzene, xylene, cumene, aniline, chlorobenzene, m-nitroaniline, m-dinitrobenzene

Ans. (4)

Sol. Compounds which can not undergo Friedel Crafts reaction are



Nitrobenzene Aniline m-nitroaniline m-dinitrobenzene

- 89.** Total number of electron present in (π^*) molecular orbitals of O_2 , O_2^+ and O_2^- is _____.

Ans. (6)

$$\text{Sol. } O_2 (16e) : (\sigma_{1s})^2 (\sigma_{1s}^*)^2 (\sigma_{2s})^2 (\sigma_{2s}^*)^2 \\ (\sigma_{2p})^2 (\pi_{2p})^2 = (\pi_{2p})^2, (\pi_{2p}^*)^1 = (\pi_{2p}^*)^1$$

$$\text{Number of } e^- \text{ present in } (\pi^*) \text{ of } O_2 = 2$$

$$\text{Number of } e^- \text{ present in } (\pi^*) \text{ of } O_2^+ = 1$$

$$\text{Number of } e^- \text{ present in } (\pi) \text{ of } O_2^- = 3$$

$$\text{So total } e^- \text{ in } (\pi^*) = 2 + 1 + 3 = 6$$

- 90.** When $\Delta H_{\text{vap}} = 30 \text{ kJ/mol}$ and $\Delta S_{\text{vap}} = 75 \text{ J mol}^{-1} \text{ K}^{-1}$, then the temperature of vapour, at one atmosphere is _____ K.

Ans. (400)

$$\text{Sol. At equilibrium } \Delta G_{\text{PT}} = 0$$

$$\Delta H_{\text{vap}} = T \Delta S_{\text{vap}}$$

$$30 \times 1000 = T \times 75$$

$$T = 400 \text{ K}$$